

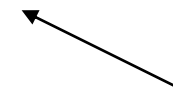
CDL Advances Chemical Studies (ACS)

Metabolic Biochemistry 7 CFU = 6+1

Lecture



Laboratory



Prof.ssa **Alessandra Olianas**

Prof.ssa **Francesca Pintus**

Prof.ssa Alessandra Olianas

Department of Life and Environmental Science
Biomedical Section (Biochemistry Laboratory)

Dip. Scienze della Vita e dell'Ambiente
Sezione Biomedica (laboratorio di Biochimica)

Tel. 0706754507

Students reception: **by appointment**

olianas@unica.it

Course Contents

We will review the structures of the simplest compounds and then continue with more complex compounds

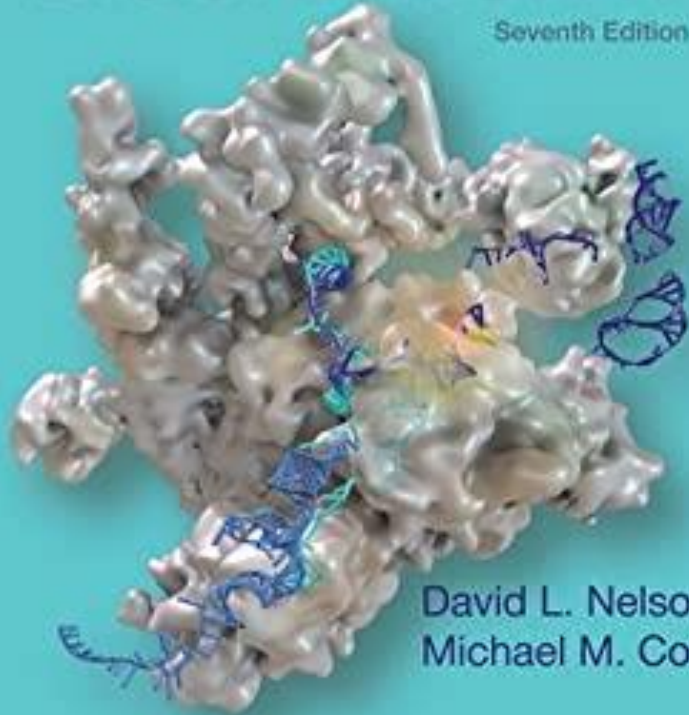
Complex compounds

- **Carbohydrates:** glycosaminoglycans and proteoglycans; protein glycosylation (N-, O- glycosylation and GPI-linked proteins);
- **Lipids:** polyunsaturated fatty acids; arachidonic acid cascade; eicosanoids; phospholipases.

INTERNATIONAL
EDITION

Lehninger
**PRINCIPLES of
BIOCHEMISTRY**

Seventh Edition



David L. Nelson
Michael M. Cox

LIPIDS

Lipids are a class of macromolecules that are nonpolar and hydrophobic in nature.

Lipids are organic molecules insoluble in water, soluble in non-polar solvents such as ether, chloroform, etc...

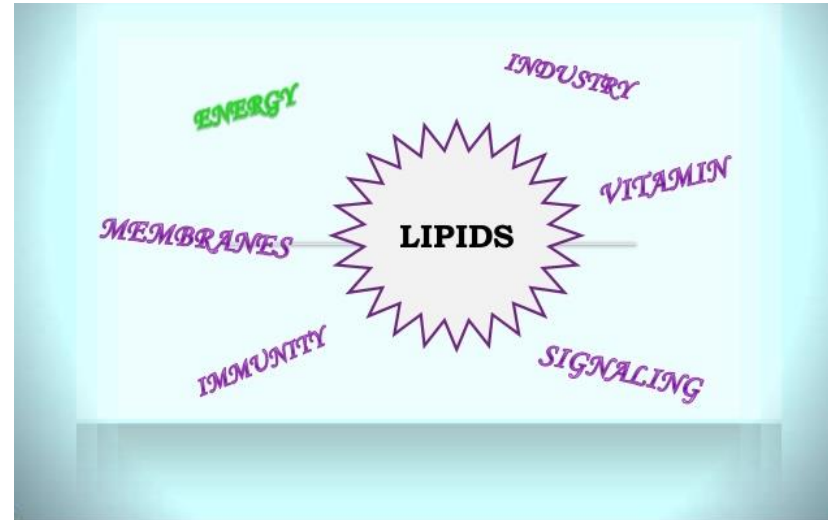
Functions of Lipids

Energy Storage

Large amount of energy is released from their oxidation.

(Triacylglycerols),

Surface coating and protection, thermal insulation (waxes).



Structural: Fats also make cell membranes and coatings (i.e. fruit coats)(glycerophospholipids Sphingolipids)

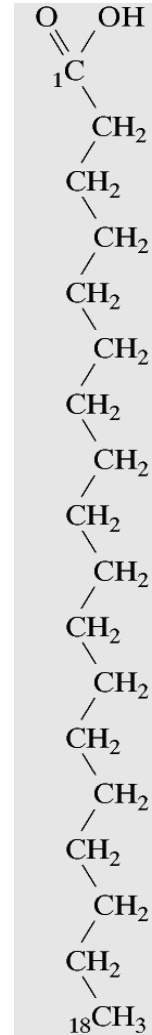
Metabolic regulators , hormonal type, cell recognition

The basic structure is formed by FATTY ACID

A fatty acid molecule has two distinct regions:

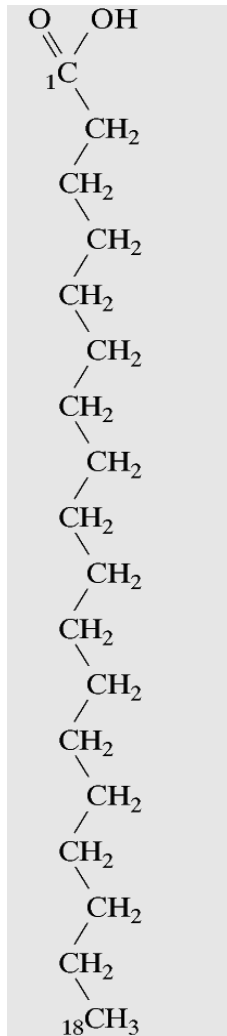
- a long, hydrophobic hydrocarbon chain,
- a carboxylic acid group, (hydrophilic portion)

Consequently, fatty acids are amphipathic because contain two distinct regions (hydrophobic and hydrophilic).

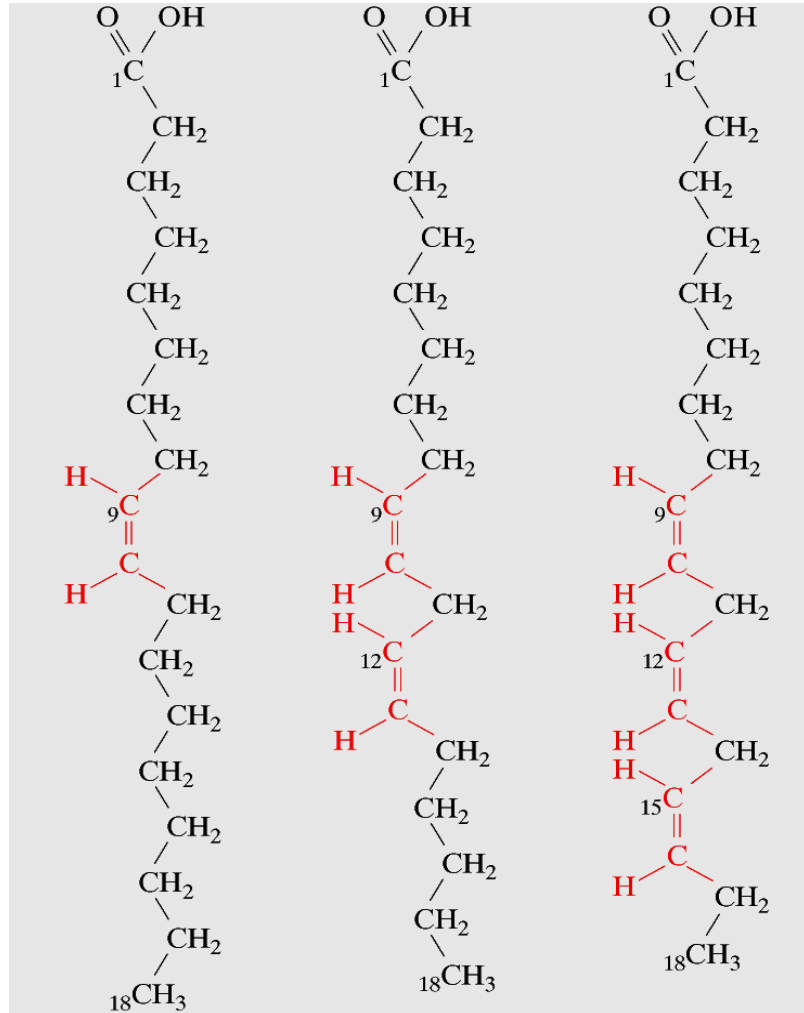


Fatty acids can be distinguished in

- Saturated Fatty Acids



- Unsaturated Fatty Acids
(mono-unsaturated and poly-unsaturated)



Fatty acid: The basic structure of fats is a hydrocarbon backbone with a carboxyl group attached (-COOH in C 1)

They usually contain an even number of carbon atoms (typically 12–20).

They can be distinguished

Saturated,
Unsaturated : mono-unsaturated
poly-unsaturated

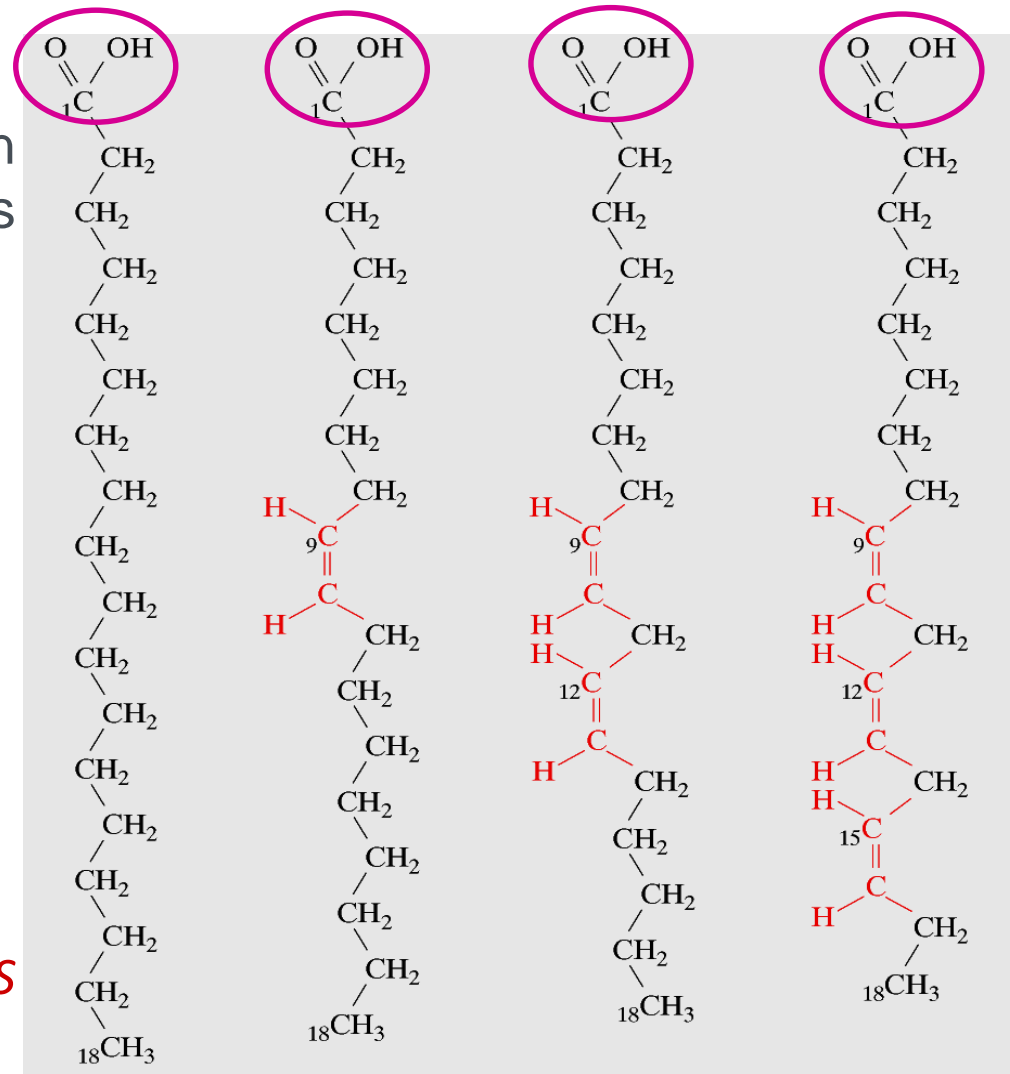
double bond:

- 9-10 position ,
- 12-13 position,
- 15-16 position

Commonly, double bond is in **CIS** conformation

Indicate with Δ^n

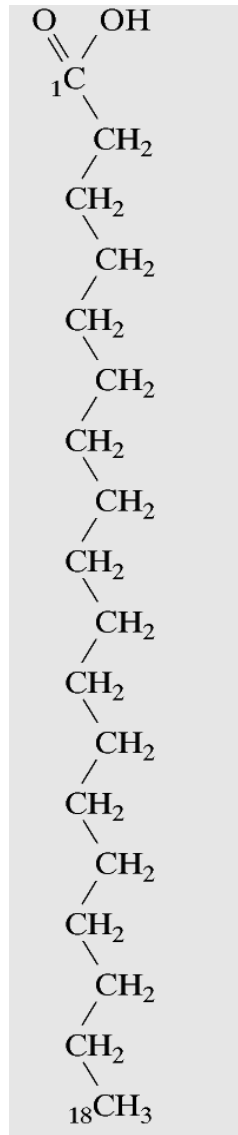
n = double bond position



Octadecanoic \rightarrow Stearic acid. Oleic acid \rightarrow $cis-\Delta^9$ -octadecanoic
 Linoleic acid \rightarrow $cis-\Delta^{9,12}$ -octadecanoic
 linolenic acid \rightarrow $cis-\Delta^{9,12,15}$ -octadecanoic

Saturated Fatty Acids

Hydrocarbon tail

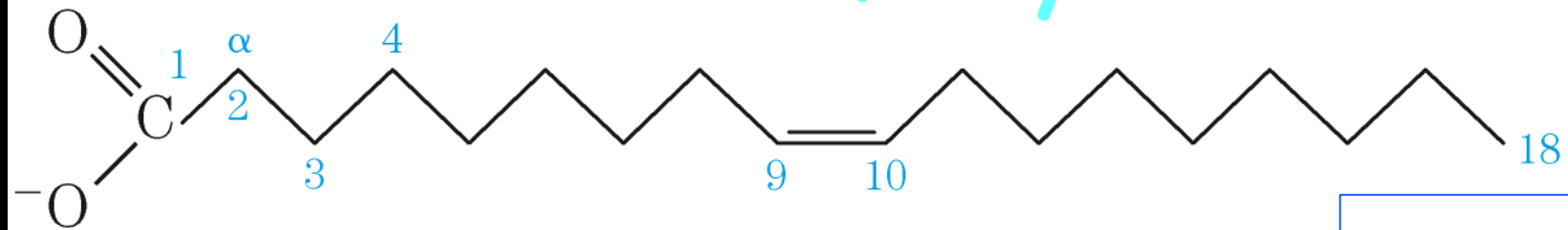


Abbreviation: 18:0

Common name: Stearic acid

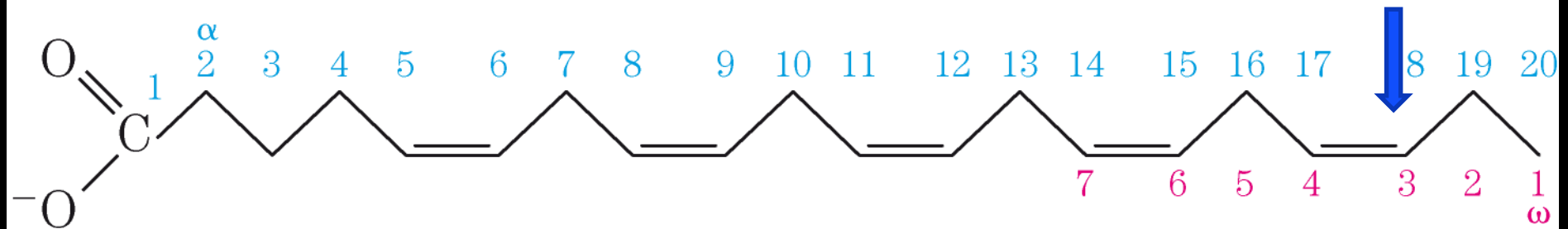
IUPAC: n-Octadecanoic acid

Two conventional nomenclature of natural fatty acid



(a) 18:1(Δ^9) *Cis*- Δ^9 -Octadecanoic acid
(Oleic acid)

The numbering of the chain starts on the opposite side with respect to -COOH

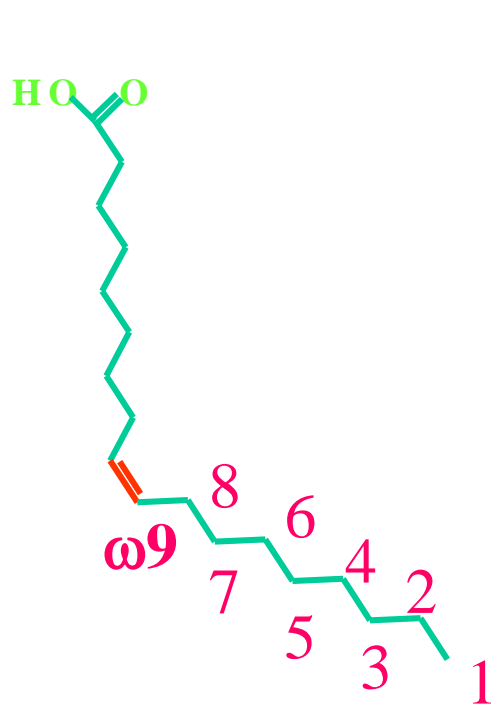


(b) 20:5($\Delta^{5,8,11,14,17}$) Eicosapentanoic acid (EPA),

ω -3

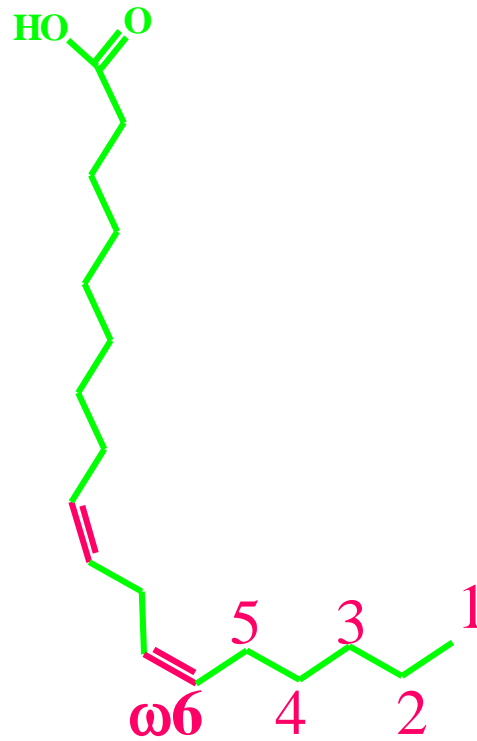
ω

Unsaturated Fatty Acids



Oleic acid
(18:1 Δ^9)

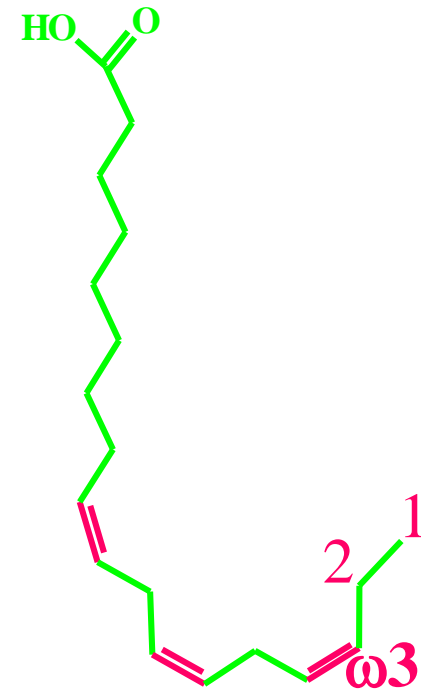
ω-9



Linoleic acid
(18:2 $\Delta^{9,12}$)

ω-6

vegetable oils



α -Linolenic acid
(18:3 $\Delta^{9,12,15}$)

ω-3

Fishes's fat

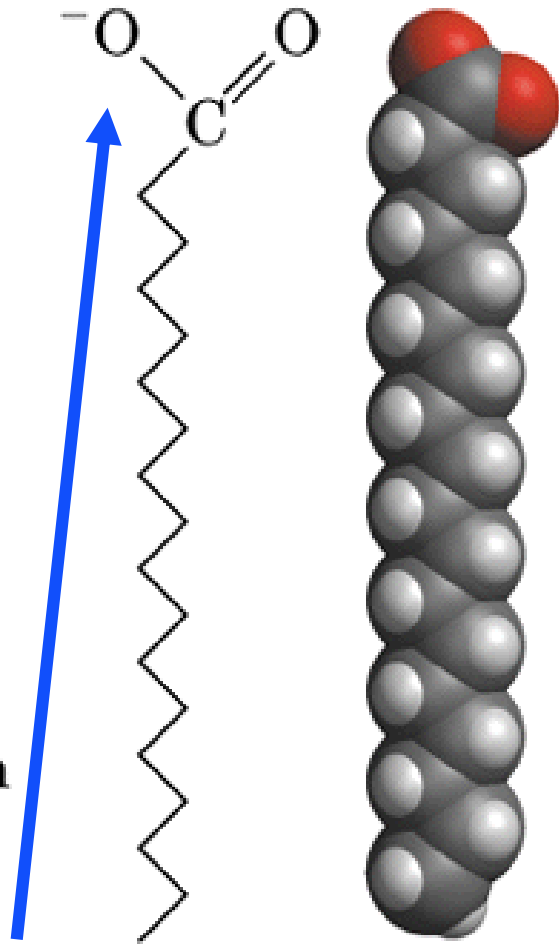
Essential fatty acids

| Carbon skeleton | Structure ^a | Systematic name ^b | Common name (derivation) | Melting point (°C) | Solubility at 30 °C (mg/g solvent) | |
|-------------------------------|--|--|--|--------------------|------------------------------------|---------|
| | | | | | Water | Benzene |
| 12:0 | CH ₃ (CH ₂) ₁₀ COOH | <i>n</i> -Dodecanoic acid | Lauric acid (Latin <i>laurus</i> , "laurel plant") | 44.2 | 0.063 | 2,600 |
| 14:0 | CH ₃ (CH ₂) ₁₂ COOH | <i>n</i> -Tetradecanoic acid | Myristic acid (Latin <i>Myristica</i> , nutmeg genus) | 53.9 | 0.024 | 874 |
| 16:0 | CH ₃ (CH ₂) ₁₄ COOH | <i>n</i> -Hexadecanoic acid | Palmitic acid (Latin <i>palma</i> , "palm tree") | 63.1 | 0.0083 | 348 |
| 18:0 | CH ₃ (CH ₂) ₁₆ COOH | <i>n</i> -Octadecanoic acid | Stearic acid (Greek <i>stear</i> , "hard fat") | 69.6 | 0.0034 | 124 |
| 20:0 | CH ₃ (CH ₂) ₁₈ COOH | <i>n</i> -Eicosanoic acid | Arachidic acid (Latin <i>Arachis</i> , legume genus) | 76.5 | | |
| 24:0 | CH ₃ (CH ₂) ₂₂ COOH | <i>n</i> -Tetracosanoic acid | Lignoceric acid (Latin <i>lignum</i> , "wood" + <i>cera</i> , "wax") | 86.0 | | |
| 16:1(Δ ⁹) | CH ₃ (CH ₂) ₅ CH=CH(CH ₂) ₇ COOH | <i>cis</i> -9-Hexadecenoic acid | Palmitoleic acid | -0.5 | | |
| 18:1(Δ ⁹) | CH ₃ (CH ₂) ₇ CH=CH(CH ₂) ₇ COOH | <i>cis</i> -9-Octadecenoic acid | Oleic acid (Latin <i>oleum</i> , "oil") | 13.4 | | |
| 18:2(Δ ^{9,12}) | CH ₃ (CH ₂) ₄ CH=CHCH ₂ CH=CH(CH ₂) ₇ COOH | <i>cis</i> -, <i>cis</i> -9,12-Octadecadienoic acid | Linoleic acid (Greek <i>linon</i> , "flax") | -5 | | |
| 18:3(Δ ^{9,12,15}) | CH ₃ CH ₂ CH=CHCH ₂ CH=CHCH ₂ CH=CH(CH ₂) ₇ COOH | <i>cis</i> -, <i>cis</i> -, <i>cis</i> -9,12,15-Octadecatrienoic acid | α-Linolenic acid | -11 | | |
| 20:4(Δ ^{5,8,11,14}) | CH ₃ (CH ₂) ₄ CH=CHCH ₂ CH=CHCH ₂ CH=CHCH ₂ CH=CH(CH ₂) ₃ COOH | <i>cis</i> -, <i>cis</i> -, <i>cis</i> -, <i>cis</i> -5,8,11,14-Icosatetraenoic acid | Arachidonic acid | -49.5 | | |

Conformations

Carboxyl group

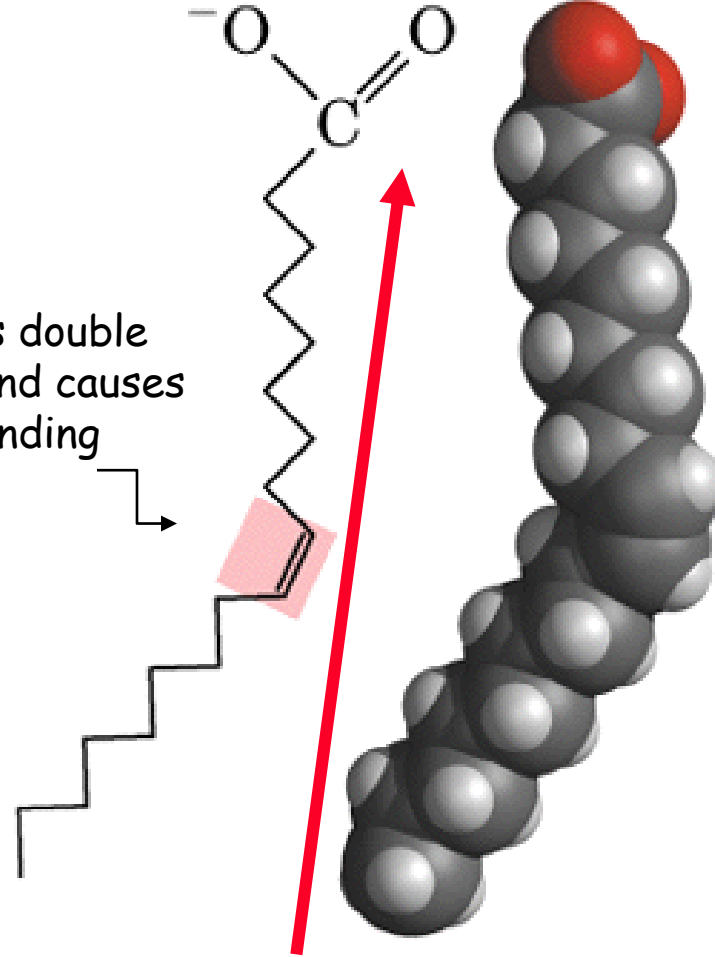
Hydrocarbon chain



Stearic Acid

(a)

Cis double bond causes bending



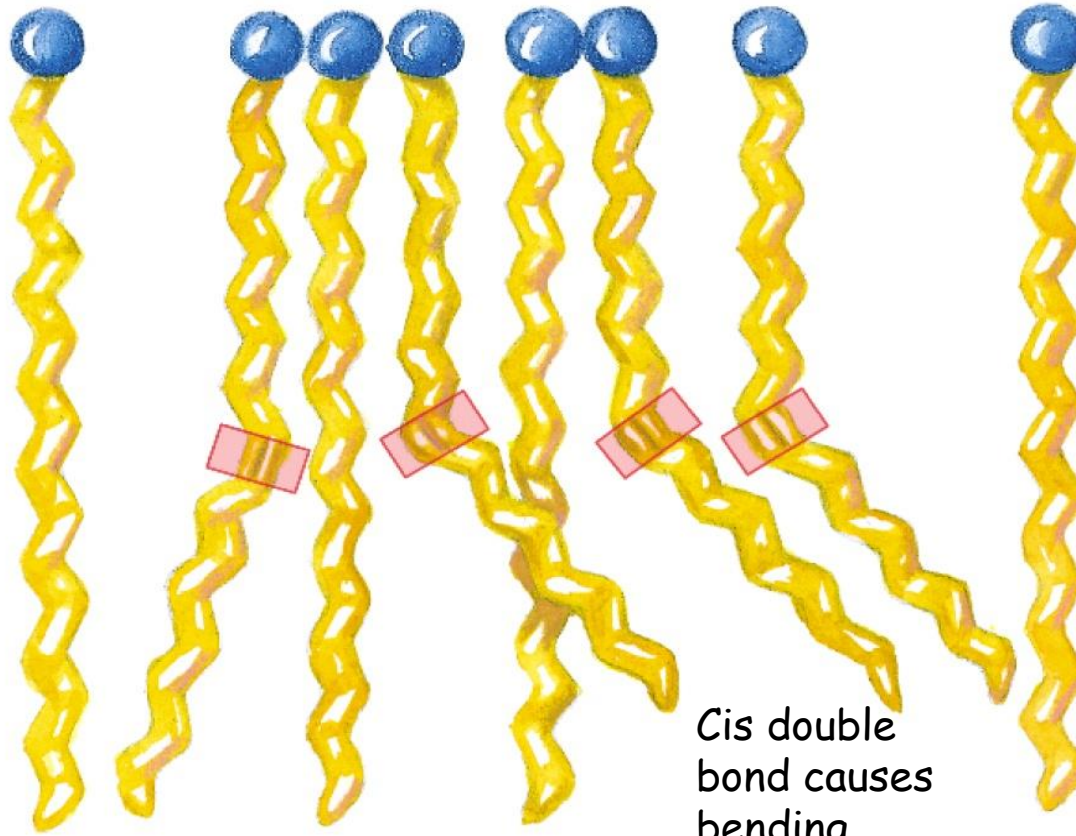
Oleic acid

(b)



Saturated
fatty acids

(c)



Cis double
bond causes
bending

Mixture of saturated and
unsaturated fatty acids

(d)

Unsaturated

Oil



Liquid at room temperature

Saturated

Fat



Solid at room temperature

Natural fatty acids

SATURATED

| Carbon skeleton | Structure* | Systematic name [†] | Common name (derivation) | Melting point (°C) | Solubility at 30 °C (mg/g solvent) | |
|------------------------------|--|--|--|--------------------|------------------------------------|---------|
| | | | | | Water | Benzene |
| 12:0 | $\text{CH}_3(\text{CH}_2)_{10}\text{COOH}$ | <i>n</i> -Dodecanoic acid | Lauric acid (Latin <i>laurus</i> , "laurel plant") | 44.2 | 0.063 | 2,600 |
| 14:0 | $\text{CH}_3(\text{CH}_2)_{12}\text{COOH}$ | <i>n</i> -Tetradecanoic acid | Myristic acid (Latin <i>Myristica</i> , nutmeg genus) | 53.9 | 0.024 | 874 |
| 16:0 | $\text{CH}_3(\text{CH}_2)_{14}\text{COOH}$ | <i>n</i> -Hexadecanoic acid | Palmitic acid (Latin <i>palma</i> , "palm tree") | 63.1 | 0.0083 | 348 |
| 18:0 | $\text{CH}_3(\text{CH}_2)_{16}\text{COOH}$ | <i>n</i> -Octadecanoic acid | Stearic acid (Greek <i>stear</i> , "hard fat") | 69.6 | 0.0034 | 124 |
| 20:0 | $\text{CH}_3(\text{CH}_2)_{18}\text{COOH}$ | <i>n</i> -Eicosanoic acid | Arachidic acid (Latin <i>Arachis</i> , legume genus) | 76.5 | | |
| 24:0 | $\text{CH}_3(\text{CH}_2)_{22}\text{COOH}$ | <i>n</i> -Tetracosanoic acid | Lignoceric acid (Latin <i>lignum</i> , "wood") | 86.0 | | |
| 16:1(Δ^9) | $\text{CH}_3(\text{CH}_2)_5\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$ | <i>cis</i> -9-Hexadecenoic acid | Palmitoleic acid | -0.5 | | |
| 18:1(Δ^9) | $\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$ | <i>cis</i> -9-Octadecenoic acid | Oleic acid (Latin <i>oleum</i> , "oil") | 13.4 | | |
| 18:2($\Delta^{9,12}$) | $\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$ | <i>cis</i> -, <i>cis</i> -9,12-Octadecadienoic acid | Linoleic acid (Greek <i>linon</i> , "flax") | -5 | | |
| 18:3($\Delta^{9,12,15}$) | $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$ | <i>cis</i> -, <i>cis</i> -, <i>cis</i> -9,12,15-Octadecatrienoic acid | α -Linolenic acid | -11 | | |
| 20:4($\Delta^{5,8,11,14}$) | $\text{CH}_3(\text{CH}_2)_3\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$ | <i>cis</i> -, <i>cis</i> -, <i>cis</i> -, <i>cis</i> -5,8,11,14-Icosatetraenoic acid | Arachidonic acid | -49.5 | | |

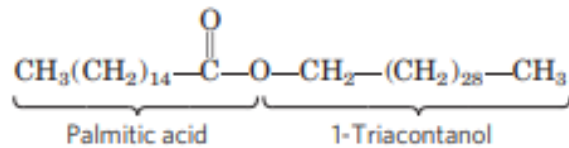
UNSATURATED

Waxes Serve as Energy Stores and Water Repellents

The term "wax" originates in the Old English weax, meaning "the material of the honeycomb"

Esters of long-chain (C14 to C36) saturated fatty acids

Esters unsaturated of long-chain (C16 to C30) fatty acids



(a)

Triacontanoylpalmitate, the major component of beeswax, is an ester of palmitic acid with the alcohol triacontanol.



honeycomb

FUNCTIONS: related to their water-repellent properties and their firm consistency

Birds, particularly waterfowl, secrete waxes to keep their feathers water-repellent.

The shiny leaves of holly and many tropical plants are coated with a thick layer of waxes, which prevents excessive evaporation of water and protects against parasites.

Biological waxes: applications in the pharmaceutical, cosmetic, and other industries.

- Lanolin (from lamb's wool),
- Beeswax
- Carnauba wax (from a Brazilian palm tree)
- Wax extracted from spermaceti oil (from whales)

widely used in the manufacture of lotions, ointments, and polishes.

Lipids Function

1) Storage Lipids and COATING
triacylglycerols (triglycerides): fats, oils and waxes

2) Membrane Lipids
glycerophospholipids
sphingolipids

3) Steroids
Cholesterol and their derivatives:
cholesterols esters
vitamin D
hormones
bile acid

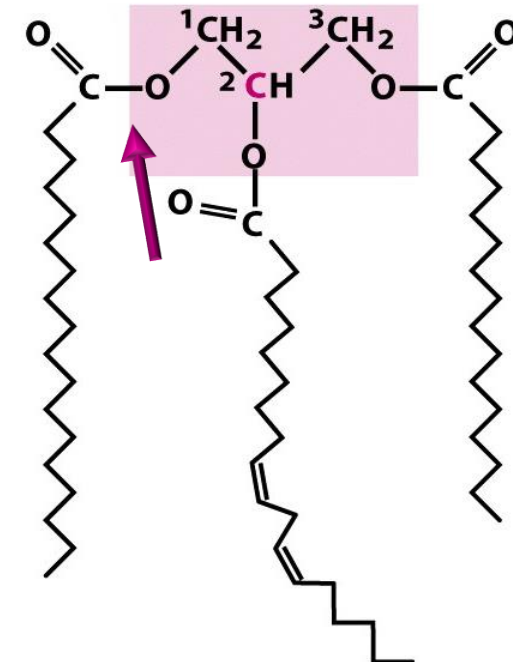
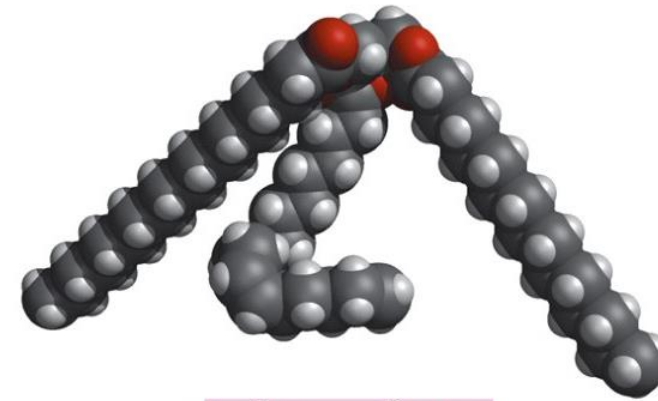
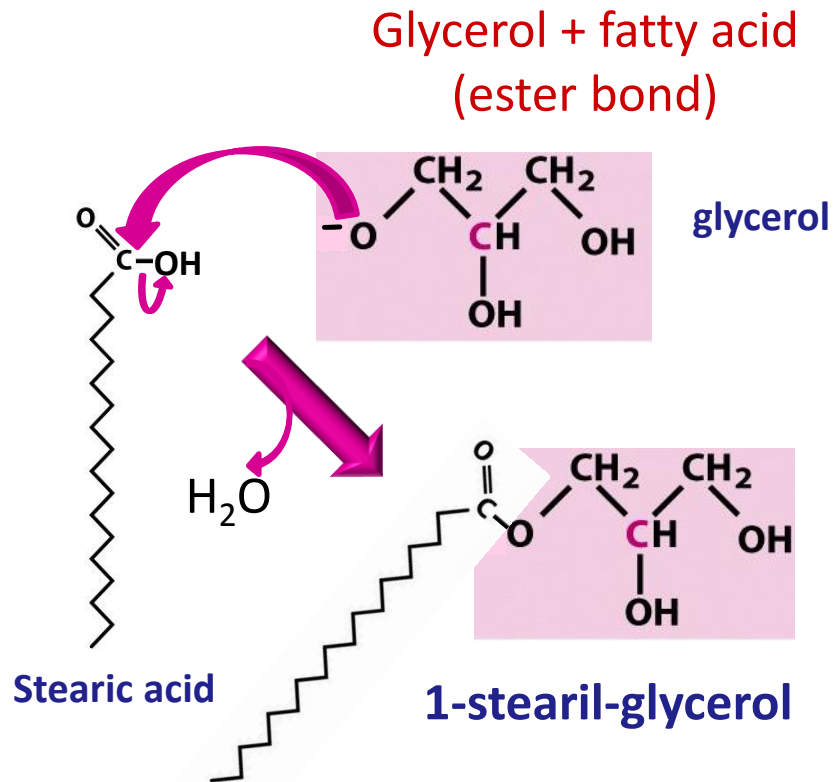
1) STORAGE LIPIDS : TRIGLYCERIDES

Energy is released from their oxidation.

➔ Neutral lipids, apolar

Stored in anhydrous form in fat cells in membrane-limited vacuoles,

Mobilized to produce energy under fasting conditions (for example during sleep).



1-stearil,2-linoleil,3-palmitoil glycerol

Figure 10-3
Lehninger Principles of Biochemistry, Fifth Edition
© 2008 W. H. Freeman and Company

Triacylglycerol, Diacylglycerol, and
Monoacylglycerol.

Membrane Lipids: glycerophospholipids sphingolipids, cholesterol

Lipids and proteins are the two major components of all membranes ;

Lipids are major components of Membranes

Proteins

From about 20% in the myelin sheath to over 70% in the inner membrane of mitochondria

Their concentrations vary greatly between different membranes:

Contain:

- One Polar head
- Two Hydrophobic tails

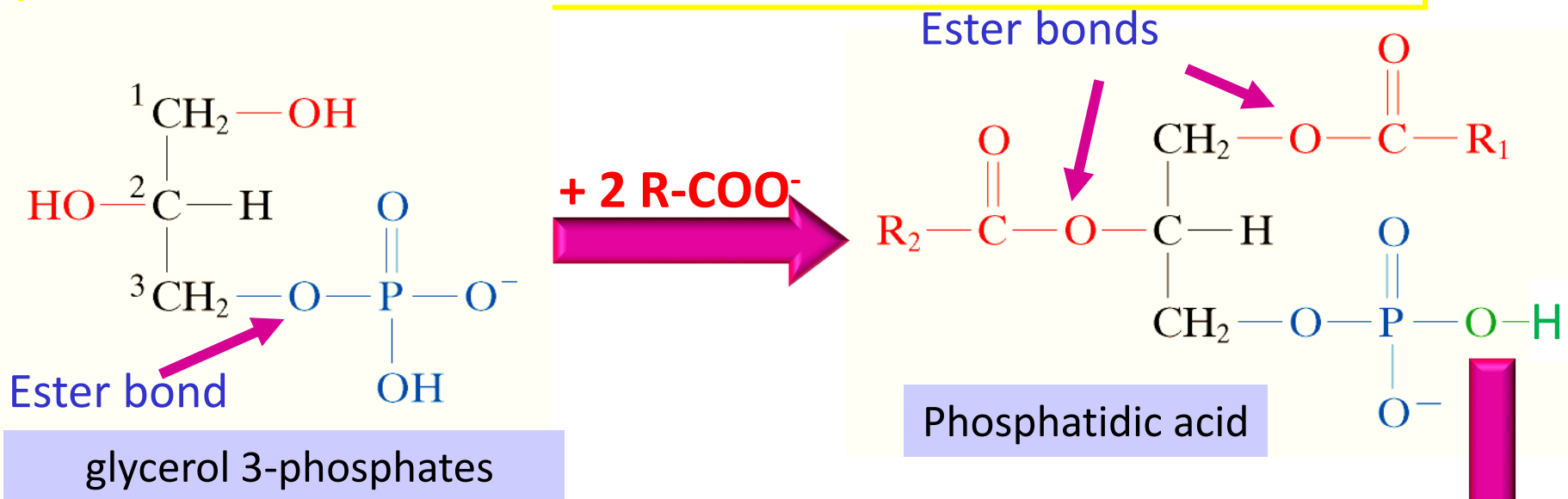
Three major lipids components of eucariotic mambranes:

Membrane Lipids: glycerophospholipids
sphingolipids,
cholesterol

- **Glycerophospholipids (phosphoglycerides)**

They have a glycerol molecule with a phosphate esterified at carbon atom (C3)

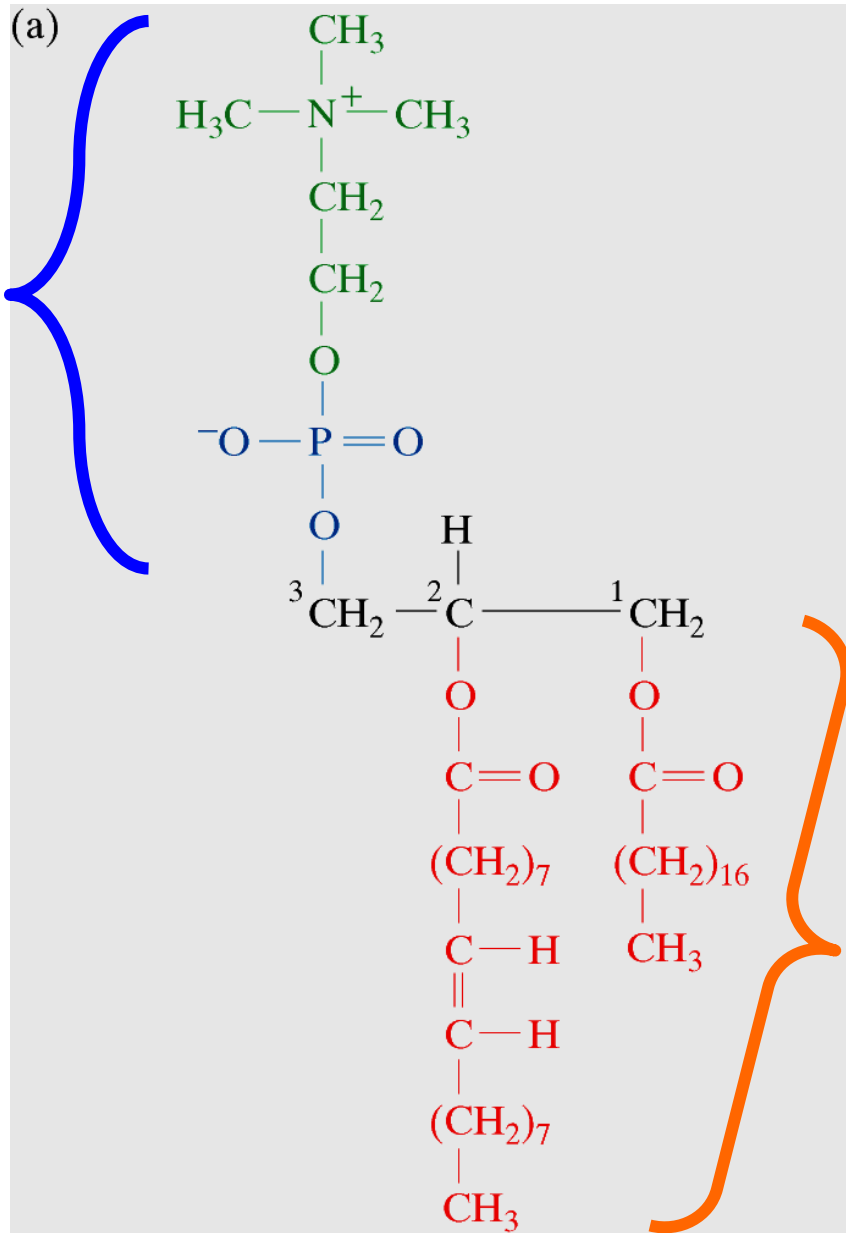
and two long chains of fatty acid esterified to the remaining carbon atoms



The phosphate usually forms a second bond with an alcoholic group of another molecule to form "polar head" of glycerophospholipids.

GLYCEROPHOSPHOLIPIDS ARE Amphipatic MOLECULES

Polar Head.
hydrophilic:
phosphates
+
Choline



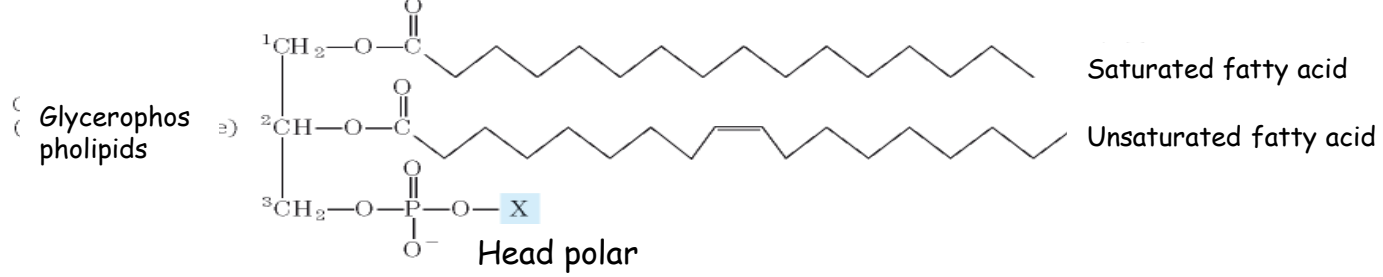
Glycerophospholipids differ for head polar and the type of fatty acid.

Fatty acid differ for the number of carbon atoms and unsaturated

Hydrophobic tail:
glycerol + hydrocarburic chain (stearic in C1 e oleic in C2)
-on C1 usually saturated acilic chain
-on C2 usually unsaturated chain

1-stearil-2-oleil-3-phosphatydil-choline

GLYCEROPHOSPHOLIPIDS



| name | Name X-O | X | Charge at pH 7 |
|--|----------------------------|--|----------------|
| Phosphatidic acid | — | — H | -1 |
| Phosphatidylethanolamine | Ethanolamine | — CH ₂ -CH ₂ -N ⁺ H ₃ | 0 |
| Phosphatidylcholine | Choline | — CH ₂ -CH ₂ -N ⁺ (CH ₃) ₃ | 0 |
| Phosphatidylserine | Serine | — CH ₂ -CH(NH ₃ ⁺)COO ⁻ | -1 |
| Phosphatidylglycerol | Glycerol | — CH ₂ -CH(OH)-CH ₂ -OH | -1 |
| Phosphatidylinositol 4-5 bisphosphates | Inositol 4-5 bisphosphates | | -4 |
| Dihosphatidylglycerol (cardiolipin) | Phosphatidylglycerol | | -2 |

Sphingolipids

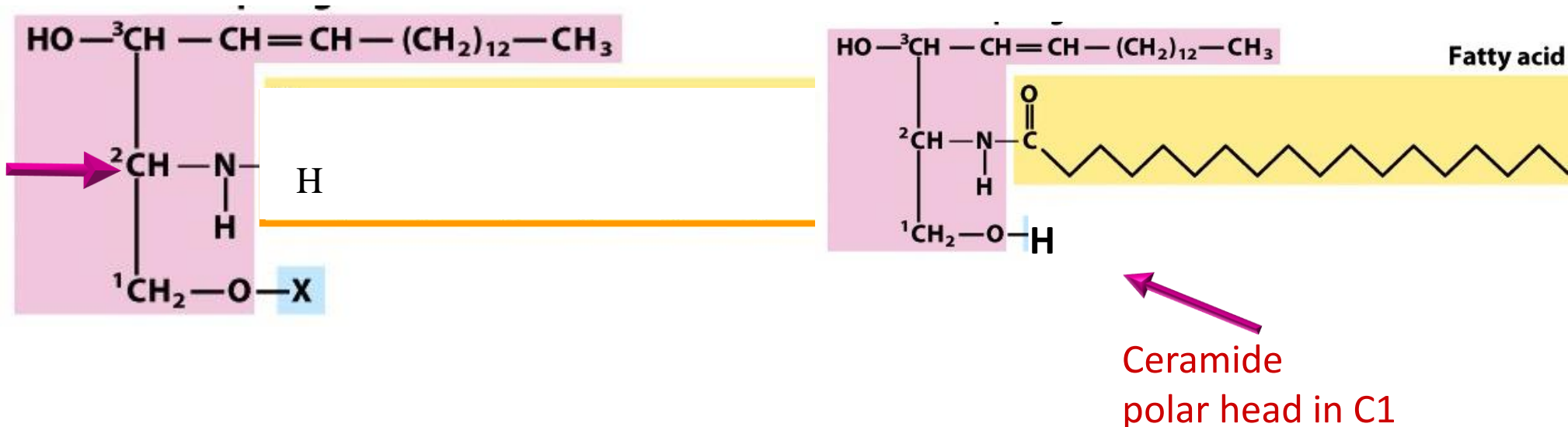
Sphingolipids contain a molecule of sphingosine instead of glycerol

Sphingosine (ammino alcohol of 18 carbon atoms)

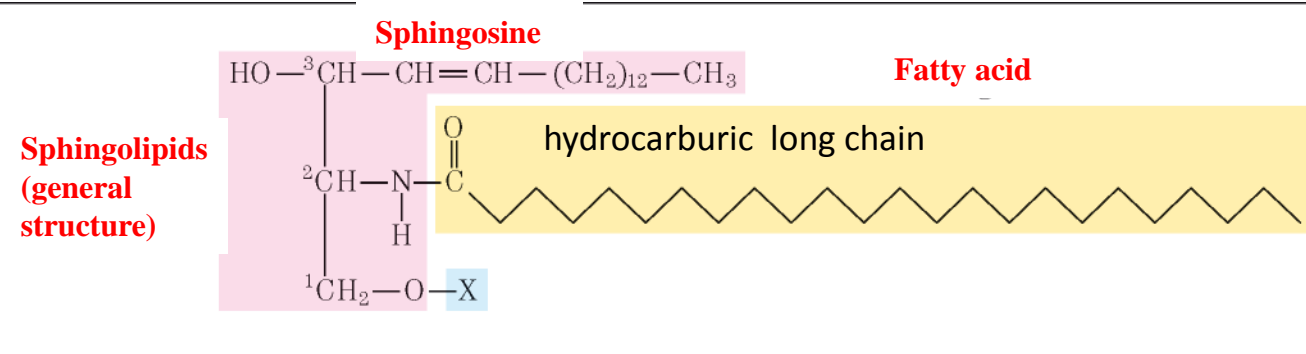
- a monounsatur hydrocarburic long chain in C-3,
- two OH- groups in C-3 and C-1
- one aminic group in C-2.

Sphingosine: fatty acid in C2 and H atom in C1 (CERAMIDE).

CERAMIDE is the basic structural of sphingolipids: fatty acid is linked to amminic group by CARBAMIDIC bond.



sphingolipids



| name | Name X-O | X |
|------------------|---------------------|--|
| ceramide | — | — H |
| 1) Sphingomielin | Phosphatidilcholine | $ \begin{array}{c} \text{O} \\ \\ -\text{P}-\text{O}-\text{CH}_2-\text{CH}_2-\text{N}^+(\text{CH}_3)_3 \\ \\ \text{O}^- \end{array} $ |

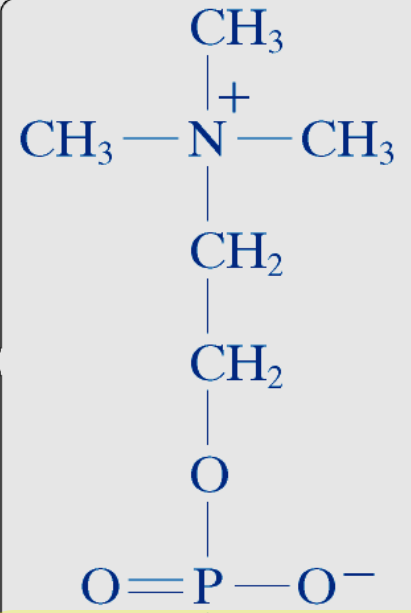
Sphingosine (18 carbon atoms)

- 2 OH- groups in C-3 and C-1
- 1 aminic group in C-2.

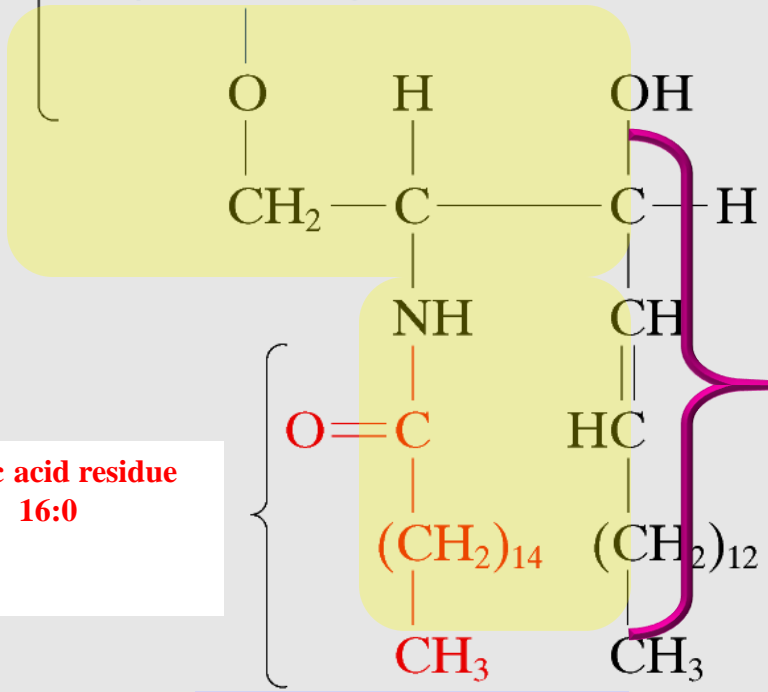
(a)

Sphingolipids
(general structure)

polar head
(phosphocholine)



Sphingomyelins:
(ANPHIPATIC)
They are Phospholipids_



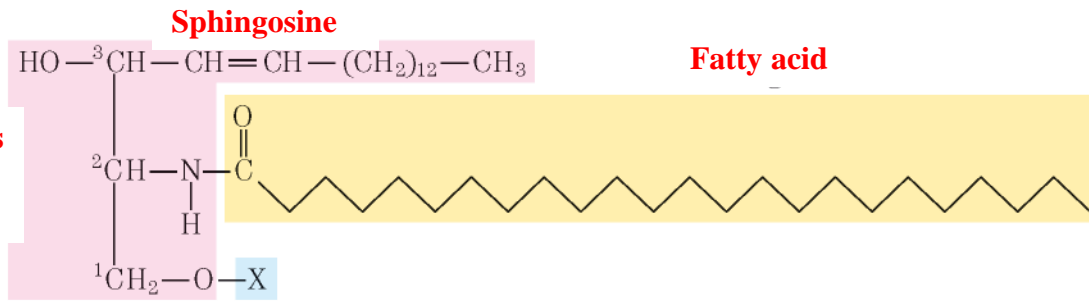
Palmitic acid residue
16:0

Unpolar tails:
Hydrocarburic chain
of sphingosine and
acylic chain of fatty
acid

Sphingomyelin

Sphingolipids

**Sphingolipids
(general
structure)**



1) **name**

Name X-O

X

ceramide

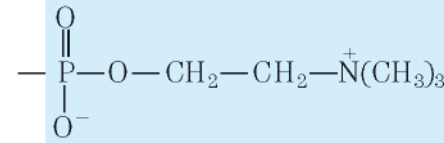
—

— H

1)

Sphingomyelin

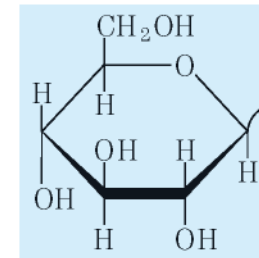
Phosphatidylcholine



2)

**Neutral glycolipids
(Glycosilcerebroside)**

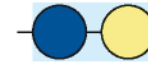
Glucose



3)

**Lactosilceramide
(globoside)**

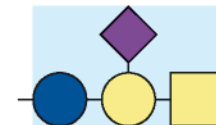
**Di-tri or
tetrasaccharides**



4)

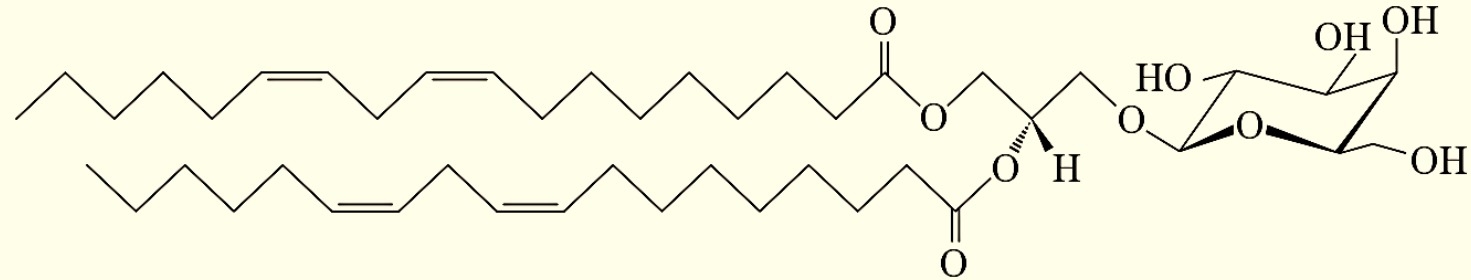
**Ganglioside
GM2**

complex oligosaccharides



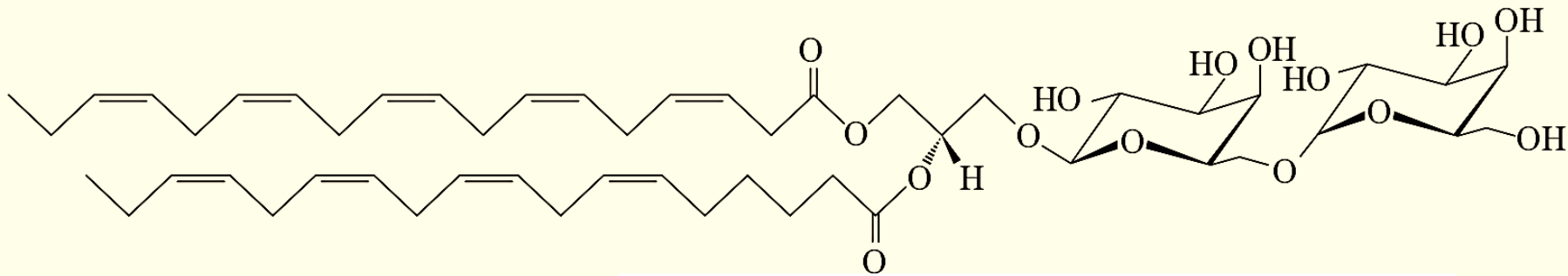
GLYCOLIPIDS: glycerol and polar head with monosaccharides or disaccharides.

They are not phosphorilated



GALACTOLIPIDS

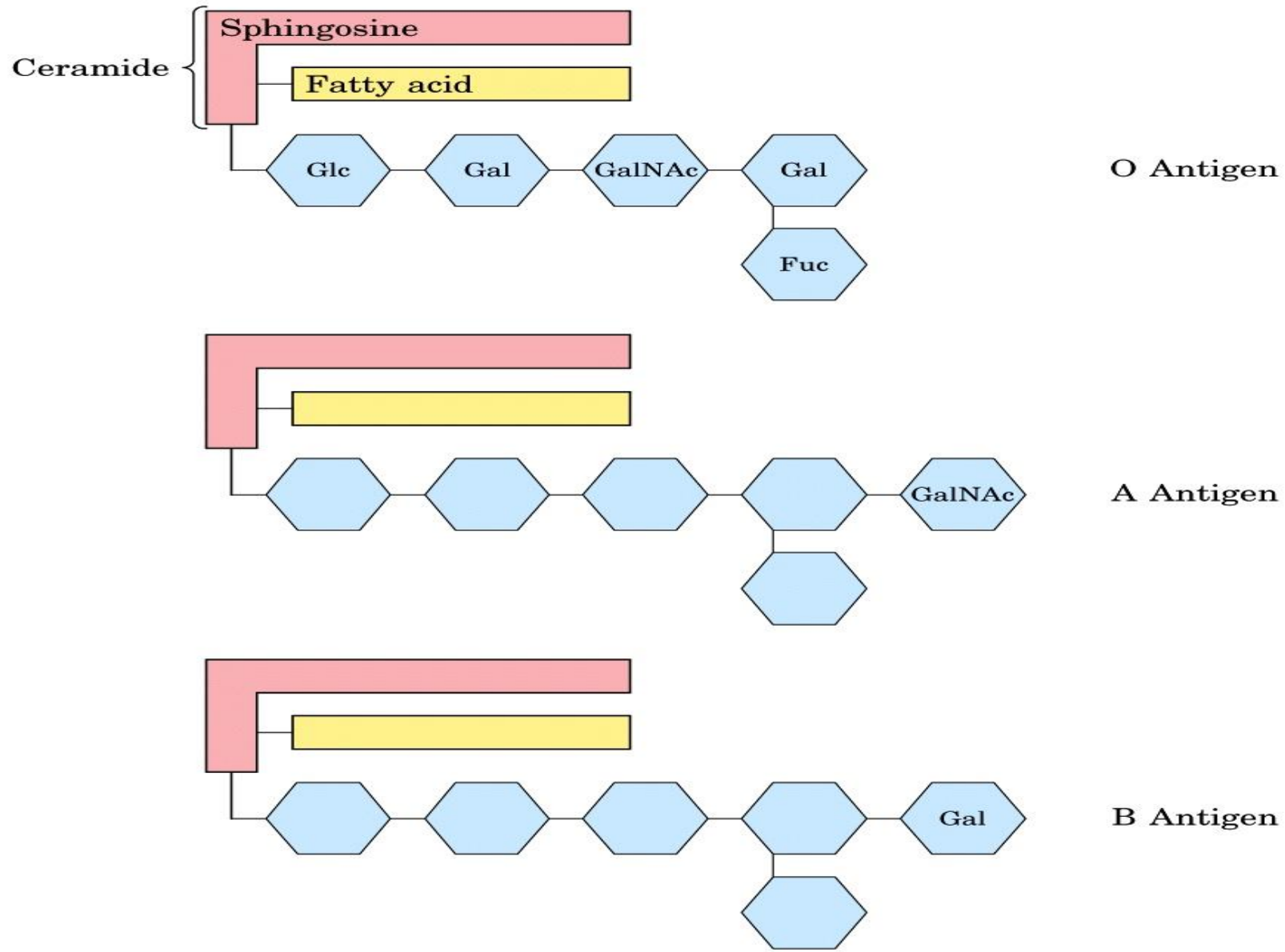
Monogalactosyl diacylglycerol (MGDG)



Digalactosyl diacylglycerol (DGDG)



Glycophingolipids in Blood groups



GLYCOSPHINGOLIPIDS

Not contain phosphorus but contain carbohydrates in glycosidic linkage to the primary alcohol of a ceramide

1) Cerebrosides

2) Globosides

3) Gangliosides



NEUTRAL GLYGOLIPIDS

POLAR GLYGOLIPIDS

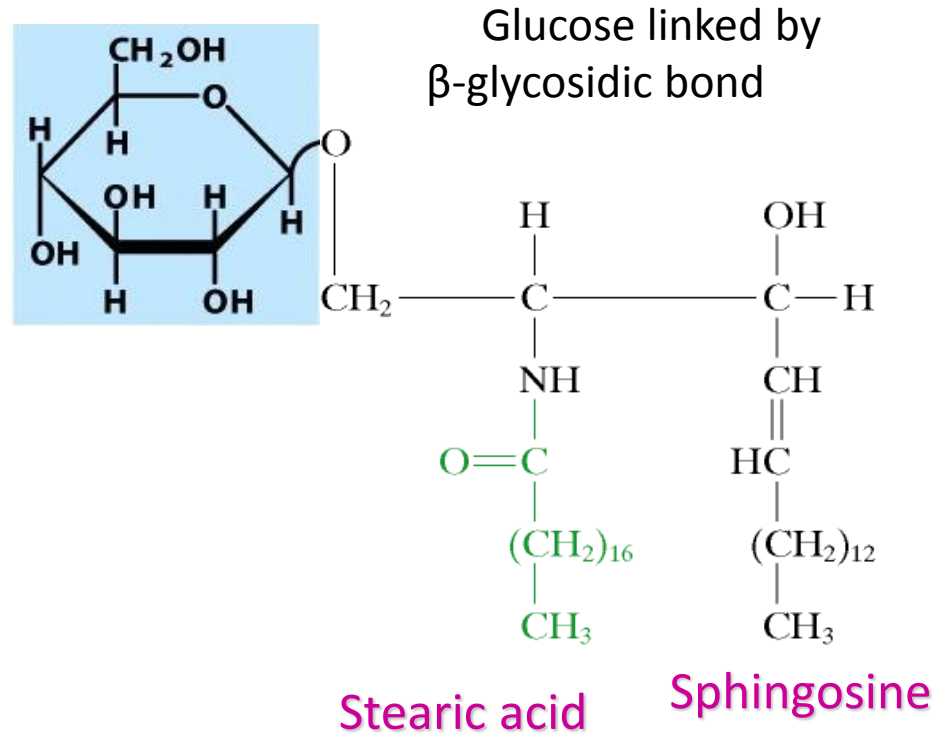


GLYCOPHINGOLIPIDS

1) Cerebrosides

a single Glucose or galactose linked to a ceramide by glycosidic bond

GLUCOSYLCERAMIDE



GLYCOSPHINGOLIPIDS

1) Cerebrosides

2) Globosides

3) Gangliosides

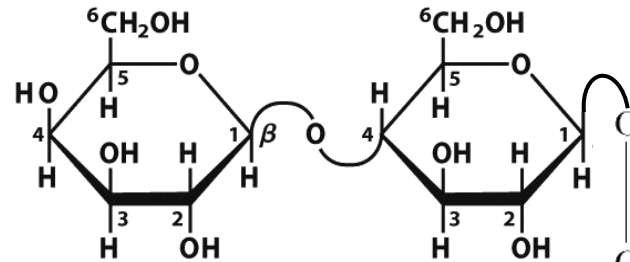


2) Globosides: are ceramide oligosaccharides

two or more neutral monosaccharides (usually galactose or glucose, or N-acetylgalactosamine) linked to a ceramide by glycosidic bond

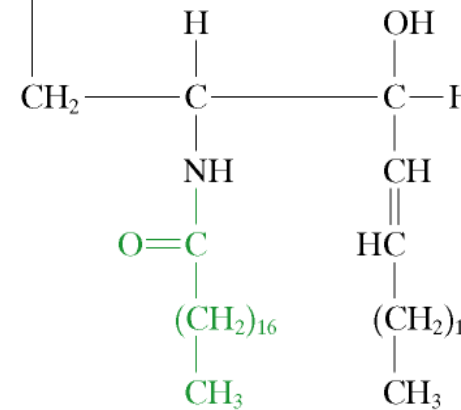
The oligosaccharides are uncharged

Lactosylceramide is present in the erythrocyte membrane



Globosidi:
LACTOSYLCERAMIDE

Another globoside of the nervous system is ceramide-tri-hexoside



Stearic acid

Sphingosine

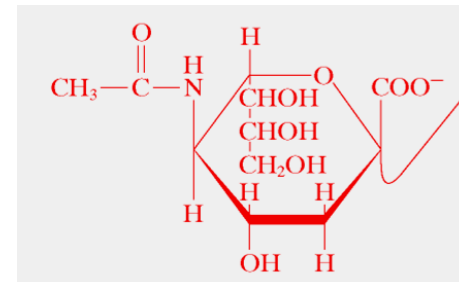
Ceramide-hexoside accumulates in kidney of patients with Fabry's disease, A deficiency of lysosomal alpha-galactosidase A.

GLYCOSPHINGOLIPIDS

3) Gangliosides: First identification in high concentrations in the central nervous system ganglia cells

Sialic acid is highly contained in ganglia cells of the central nervous system

- CNS (50% of total sialic acid is represented)
- Non-nerveous cells (10% of the total sialic acid) (plasma membrane of cells or other tissue)
- Main components of cell-surface membranes
- Constitute 6% of brain lipids.



N-Acetylneuraminic
(sialic acid)
(NANA)

- 1) Cerebrosides
- 2) Globosides
- 3) Gangliosides

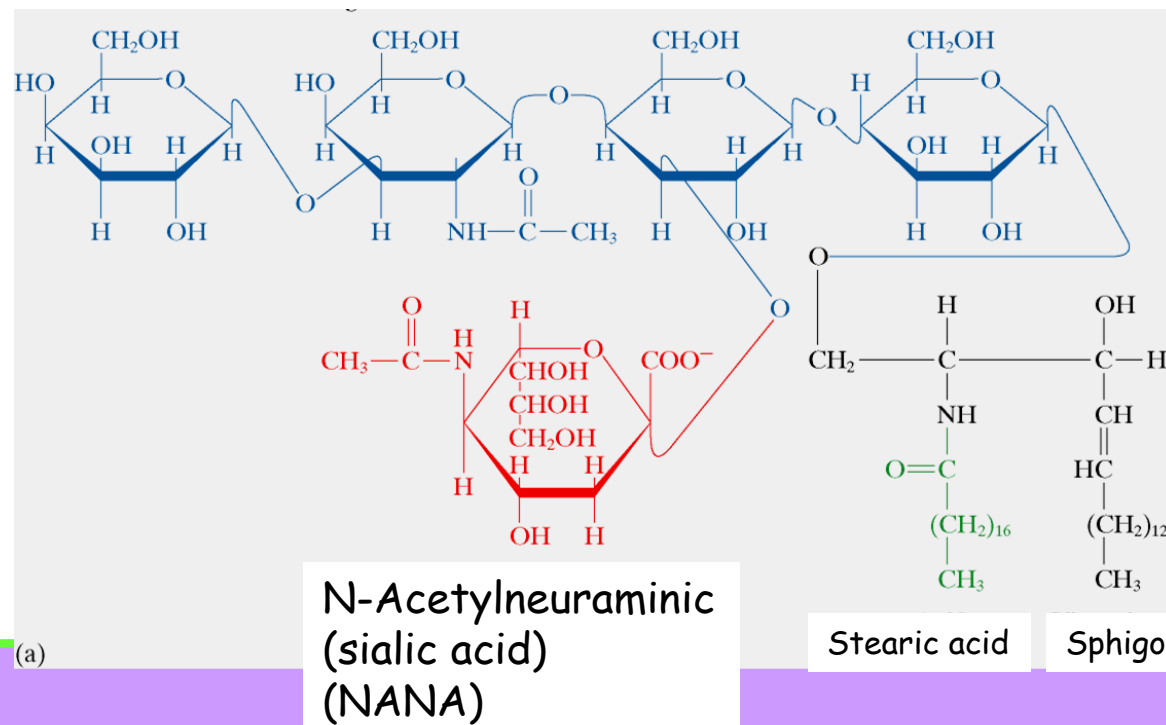


- Gangliosides have considerable physiological and medical significance.
- Receptors for bacterial protein toxins.

GLYCOSPHINGOLIPIDS

3) Gangliosides

- a oligosaccharide containing one or more **sialic acid**
- Have a polar head with negative charge at pH 7



GANGLIOSIDES

Nomenclature

Depends on the number of sialic acids

GM: one sialic acid residue

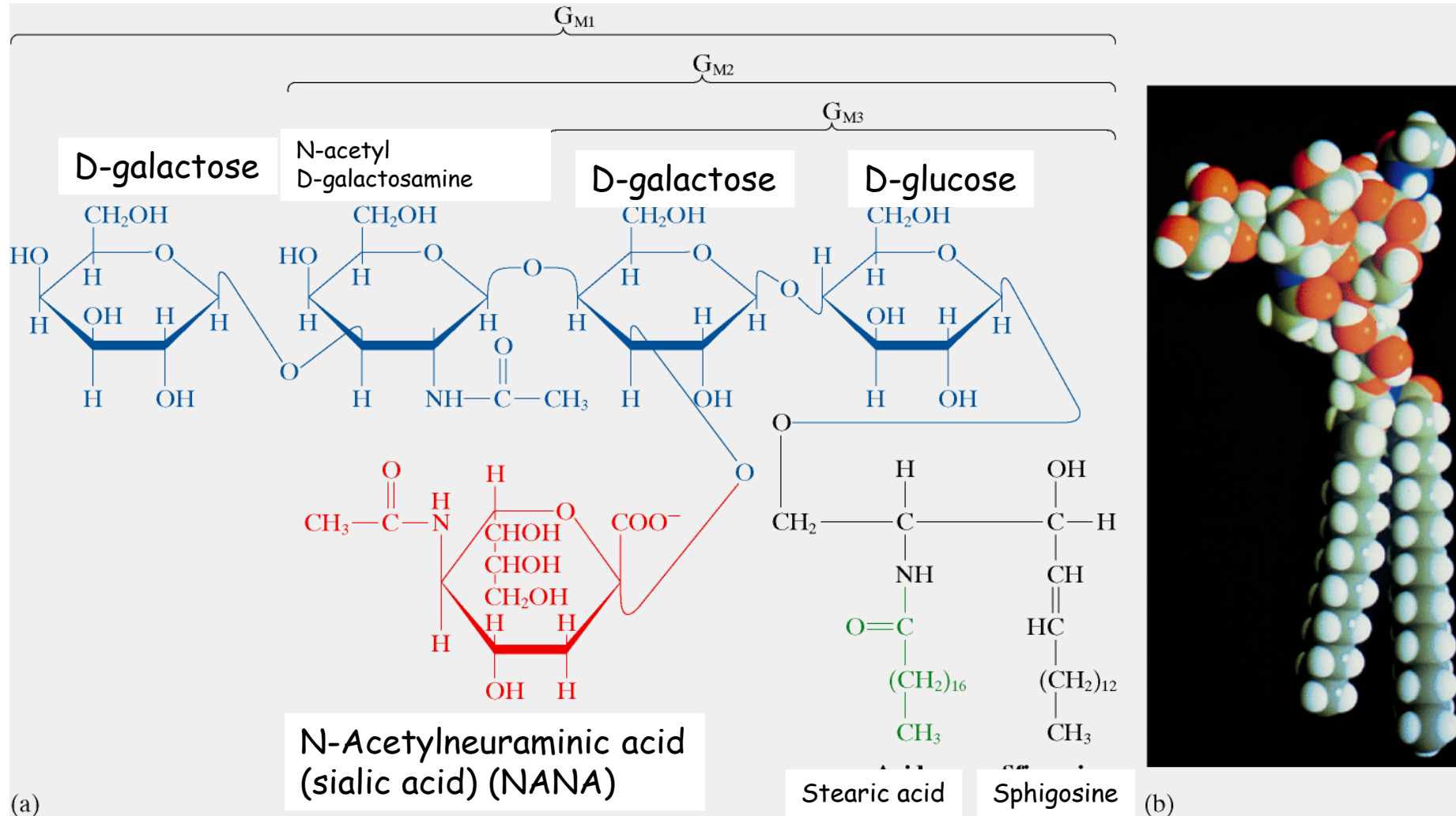
GD: two sialic acid residues

GT: three sialic acid residues

GQ: four sialic acid residues



GANGLIOSIDES

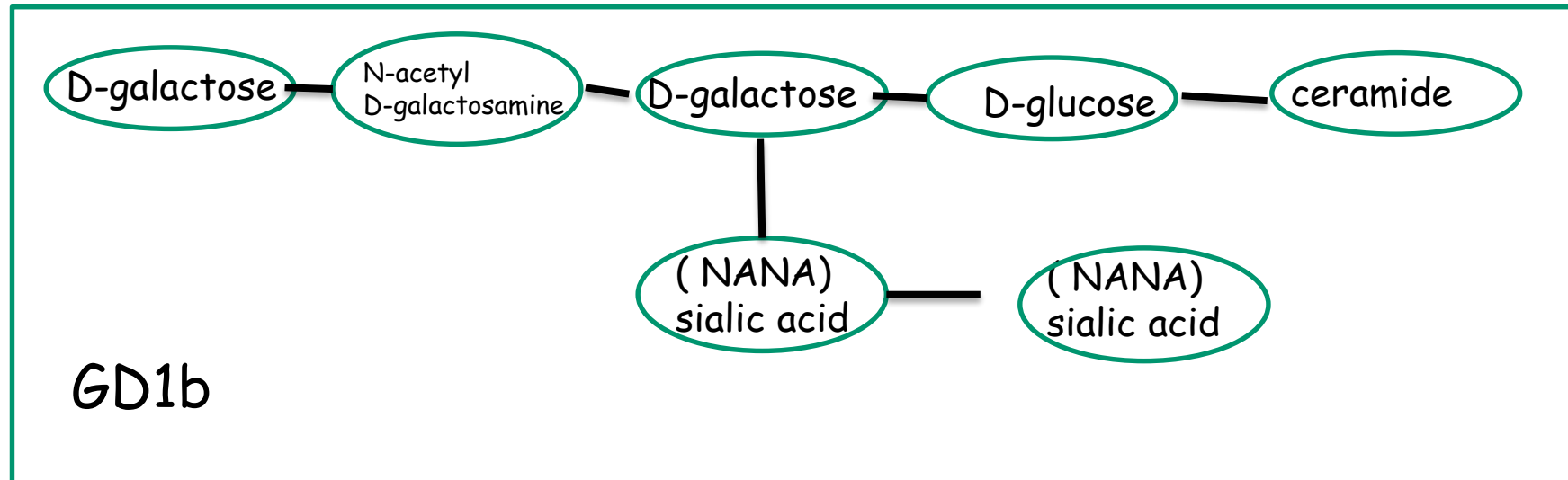
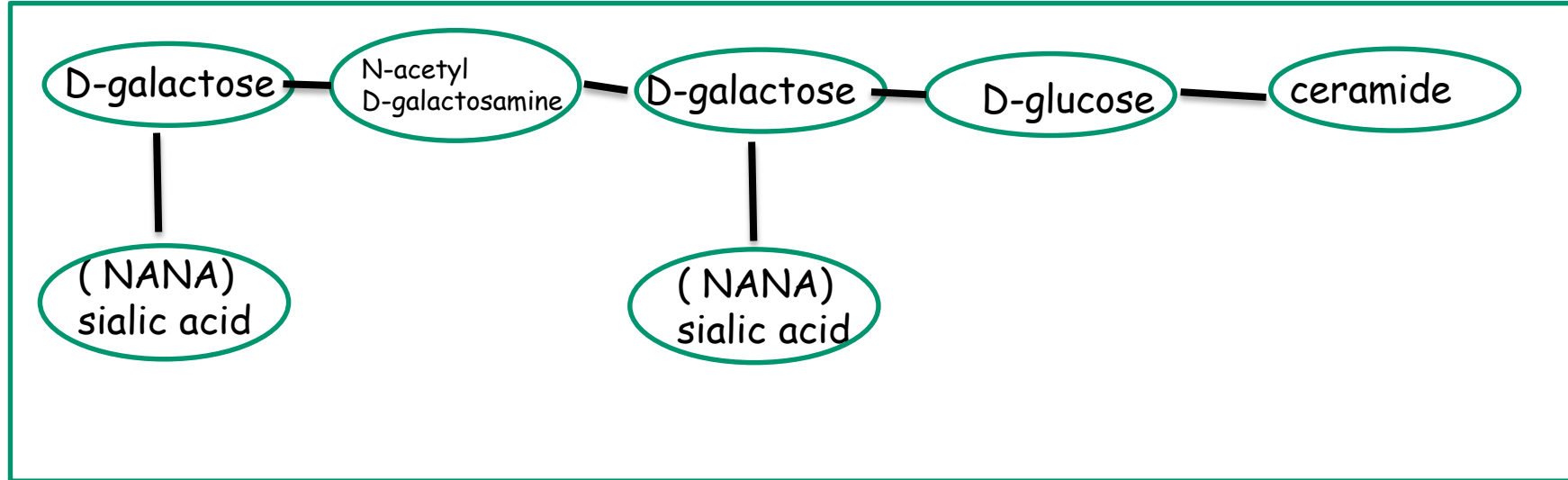


Gangliosides *GM2* e *GM3* differ from *GM1* only by the sequential absences of the terminal D-galactose and N-acetyl D-galactosamine residues.

GANGLIOSIDES

Gangliosides GD1a differ from GD1b on the basis of the sialic acid position

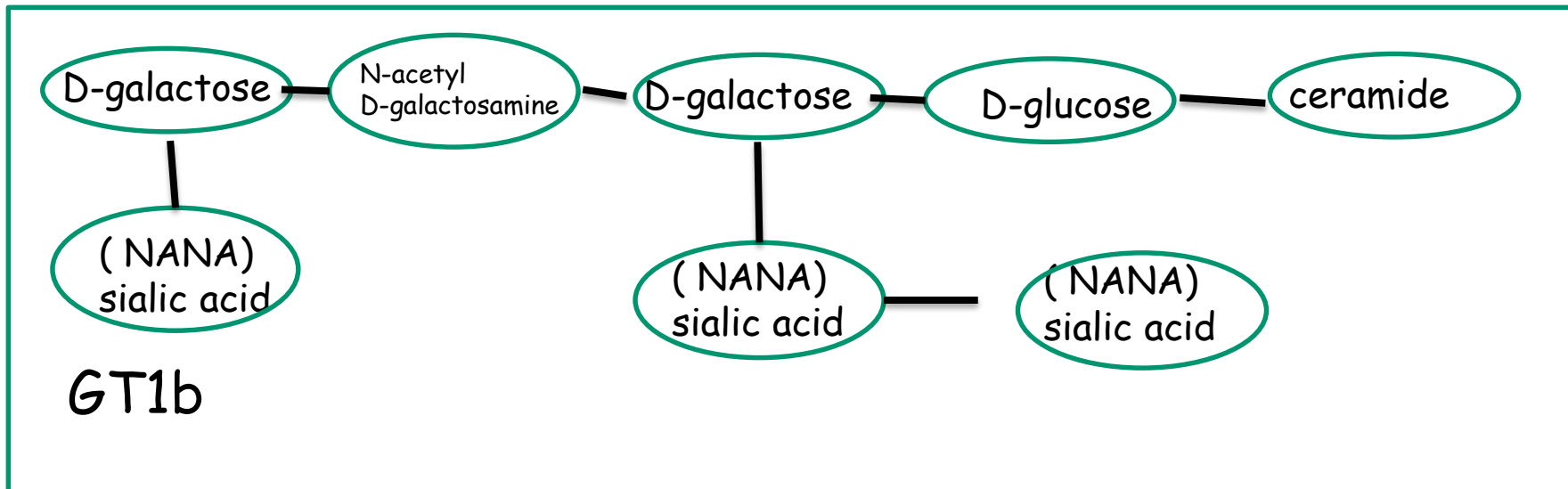
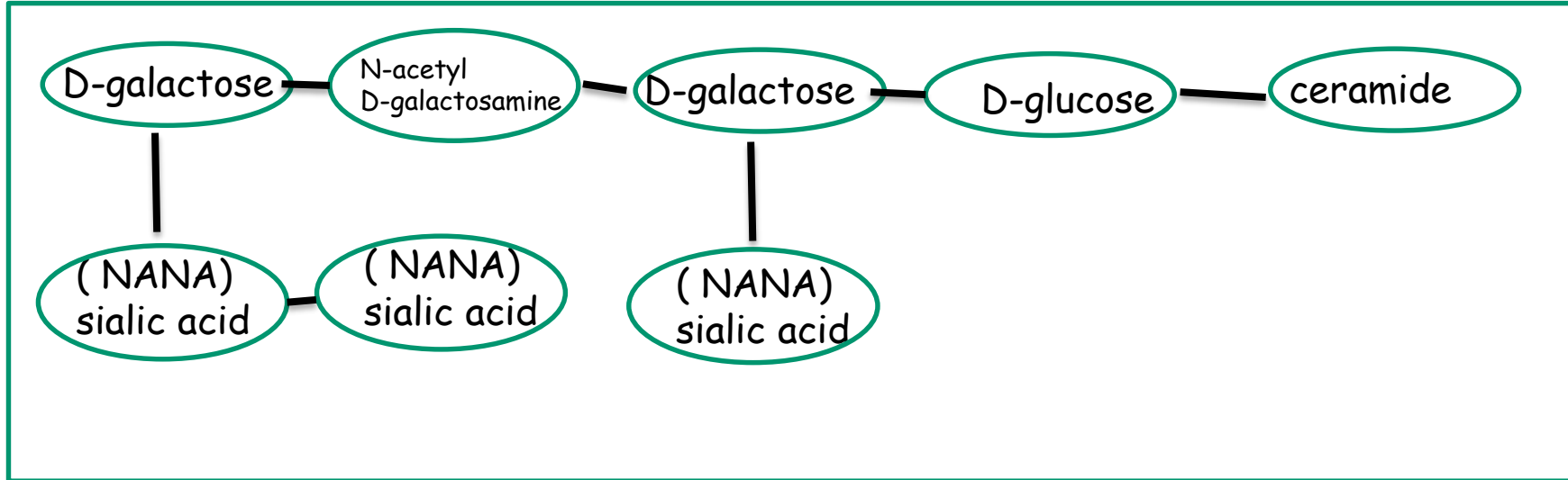
GD1a



GANGLIOSIDES

Gangliosides GT1a differ from GT1b on the basis of the sialic acid position

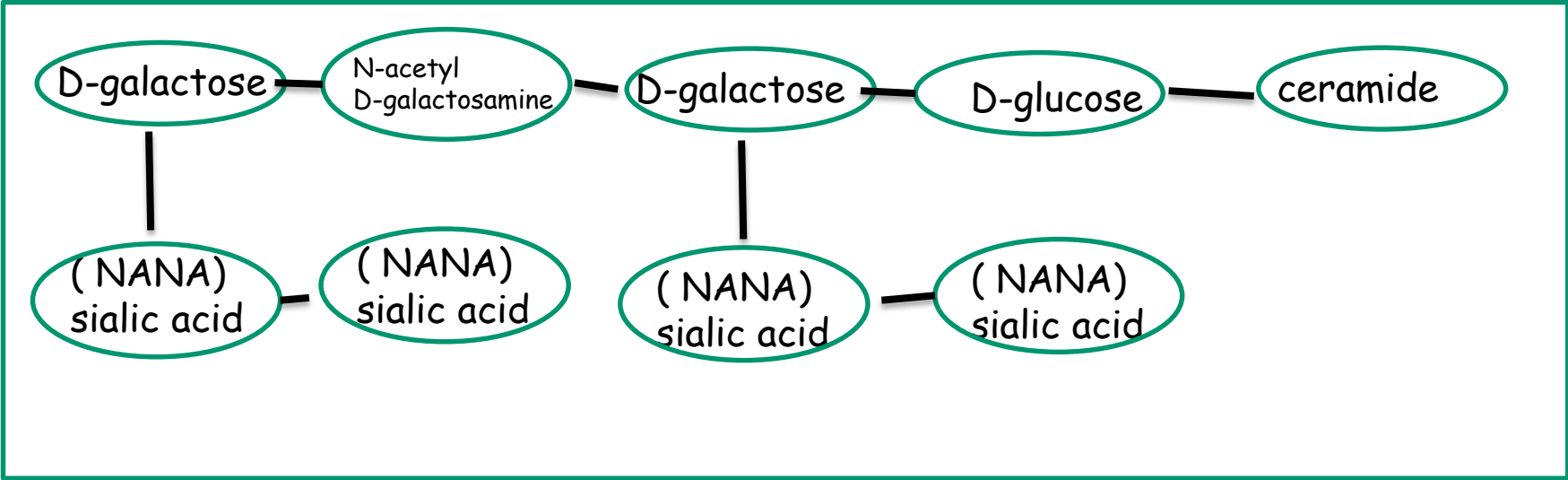
GT1a



GT1b

GANGLIOSIDES

GQ1b



GQ: four sialic acid residues

GANGLIOSIDES

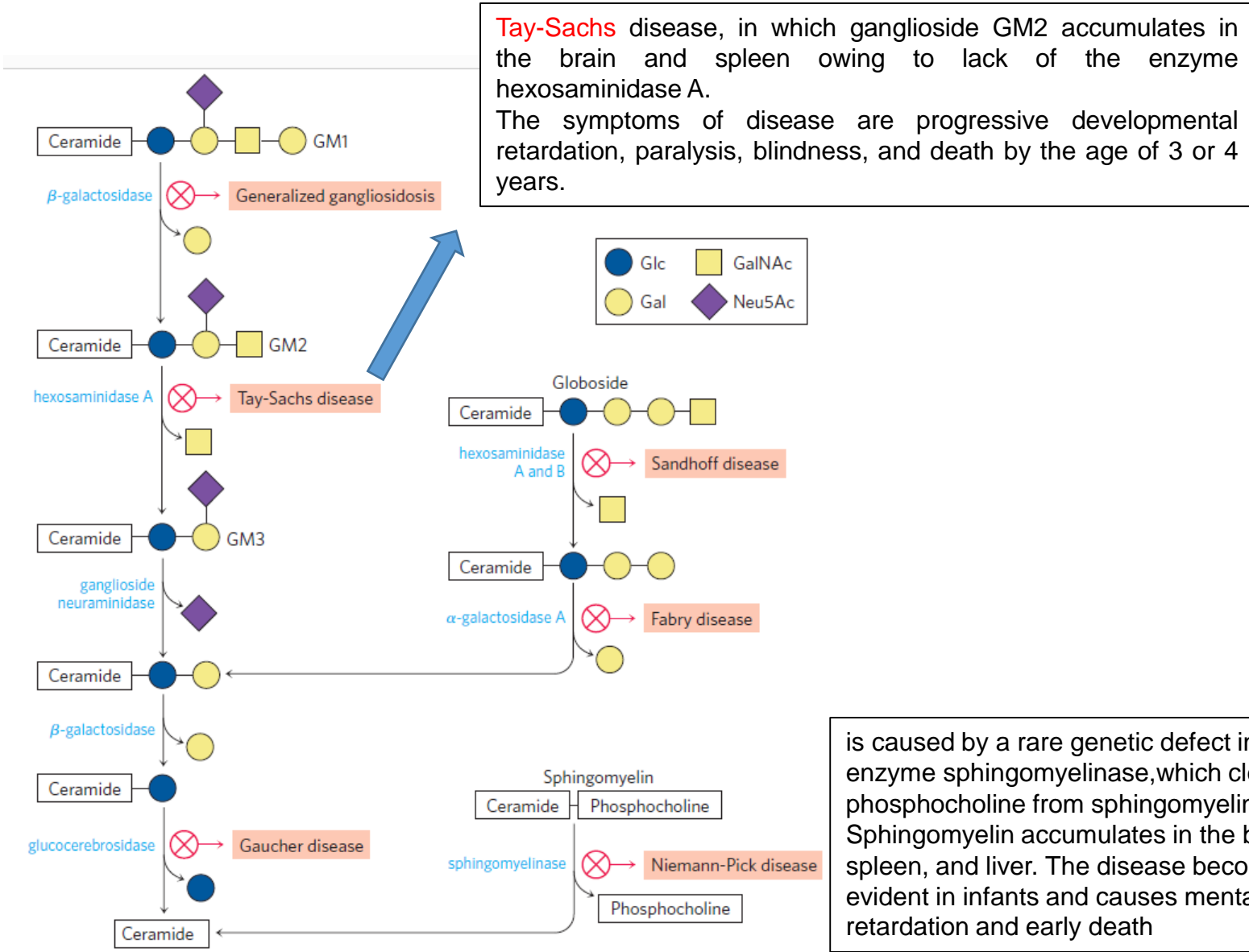
Gangliosides on intestinal mucosal cells bind cholera toxin, an 84-kDa protein secreted by pathogen *Vibrio cholerae*

The toxin stimulates secretion of chloride ions into the gut lumen, resulting in the copious diarrhea of cholera.

Gangliosides may also bind other toxins, such as tetanus toxin, and certain viruses, such as flu virus.

Disease:

- **GM1 Gangliosidosis** (autosomal recessive metabolic disease) characterized by:
 - Impaired psychomotor function
 - Mental retardation
 - Hepatosplenomegaly
 - Death within the first few years of life
- Accumulation of GM1 is due to a marked deficiency of B-galactosidase
-
- **Tay-Sachs disease** (accumulation of GM2 in brain)



is caused by a rare genetic defect in the enzyme sphingomyelinase, which cleaves phosphocholine from sphingomyelin. Sphingomyelin accumulates in the brain, spleen, and liver. The disease becomes evident in infants and causes mental retardation and early death

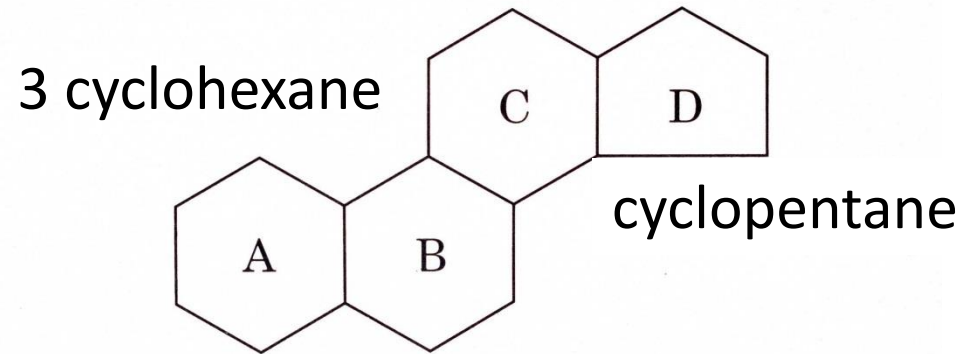
Steroids

Cholesterol and their derivatives:

cholesterols esters
vitamin D
hormones
bile acids

Cholesterol

Component of cell membranes
Starting molecule for the
synthesis of other steroid



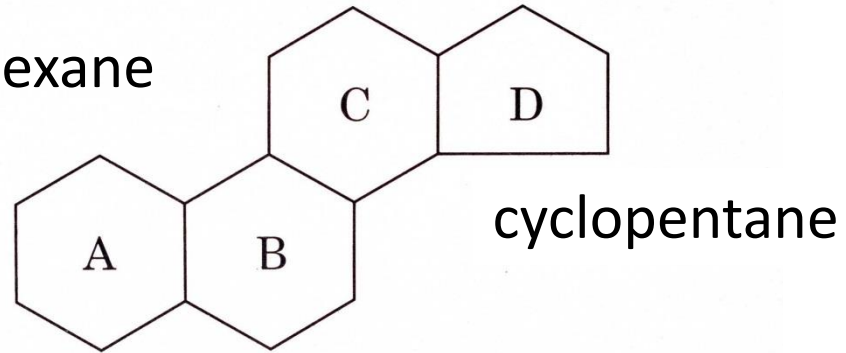
Cyclopentanoperhydrophenanthrene

- Structural lipids of membrane
- Precursors of bile acids
- Metabolic precursors of steroid hormones and vitamins
- Cofactors in more reactions

Steroids

Steroids are derivatives of cyclopentanoperhydrophenanthrene

3 cyclohexane



Cyclopentanoperhydrophenanthrene

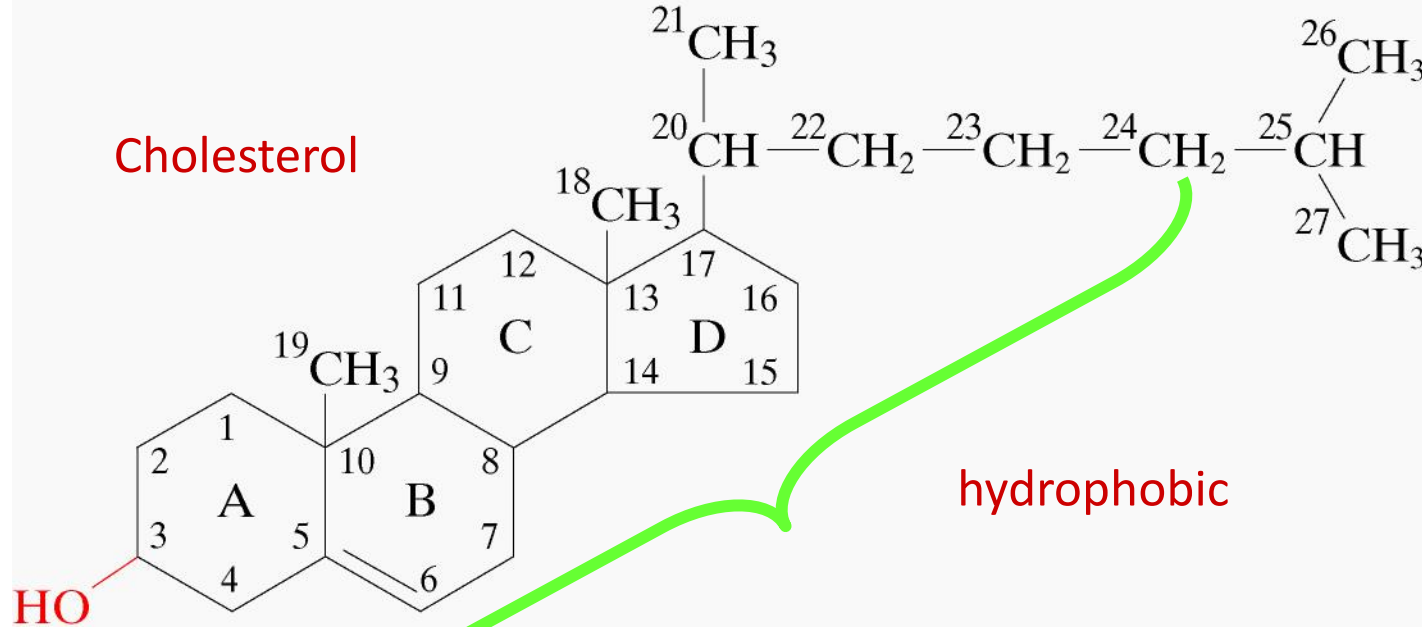
Steroids

Derivatives of Cyclopentanoperhydrophenanthrene

Steroid nucleus is rather rigid, Planar structure

(a)

Structural lipids of cellular membranes



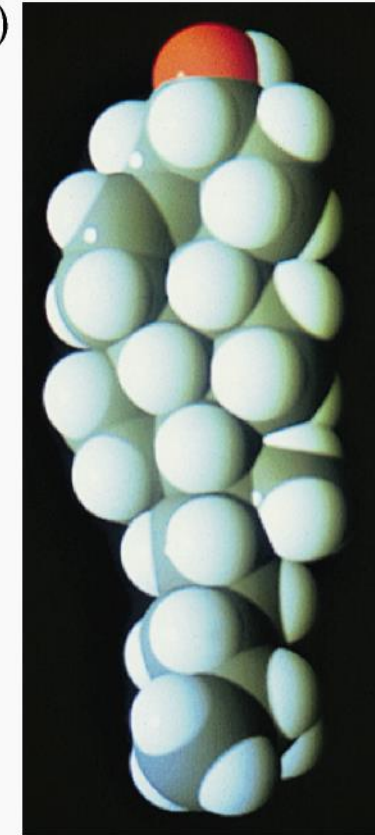
Polar head

Free cholesterol is esterified by fatty acid

Sterol abundant in animals cells membranes
Anphipatic

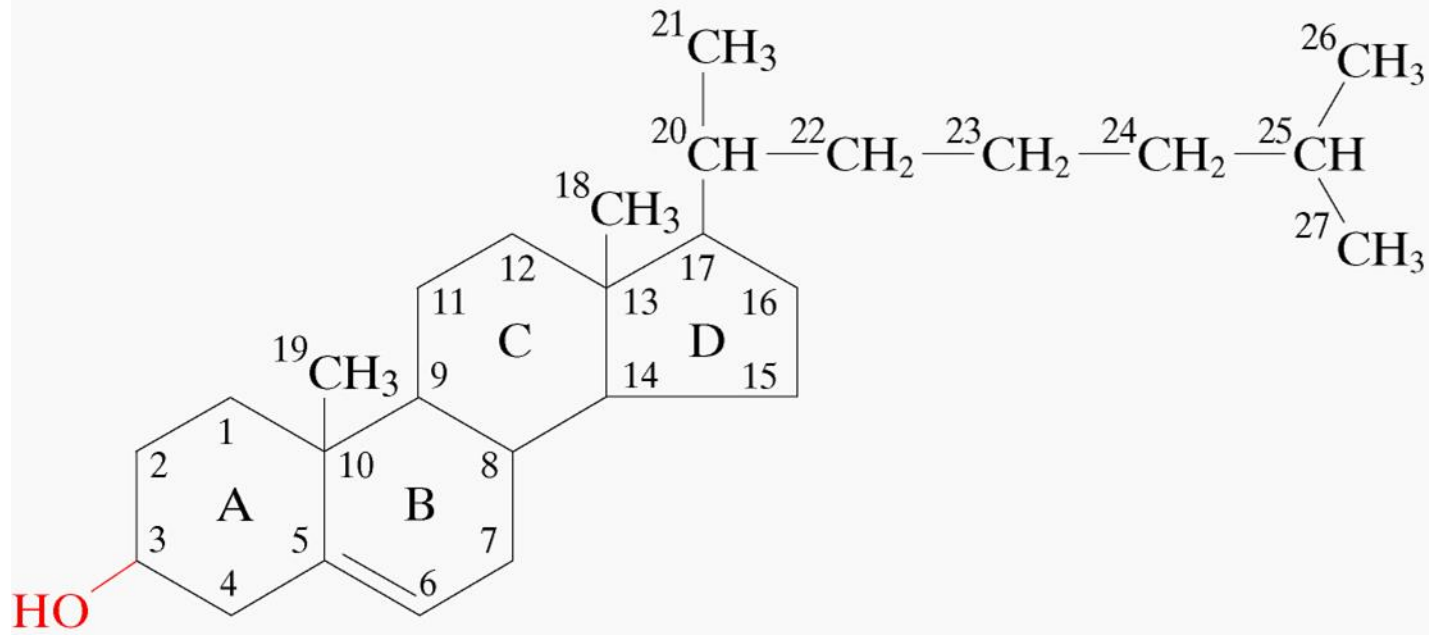
hydrophobic

(b)

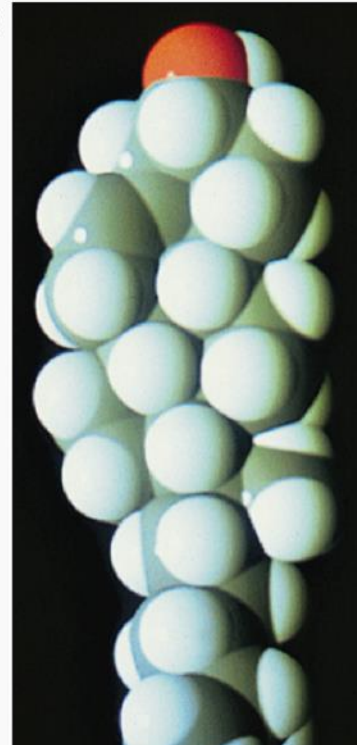


Plants= stigmasterol
mushrooms= ergosterol

(a)



(b)



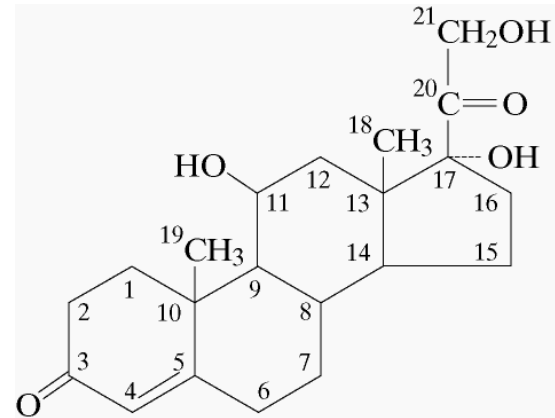
Constituent of cell membranes

It is clearly essential to life, yet its deposition in arteries has been associated with cardiovascular disease and stroke, two leading causes of death in humans

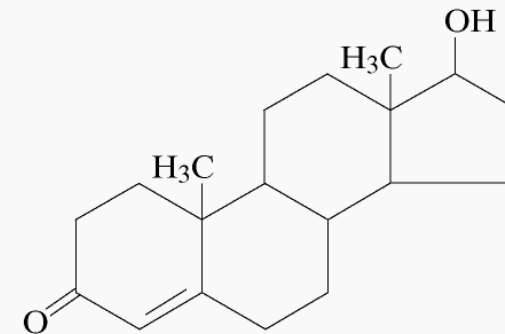
Steroids are precursors of Steroid hormones

Check the functionality of different organs.

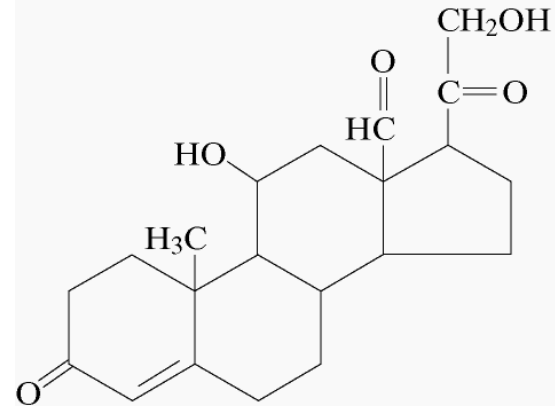
Regulate gene expression, metabolism and reproduction.



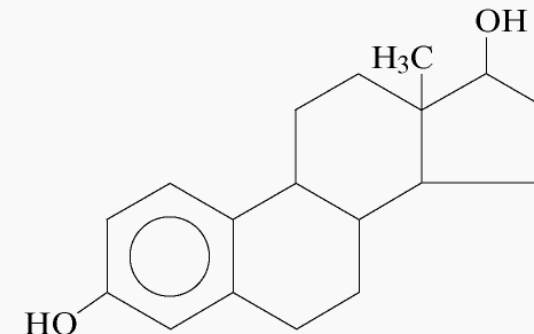
Cortisol



Testosterone



Aldosterone

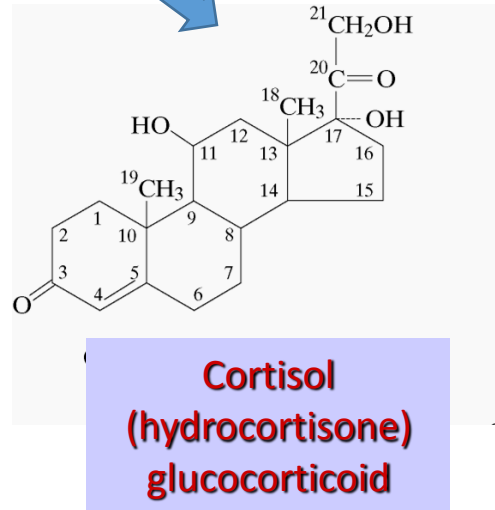
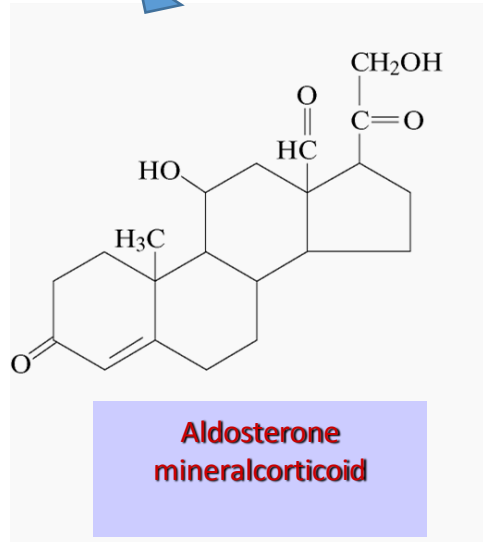


β-Estradiol

Steroid hormones

Adrenal cortex produce at least 50 different adrenocortical steroid

In the adrenal cortex

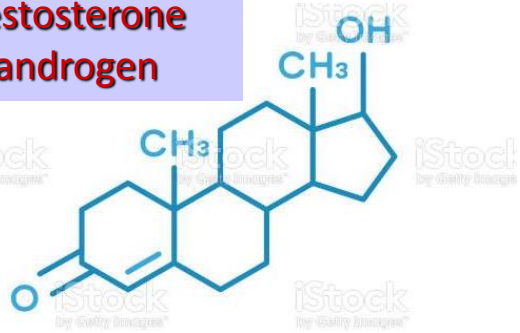


Classified according to the physiological responses they evoke:

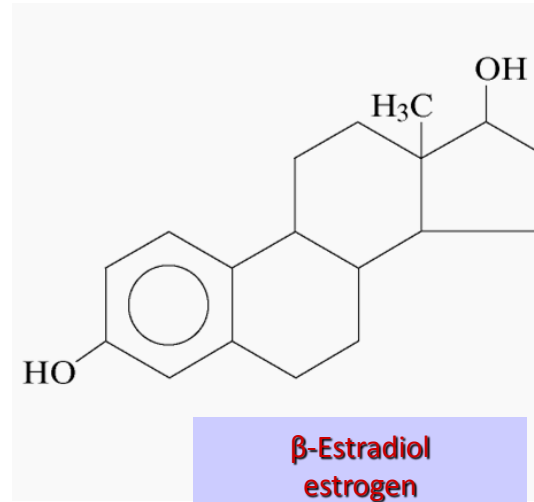
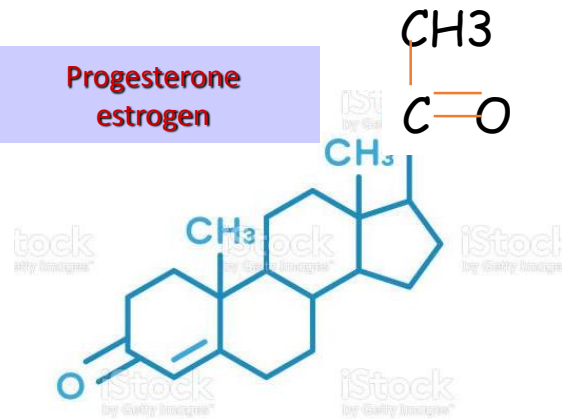
- Glucocorticoids
 - Mineralcorticoids
 - Androgens and estrogens
- Mineralcorticoids (to regulate the excretion of salt and water by the kidney)
 - Glucocorticoids (affect carbohydrates, protein and lipid metabolism and influence a wide variety of other vital functions including inflammatory reaction and the capacity to cope with stress)

- **Androgens and estrogens**
(produced by testes and ovaries)
- (affect sexual development and function.)

**testosterone
androgen**

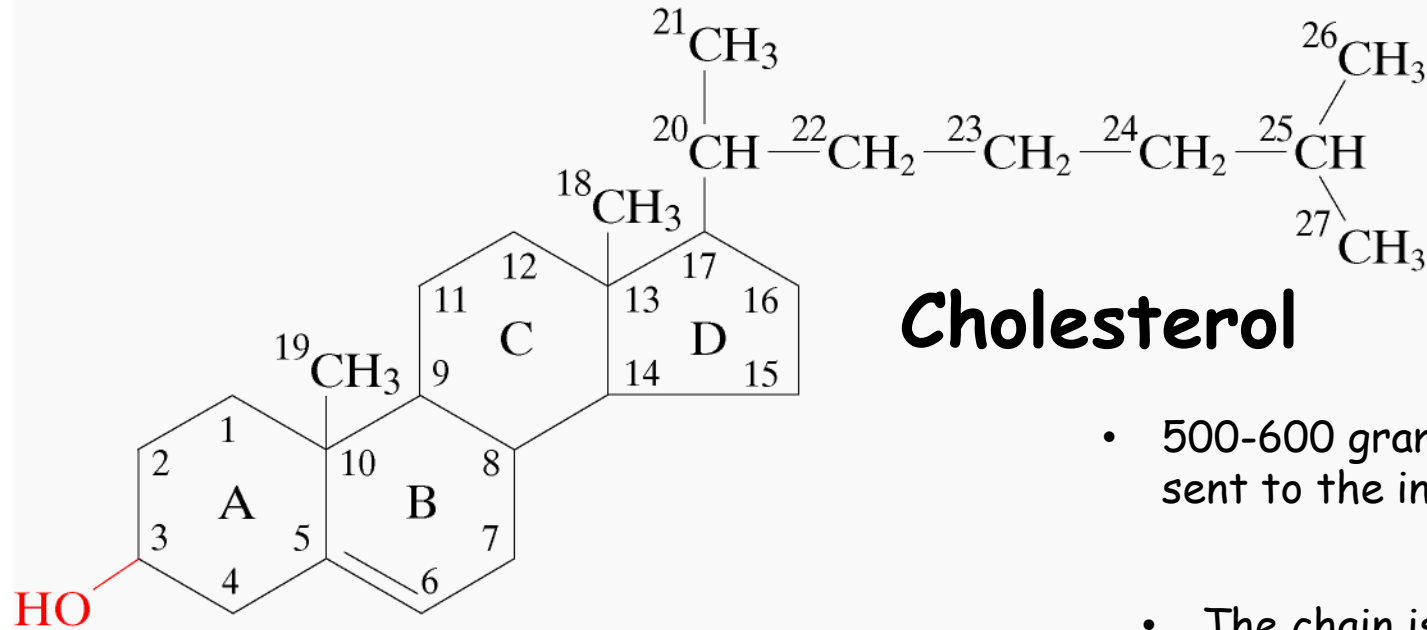


**Progesterone
estrogen**



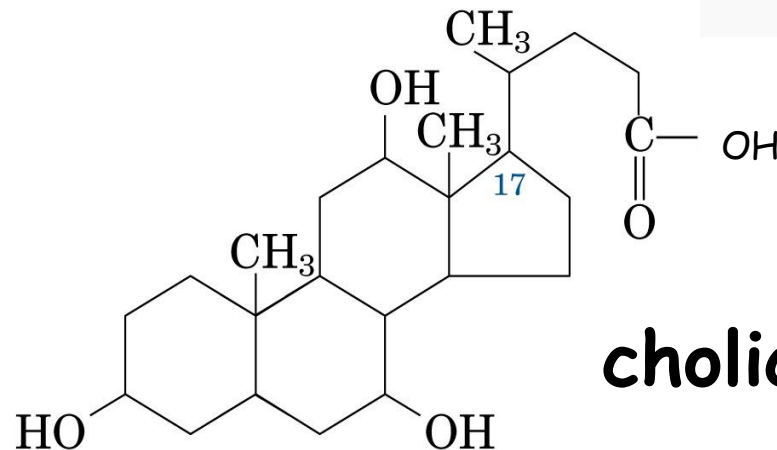
**β-Estradiol
estrogen**

Bile salts: derived from cholesterol



Cholesterol

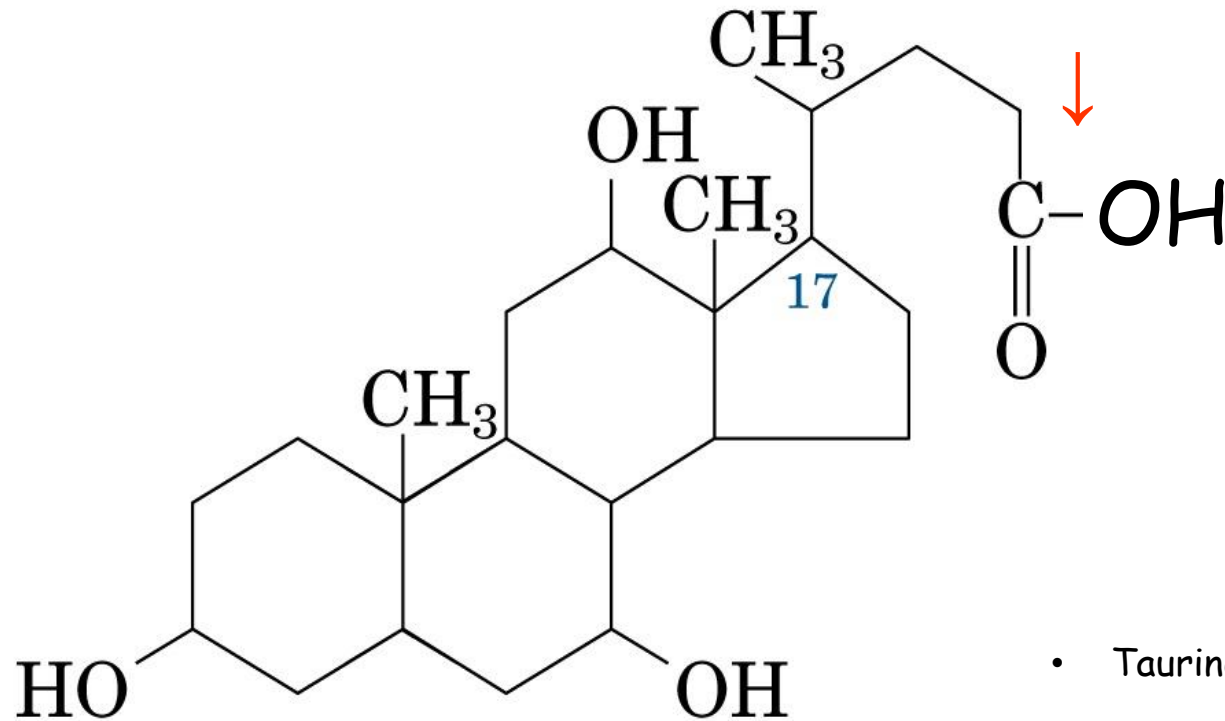
- 500-600 grams of cholesterol per day is sent to the intestine by bile
- The chain is shortened and carboxylated



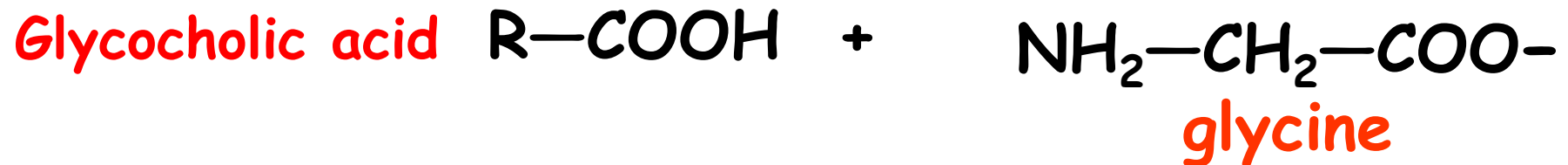
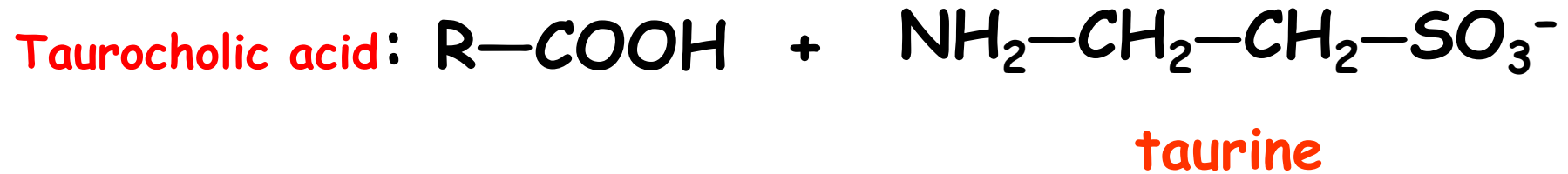
cholic acid

- A glycine or taurine molecule is bound to the COOH

Bile salts: cholic acid + taurine or glycine



- Taurine is an amine with a sulfonic group



Bile salts: cholic acid + taurine or glycine

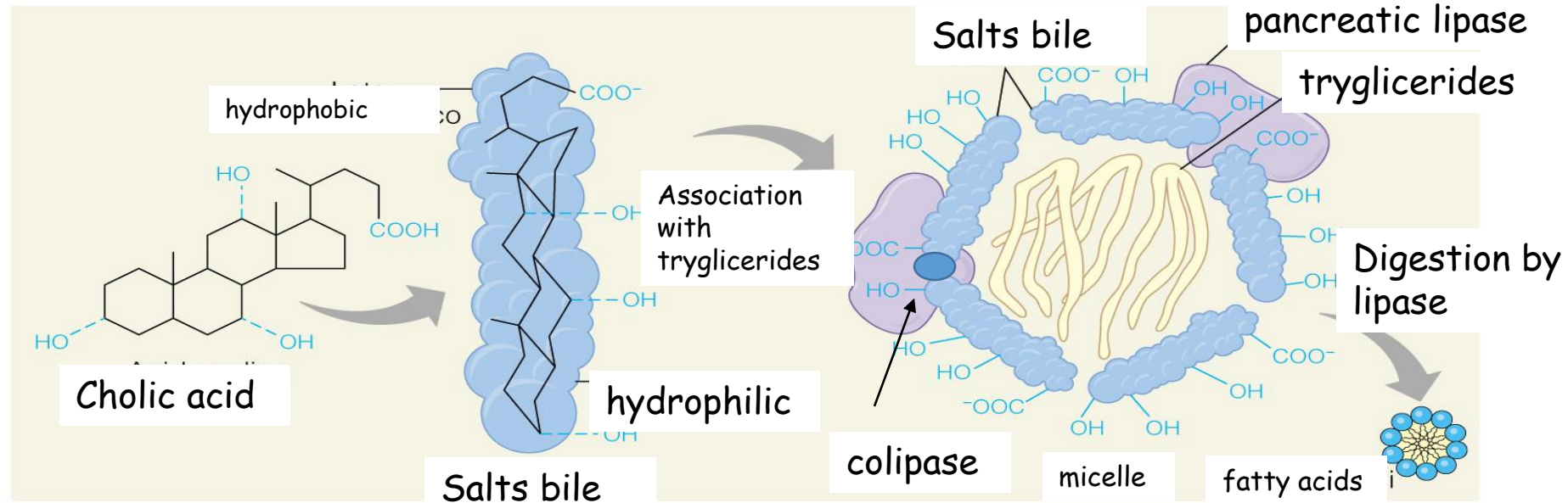
Taurocholic acid

Glycocholic acid

Function:

- Release during digestion into the duodenum.
- Solubilize ingested fat and fat-soluble vitamins, facilitating their digestion and absorption.

Emulsifying capacity of bile salts



- Polar groups are located in the same face as the steroid nucleus
- Polar face interacts with the solvent
- Apolar face makes contact with food lipids (tryglicerides and esters of cholesterol)

- **Action:**

Salts bile Fragment the large lipid drops, increasing the surface accessible to the pancreatic lipase.

Lipase converts triglycerides into free fatty acids

Colipase is needed to anchor the lipase to the droplet surface

Essential fatty acids or PUFA

Not synthesized in the body and should be supplied through diet

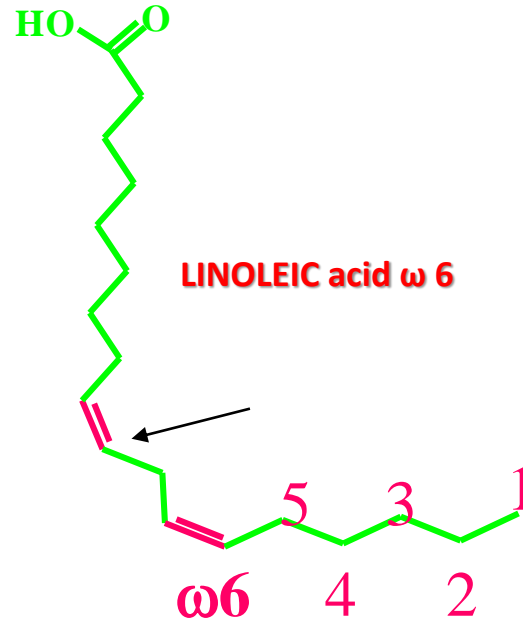
Not synthesized in human body due to lack of the **desaturase enzyme** which introduces double bonds beyond 9th and 10th carbon atom

Poly-unsaturated

Contain more than one double bond

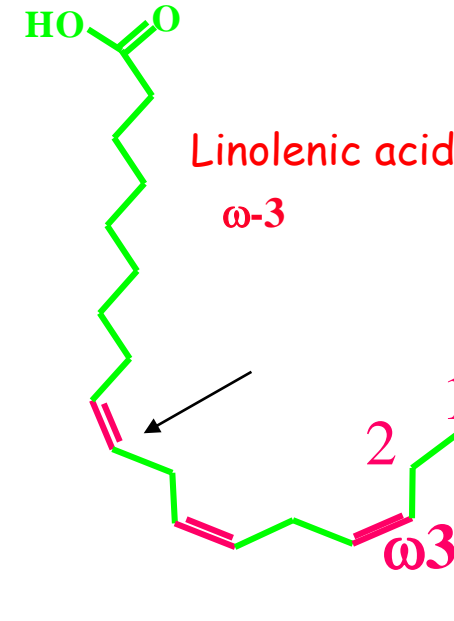
Essentially fatty acids :

Linoleic acid,
linolenic acid and
arachidonic acid.



Linoleic acid
(18:2 $\Delta^{9,12}$)

vegetable oils



α-Linolenic acid
(18:3 $\Delta^{9,12,15}$)

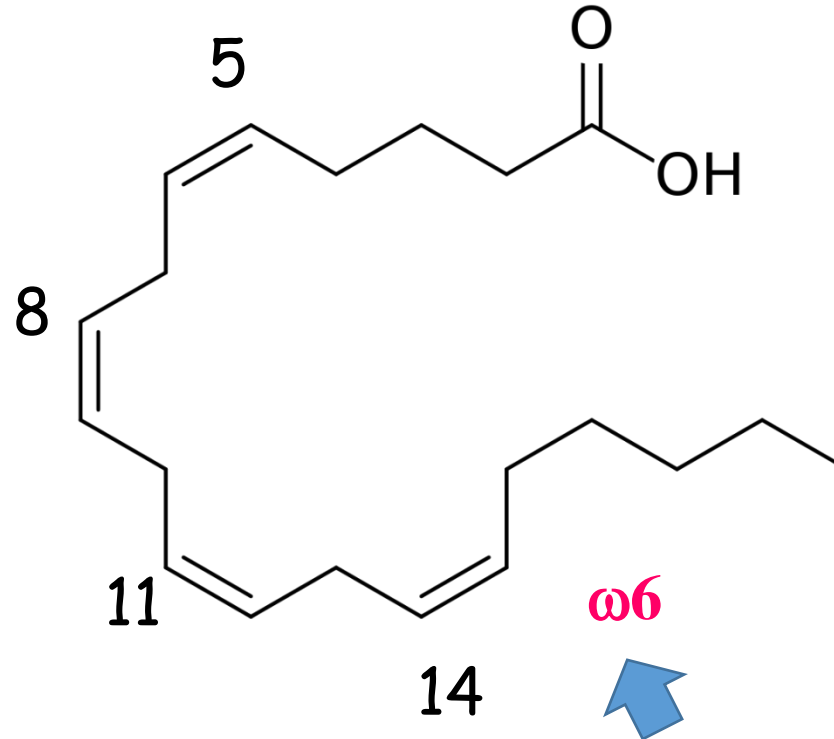
Fishes's fat

Essential fatty acids or PUFA

Arachidonic acid (Contains 20 Carbon atoms)

(ARA, 20:4 $\Delta^{5,8,11,14}$)

**Cis-,cis-,cis-,cis-5,8,11,14
Icosatetraenoic acid**

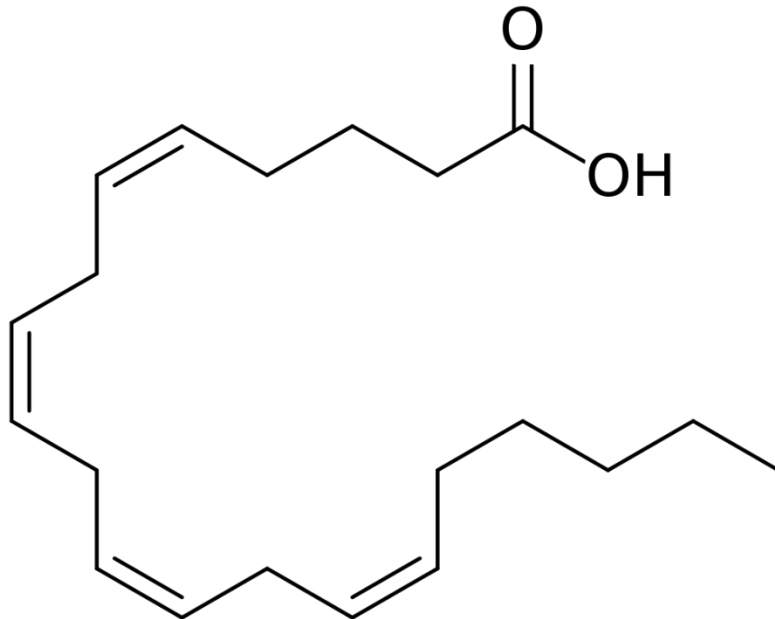


number the chain on the opposite side with respect to the carboxyl group

Eicosanoids

substances that act only on cells near the point of hormone synthesis instead of being transported in the blood to act on cells in other tissues or organs.
(Contains 20 Carbon atoms)

- Eicosanoids derive from **arachidonic acid (essential fatty acid)**
- Arachidonic acid should be supplied through diet.
- Our cells can not synthesis .
- We can produce it, starting by linoleic acid (**essential fatty acid it too**)
They control numerous important homeostatic and inflammatory processes and they are locally acting signaling molecules.

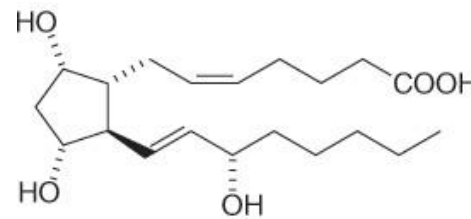


Arachidonic acid
(ARA, 20:4 $\Delta^{5,8,11,14}$)

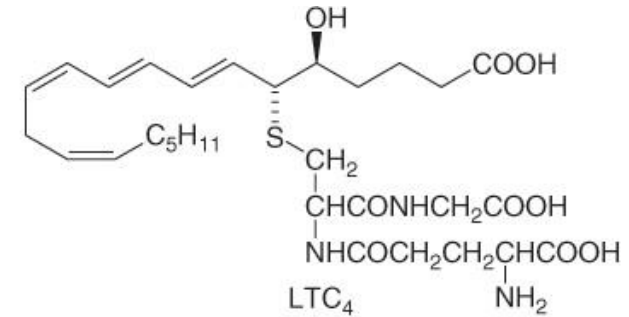
Arachidonic acid is the precursor of all eicosanoid.

Eicosanoids can be divided into several important classes:

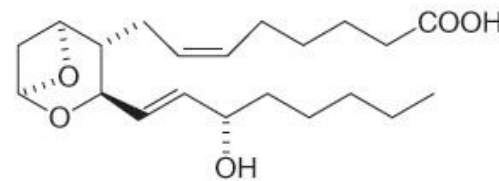
- prostaglandins,
- leukotrienes
- tromboxane
- prostacyclin
- and others.



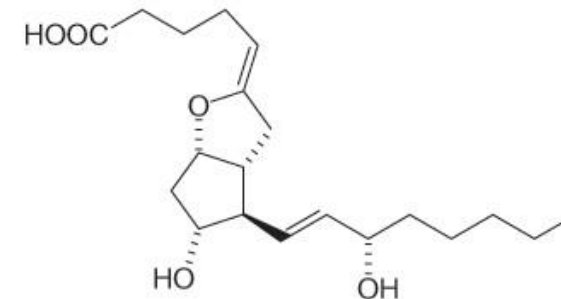
PGF_{2α}
a prostaglandin



LTC₄
a leukotriene



TA₄
a thromboxane



PGI₂
a prostacyclin

- **Prostaglandins** are important in communication between cells

by Bengt Samuelsson
and
Sune Bergström.

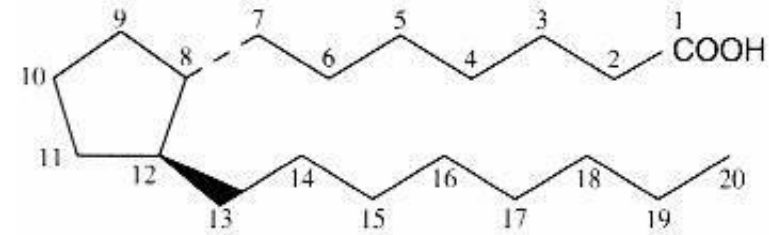
- Prostaglandins (PGs) called from prostate gland (first identified in human semen in 1930)

Contain a five-carbon ring originating from the chain of arachidonic acid.

Their name derives from the prostate gland, the tissue from which they were first isolated.

Two groups of prostaglandins were originally defined:

- PGE (ether-soluble)
- PGF (*fosfat* (Swedish for phosphate) buffer-soluble).



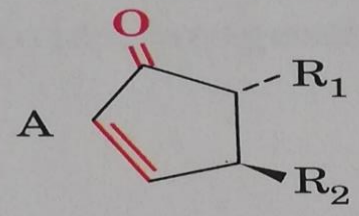
Prostanoic Acid

Prostaglandins are all derivatives of prostanoic acid (C₂₀) in which carbon atoms C₈ to C₁₂ comprise a cyclopentane ring

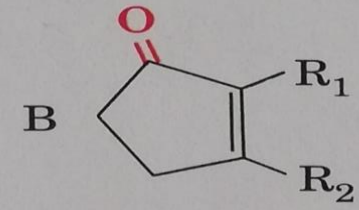
Five-carbon ring

- Prostaglandins **A** through **I** differ in the substituents on the cyclopentane ring

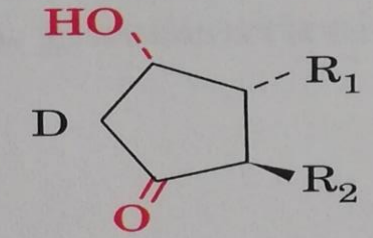
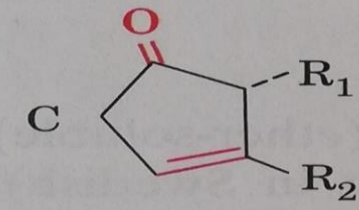
PGA
(Unsaturated ketons)



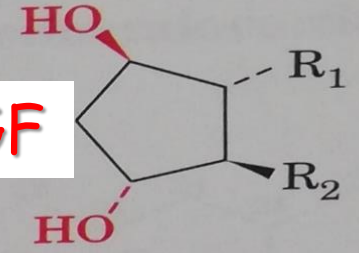
PGB
(Unsaturated ketons)



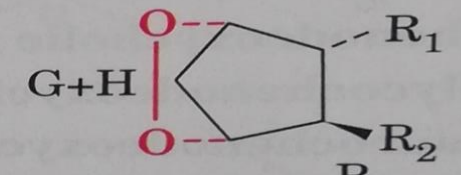
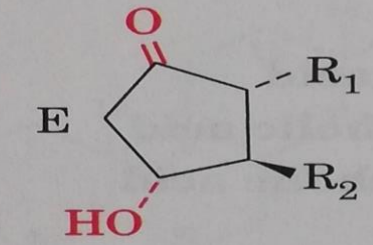
PGC
(Unsaturated ketons)



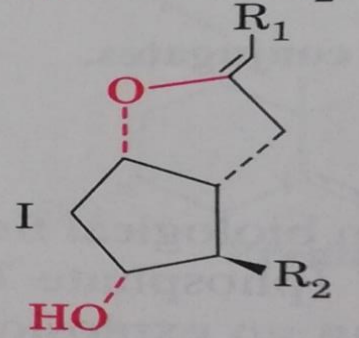
PGF



(1-3 diols)



PGD and E
(hydroxyketons)



(c)

PGE (ether-soluble) and PGF (fosfat buffer-soluble).

Each group contains numerous subtypes, named PGE1, PGE2, PGF1,

Prostaglandins are produced in:



Uterine level

promote muscle contraction during menstruation and labor.
(conception, luteolysis, menstruation, parturition)

Renal level



KIDNEYS

regulate perfusion and filtration processes

Prostaglandins have many functions:



STOMACH

Gastric level

protect the stomach wall from the action of hydrochloric acid.

Vascular level

in the processes of homeostasis and platelet aggregation.

- **Leukotrienes** First identification in leucocytes

They are powerful biological signals.

Synthesized by a variety of white blood cells, mast-cells, as well as lung , spleen, brain and heart.

Leukotrienes contribute to the asthmatic and allergic reactions (i.e caused by pollen)-

They act by supporting and **amplifying inflammatory processes**

They promote the **contraction** of the smooth muscle of the lung where they can cause **asthma attacks**, if produced in **excessive quantities**

Treatment of asthma used drugs such as **prednisone** that inhibit the synthesis of leukotrienes or their receptors.



Arachidonic acid



Lipoxygenase enzyme (LOX)

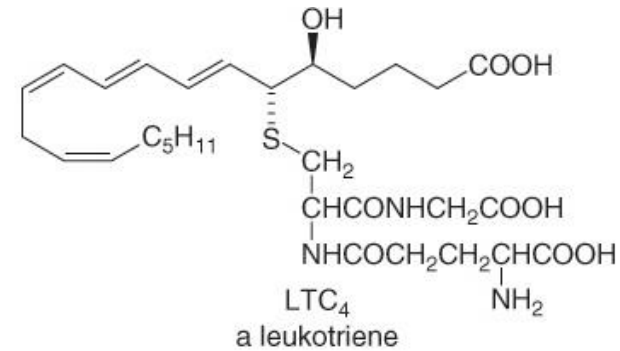
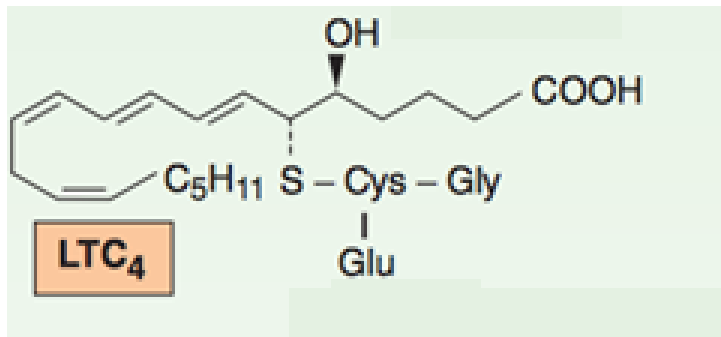
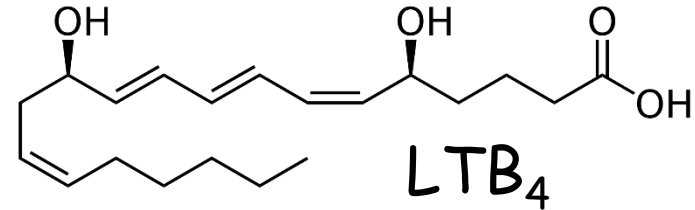
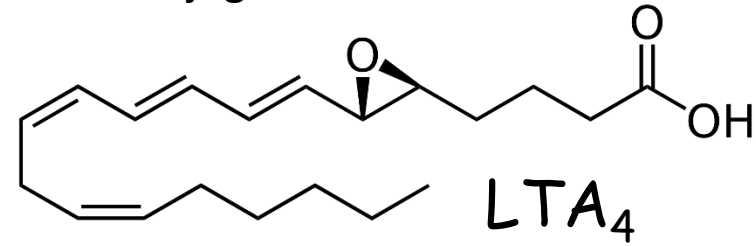
- Leukotrienes

Leukotrienes are indicated with letters LT.

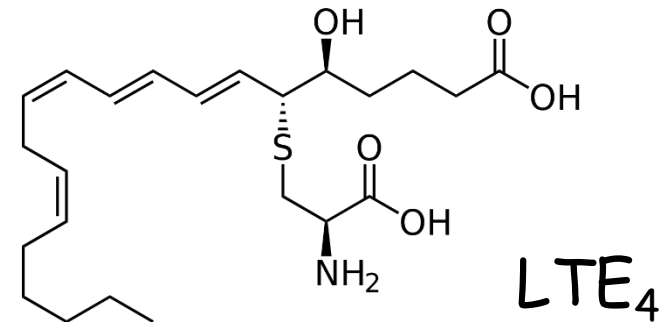
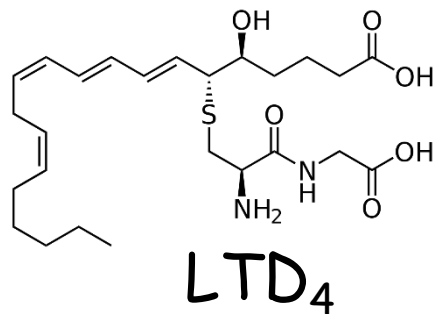
- A, B, C, D and E

Four double bonds, three of them conjugated. Common property of A₄, B₄, C₄, D₄, and E₄.

contain three conjugated double bonds.

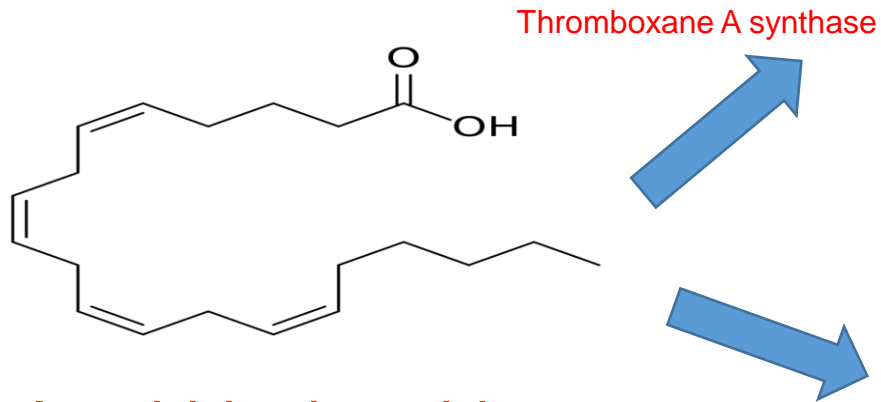


C6 (3 aminoacid bonded)



- **Tromboxane:** indicated with letters TX.

Structure:
an oxygen incorporated to form a six-membered ring



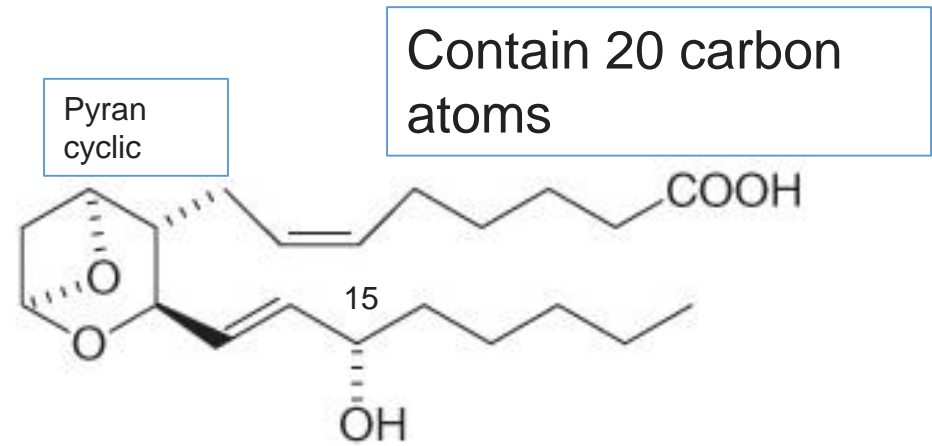
Arachidonic acid

Tromboxane

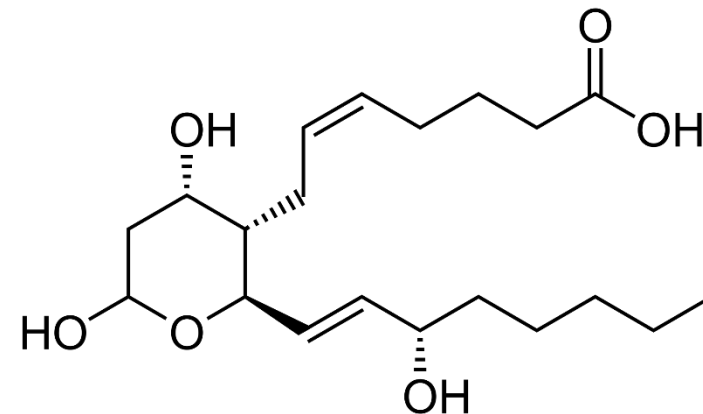
vasoconstrictor

potent hypertensive agent

they are produced by platelets and are involved in the formation of the clot



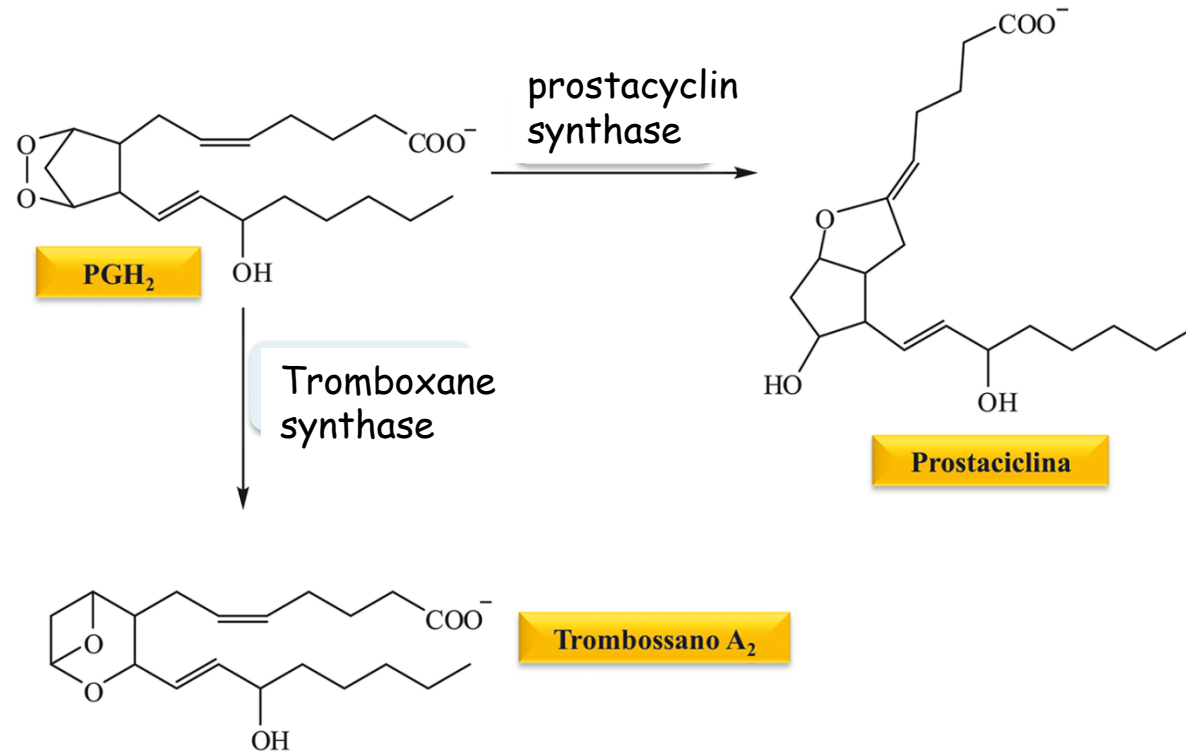
TA2.



TB2.

- Prostacyclin (PGI₂)

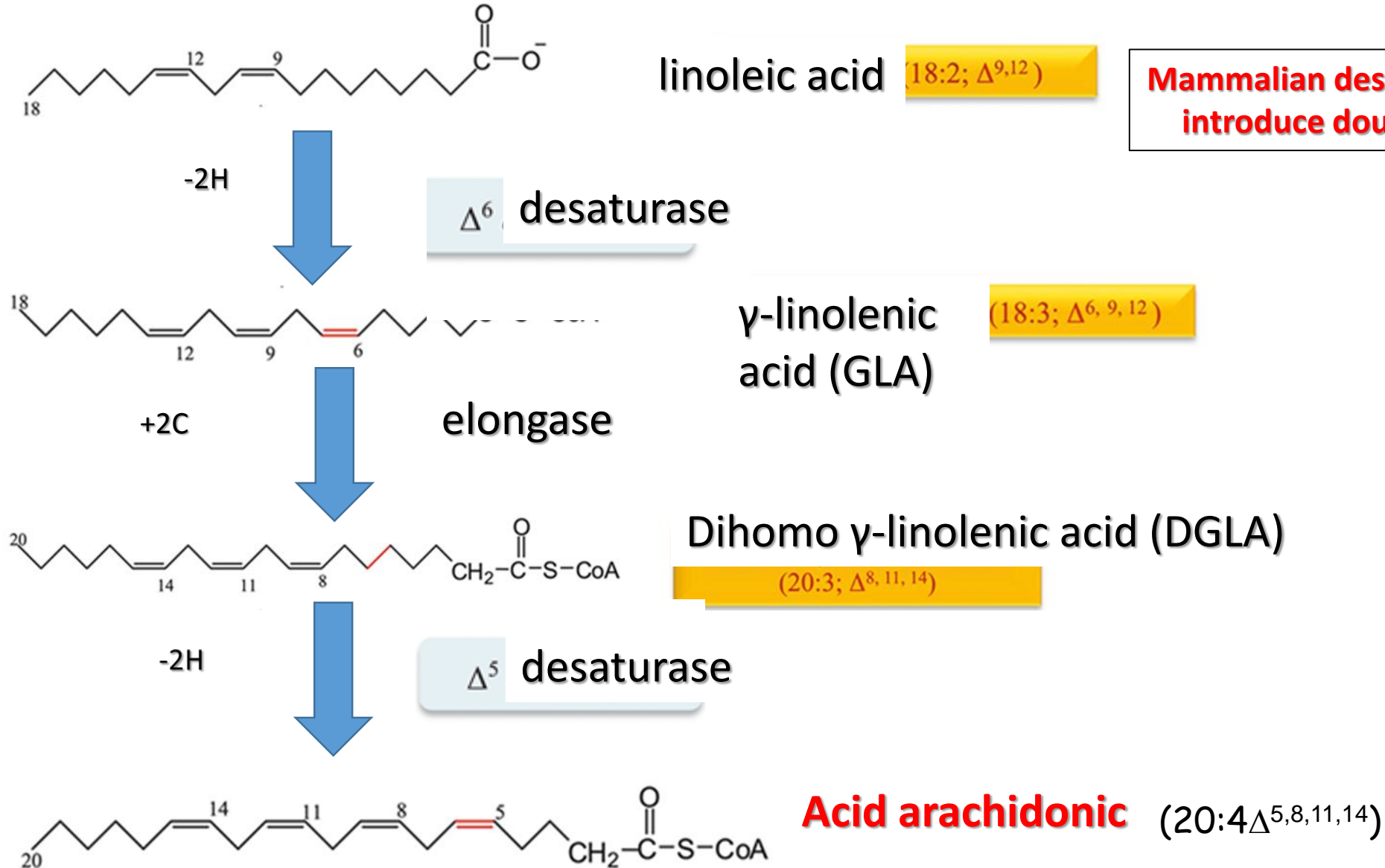
They are mainly found in the vessel wall, (**inhibits platelet aggregation and acting as a vasodilator**)



- They are found mainly in platelets(**promotes platelet aggregation and vasoconstriction**)

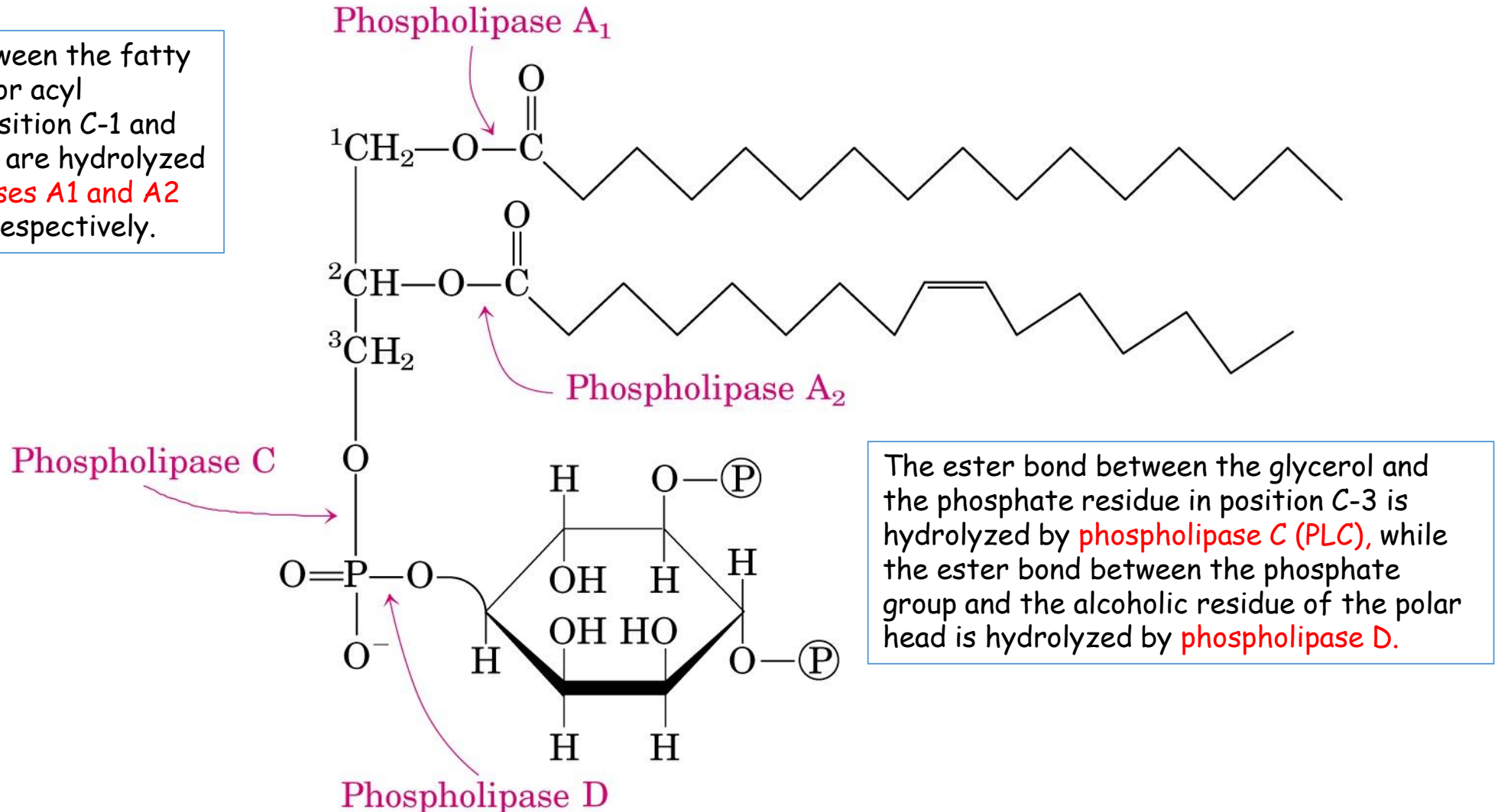
Arachidonic acid Synthesis

ARA is synthesized from linoleic acid by desaturation with a Δ^6 desaturase to yield γ -linolenic acid (GLA) followed by elongation and a second desaturation (by Δ^5 desaturase)



Phospholipids and Sphingolipids Are Degraded in Lysosomes by Phospholipases

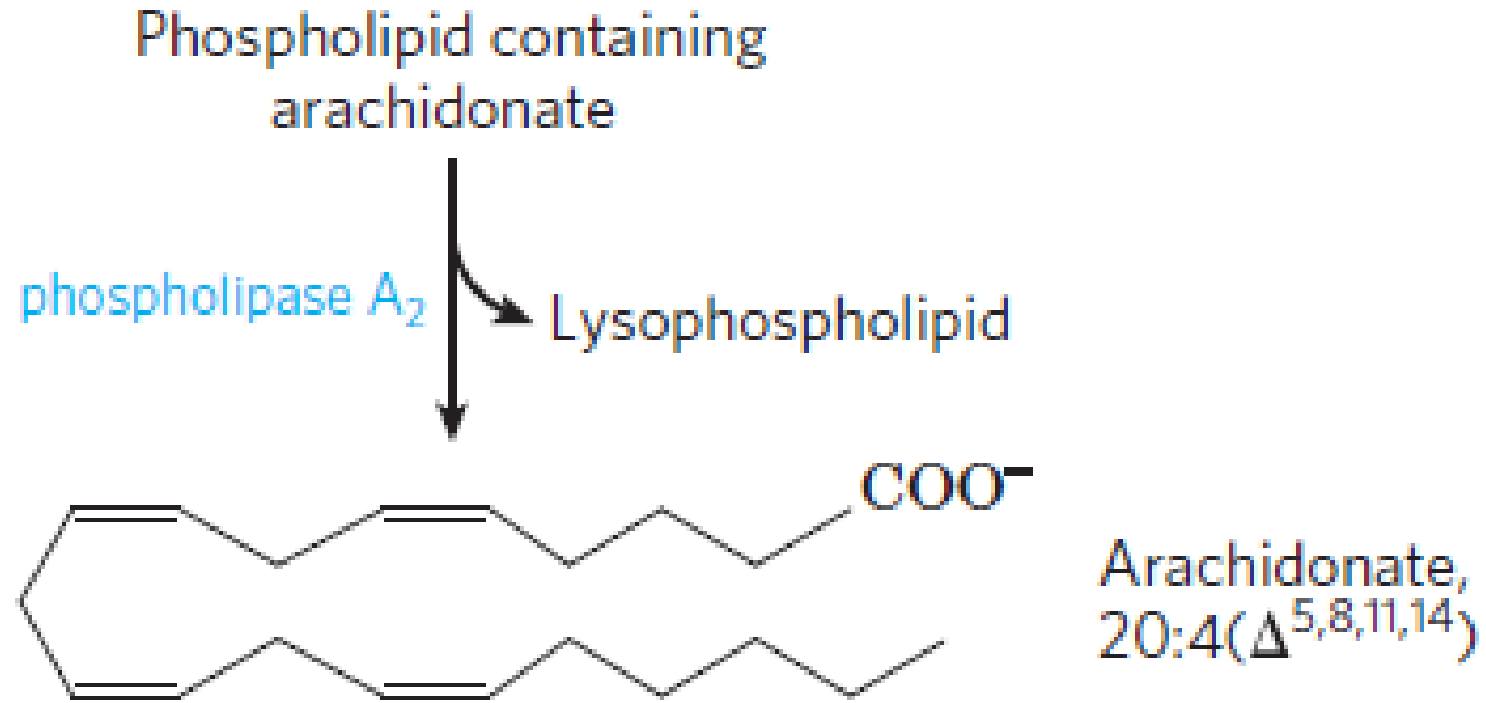
The bonds between the fatty acid residues (or acyl residues) at position C-1 and C-2 of glycerol are hydrolyzed by phospholipases A1 and A2 (PLA1, PLA2), respectively.



The ester bond between the glycerol and the phosphate residue in position C-3 is hydrolyzed by phospholipase C (PLC), while the ester bond between the phosphate group and the alcoholic residue of the polar head is hydrolyzed by phospholipase D.

C and D are considered phosphodiesterases.

Arachidonate is released from phospholipids by the action of phospholipase A₂,



Arachidonic acid is released from the phospholipid membranes by phospholipase



Arachidonic acid

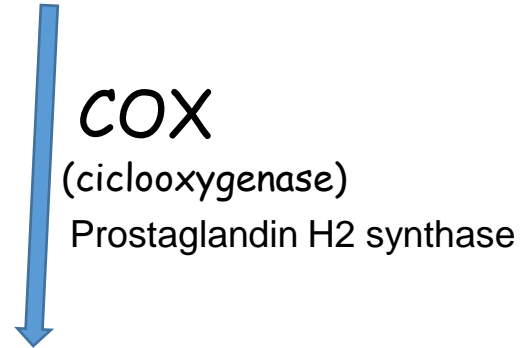
LOX (lipoxygenase)

LOX

Leukotrienes

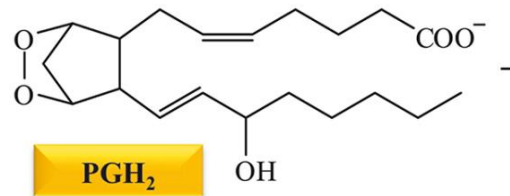
Arachidonic acid becomes the precursor of some metabolites, by 2 ways:

1) **CYCLOXYGENASE pathway**
(mainly activated in endothelial cells and platelets)

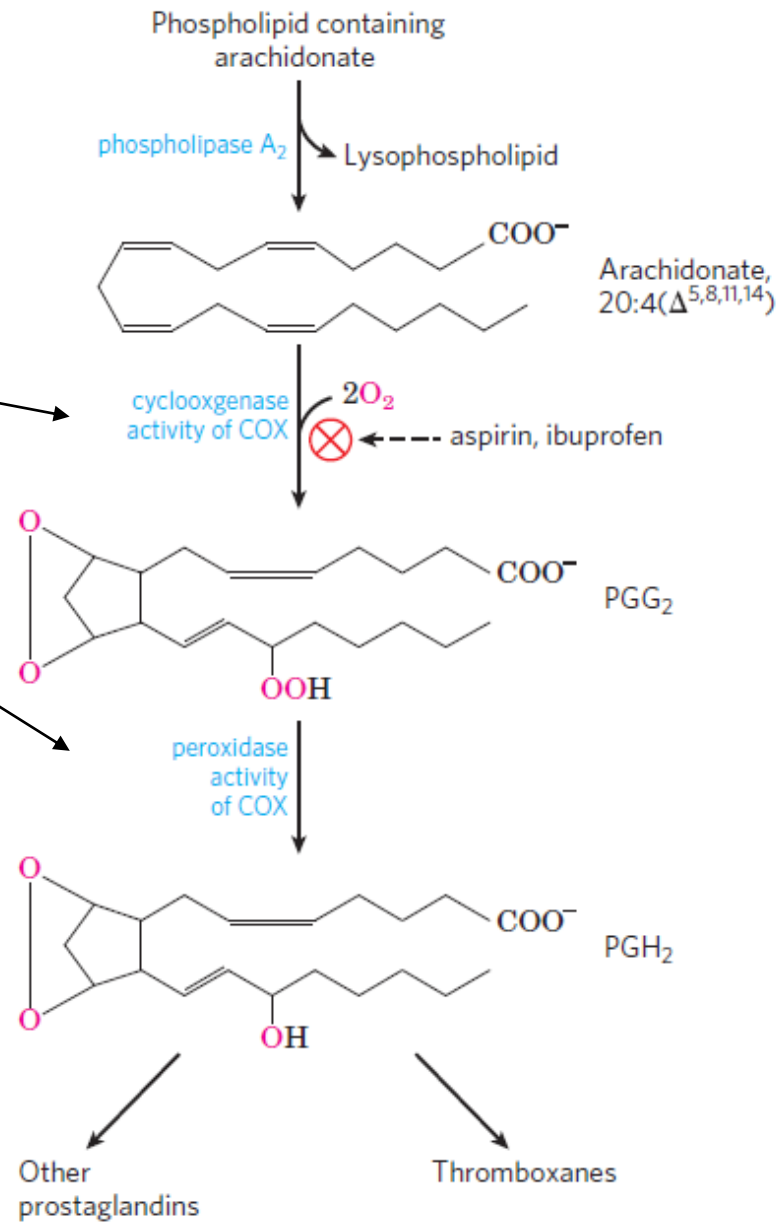


2) **LIPOXYGENASE pathway**
(mainly activated in leukocytes)

intermediate product
(prostaglandin H2)



CYCLOXYGENASE
(prostaglandin H2 synthase)



(a)

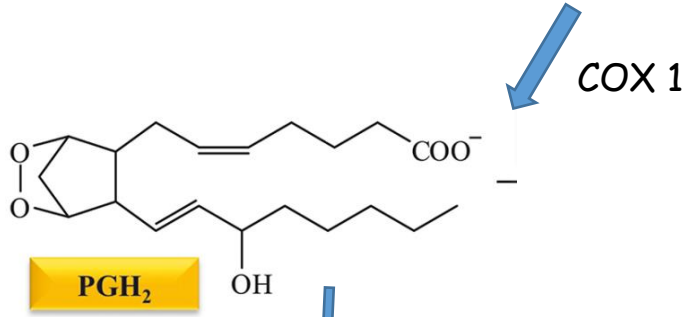
COX

Two isoforms

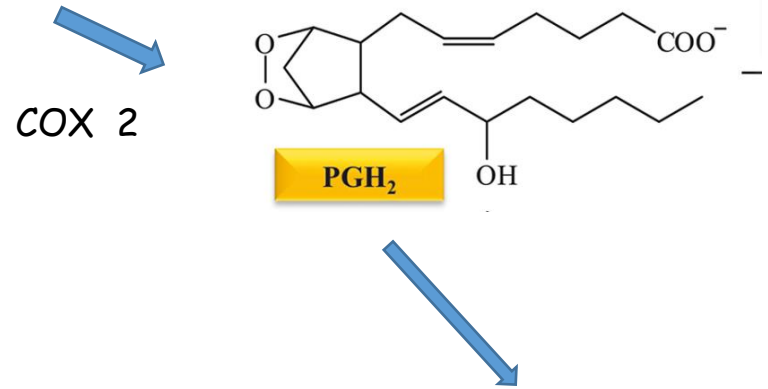
COX 1
constitutive
(ubiquitous)

COX 2
inducible
(induced expression in
inflammatory cells)

Arachidonic acid



- Prostaglandins that protect the gastric mucosa



- Prostaglandins that mediate inflammation

)

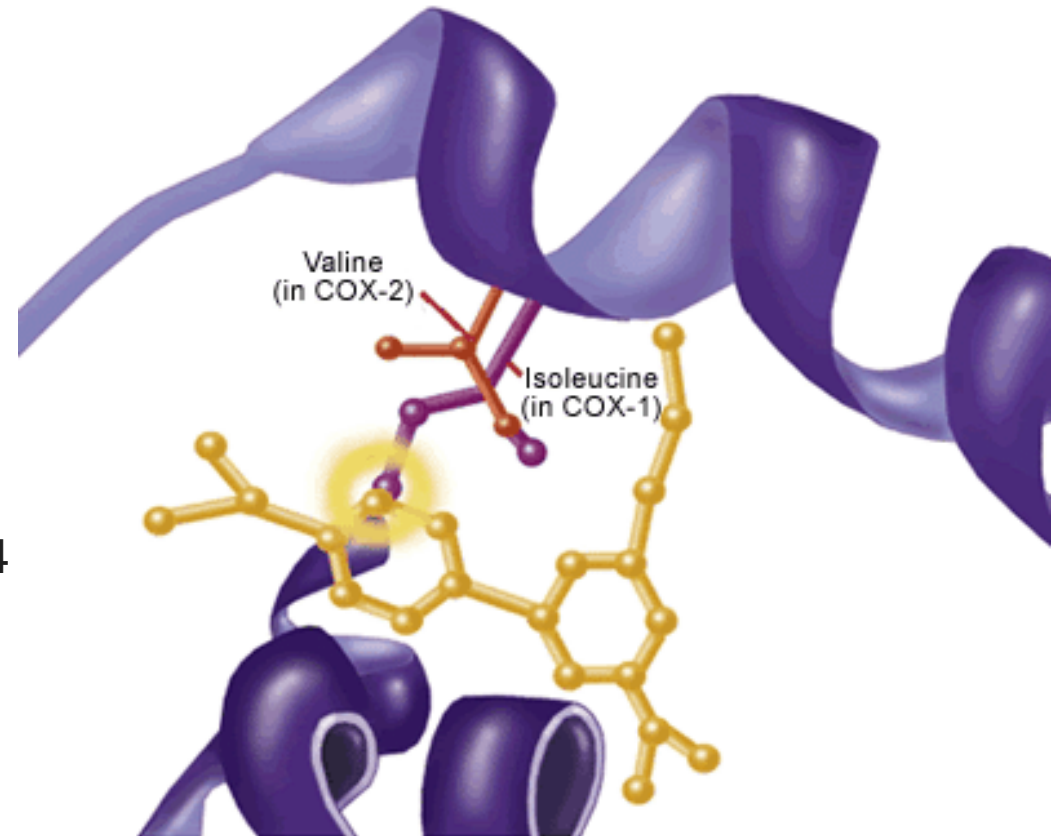
COX-1 and COX-2

They differ from amino acid substitution
Ile to Val on 523 position

They are homodimers anchored to the inner membrane of the cells

COX contains a prosthetic group, a heme group with ferrous atoms which interacting with oxygen from Fe^{2+} + becomes Fe^{3+}

Arachidonic acid (ARA) is bound in this pocket formed by 4 α helices with hydrophobic amino acids W F L V.



Arachidonic acid interacts with the catalytic site of COX (Tyr 385)

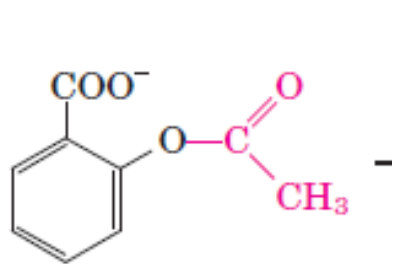
NSAIDs (non-steroidal anti-inflammatory drugs-)

Block the activity of cyclooxygenase

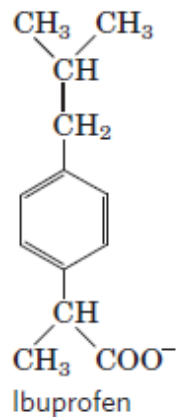
Inhibit the production of prostaglandins, thromboxanes and prostacyclins

Acetylsalicylic acid
Ibuprofen (Brufen)
Nimesulide (Aulin) } COX inhibitor

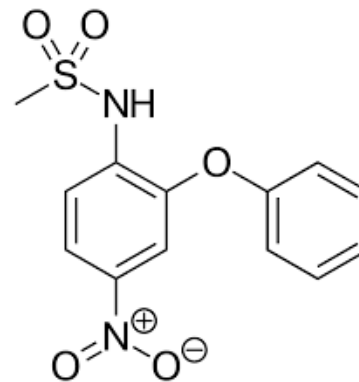
- Acetylsalicylic acid (Acetylation COX) (selective for COX 1)



Aspirin
(acetylsalicylate)

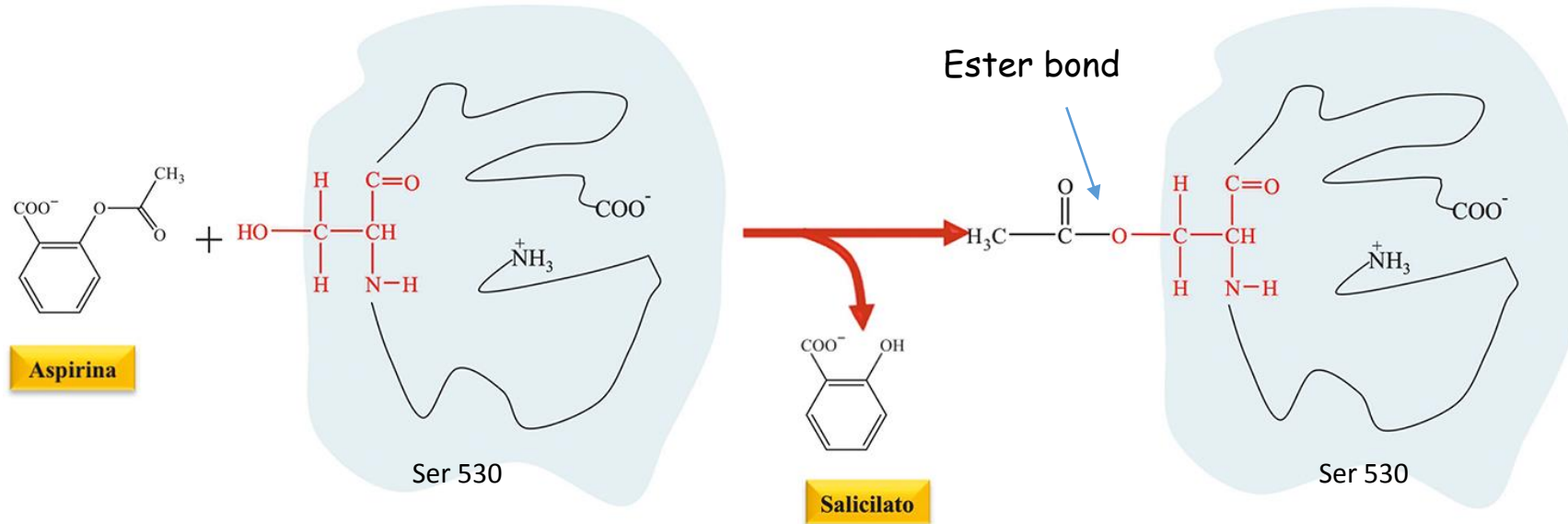


Ibuprofen



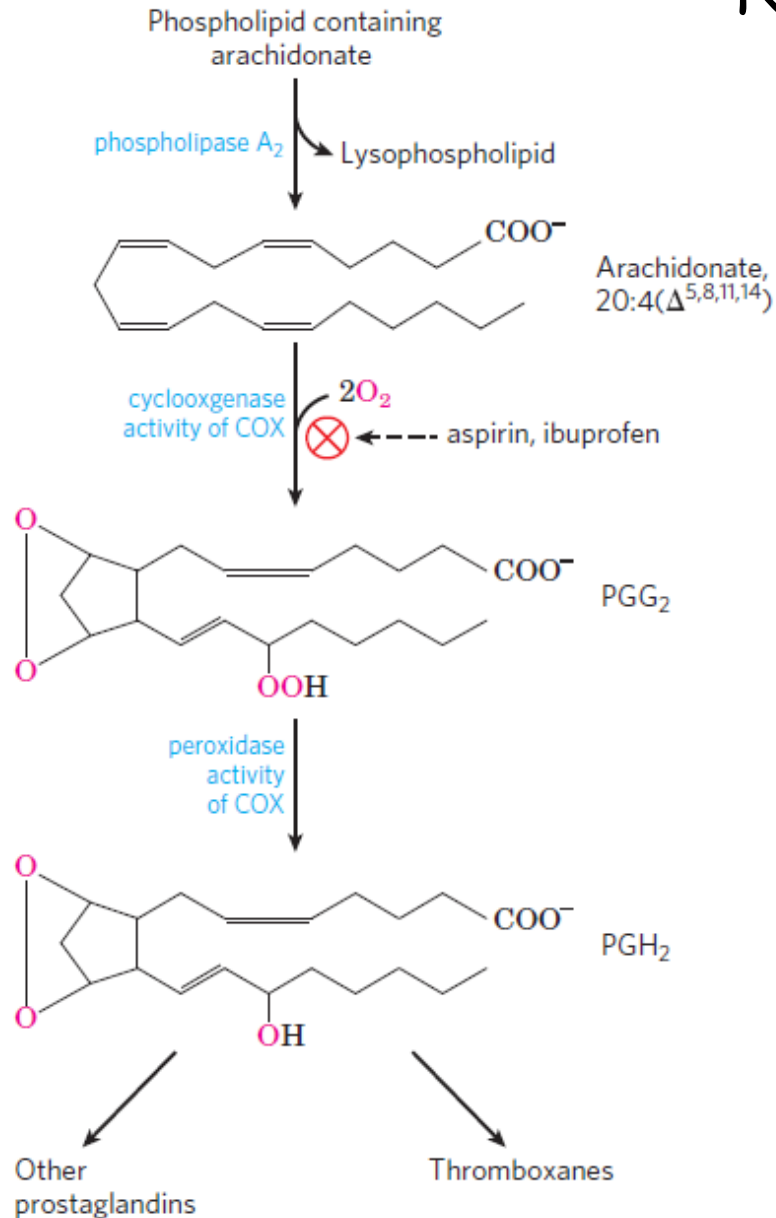
Nimesulide

- Acetylsalicylic acid (Acetylation COX) (selective for COX 1)



Aspirin inhibits COX 1 by acetylating an essential Ser residue on the enzyme.

NSAIDs (non-steroidal anti-inflammatory drugs-)



(a)

Aspirin inhibits COX 1 by acetylating an essential Ser residue on the enzyme.

Ibuprofen inhibits the same step, probably by mimicking the structure of the substrate or an intermediate in the reaction.

Steroidal anti-inflammatory: (cortisone)

Prednisone (Deltacortene), metil prednisolone, desametasone, betametasone (Bentelan)

