

Acetylsalicylic acid and salicylates

Aspirin



Willow

Papyrus of Ebers 1500 a.c. bark and willow leaves



Hippocrates: V a.c. analgesic
action of the sap extracted
from the bark of the willow
tree

Reverend Edward Stone, botanist expert, who tasted by chance a piece of bark of a willow, the bitter taste made him remember that of the cinchona from which quinine was extracted. Report (1763) to the Royal Society of London

Towards Aspirin

- 1826-29: various researchers isolate a glycoside active ingredient of the willow bark (salicin)
- In 1835, in Switzerland, from a wild bush - spirea ulmaria - a similar substance was obtained: it was called "spirsauo". A few years later it was realized that it was pure salicylic acid
- A Calabrian chemist, Raffaele Piria discovered that salicylic acid could be derived from salicin

Raffaele Piria
(1814-1865)
chemist

- Ardent patriot, involved in First Independence War in 1848
- Ordinary Member of Superior Council del Consiglio Public Education in 1859
- Senator in 1862

Together with his student Stanislao Cannizzaro promoted a real "national program" of research in chemistry, both in institutional terms and concerning research themes



Piria

Pirà gave seminal contributions to the history of aspirin



willow



willow bark

In 1838, Pirà, in Paris, isolated and purified salicylic acid from compounds extracted from willow bark (“salicilin”)

He returned to Italy in 1839

Towards Aspirin

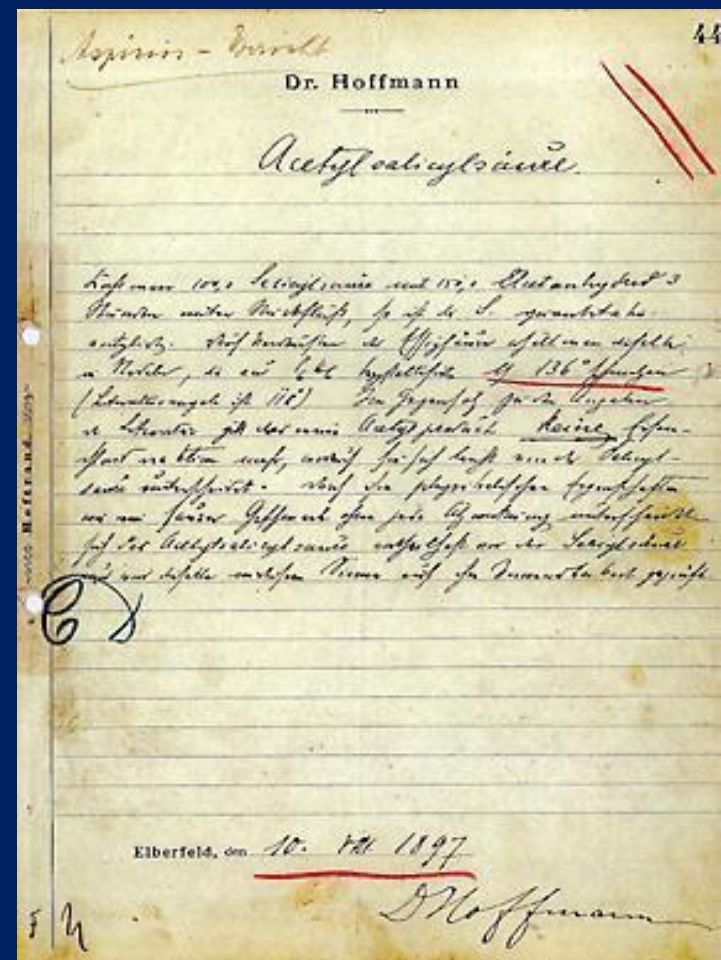
In 1853: the French chemist Charles Frédéric Gerhardt first obtained **salicylic acid**, but did not market it

Later the German Hermann Kolbe discovered the structure of **salicylic acid** and synthesized it

In 1874 the industrial production of **salicylic acid** could be started. Because of its acidity, it caused severe irritation of the mucous membranes of the mouth, throat and stomach



The laboratory note in which Hoffmann describes acetylsalicylic acid (ASA), which he synthesized in a chemically pure and stable form

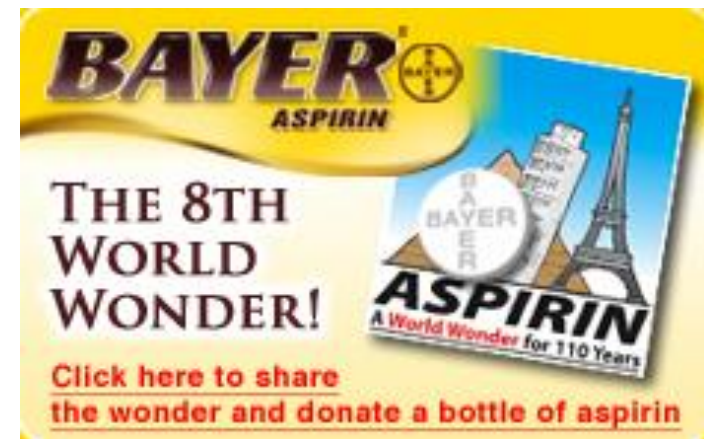


The turning point came in 1893, when Felix Hoffmann, a chemist from the German company Bayer, rediscovered Gerhardt's formula, capable of treating his father's rheumatism, developing an advantageous synthesis of acetylsalicylic acid, with the same therapeutic properties as salicylic acid, without however irritate the oral and gastric mucosa



1899 Aspirin (acetyl-spirea)

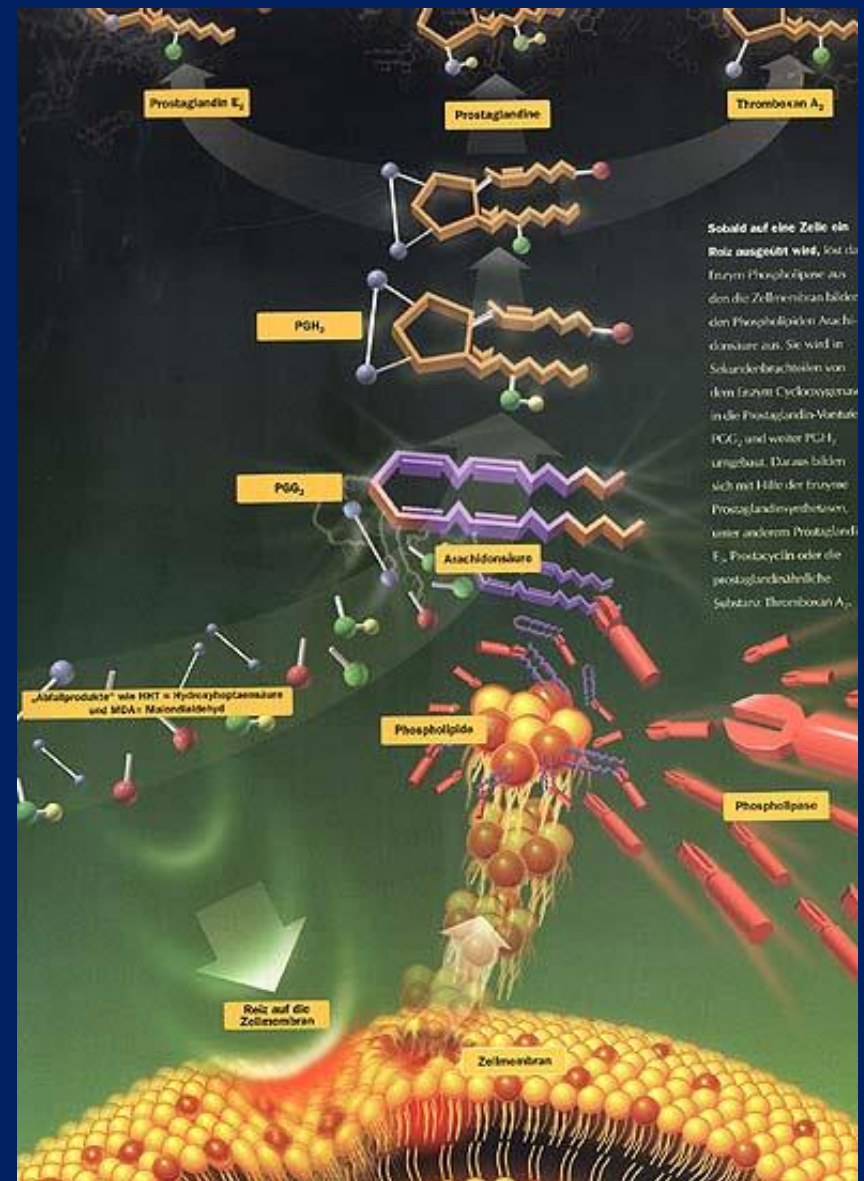
From the early 1950s Aspirin is the most well-known drug in the world for the most common disorders, from cold to migraine



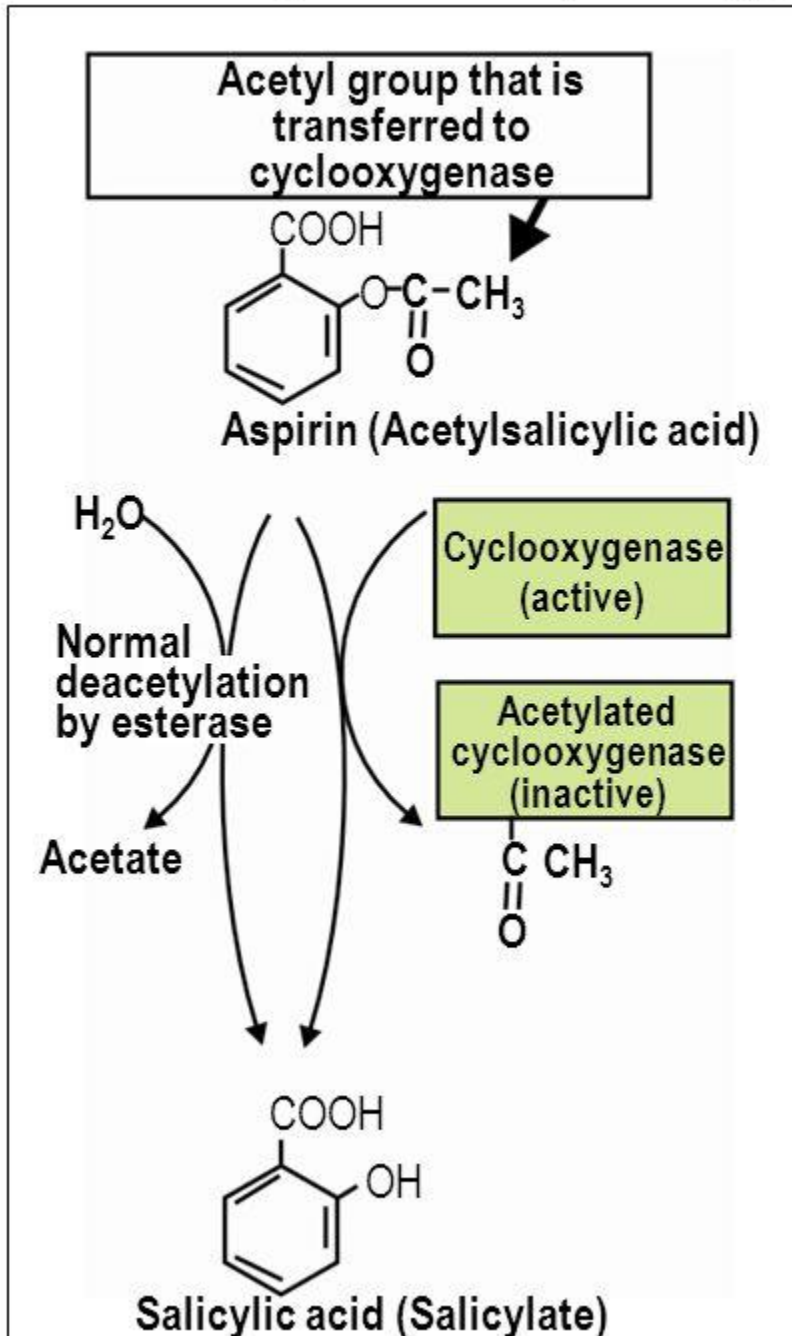
Discovery of the mechanism of action of NSAID

June 23rd, 1971

John Vane, professor of pharmacology at the Royal College of Surgery London published his studies on the mechanism of action of ASA, under the title "**Inhibition of Prostaglandin Synthesis as a Mechanism of Action of Aspirin-like Drugs**" in the journal Nature. He discovered that ASA had analgesic, antipyretic effects and anti-inflammatory properties because it inhibits the synthesis of some messenger substances (prostaglandins) in the body

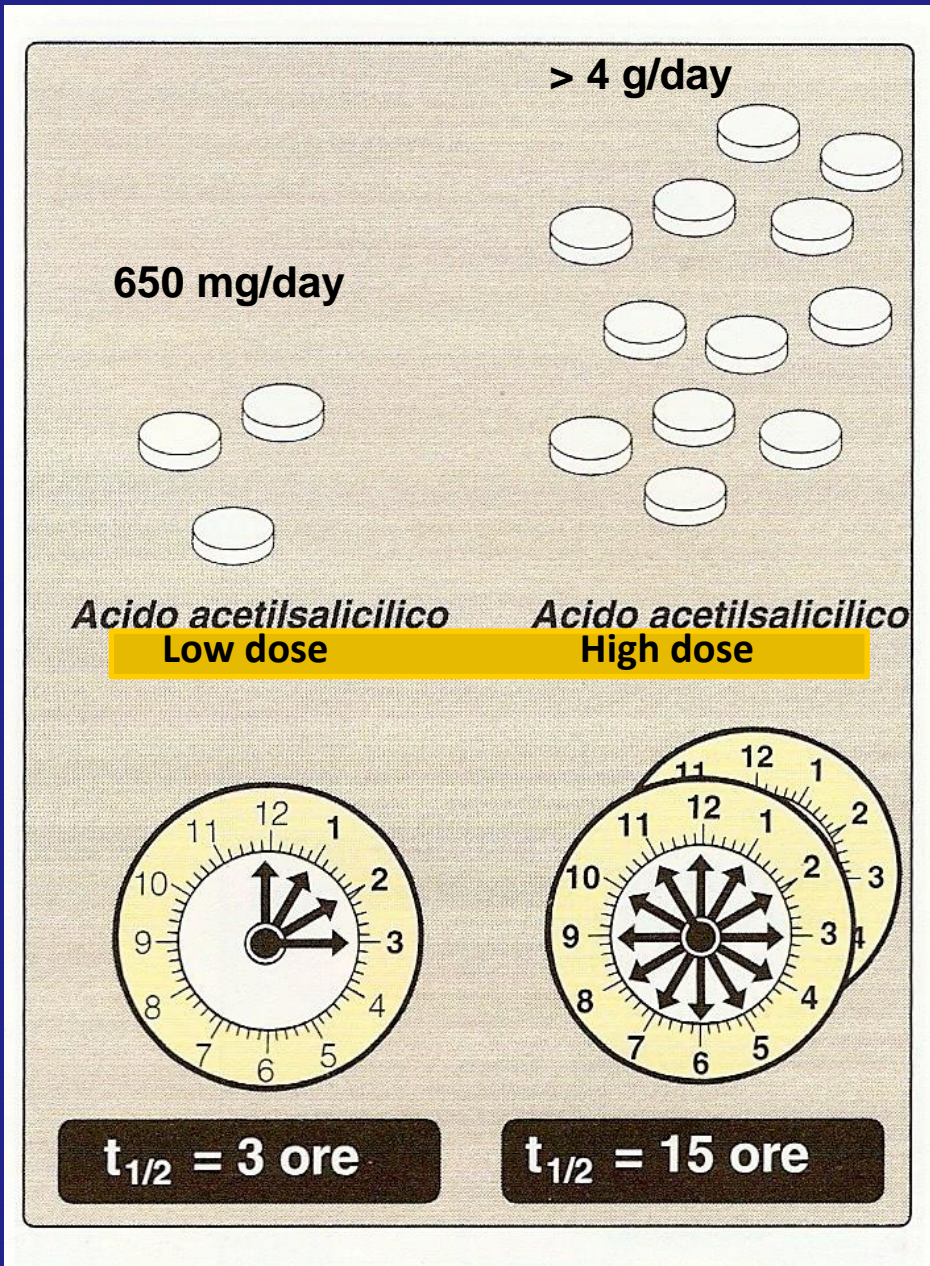


Metabolism of aspirin and acetylation of cyclooxygenase by aspirin



Irreversible
Acetylation

EFFECT OF THE DOSE ON THE HALF-LIFE OF ASPIRIN

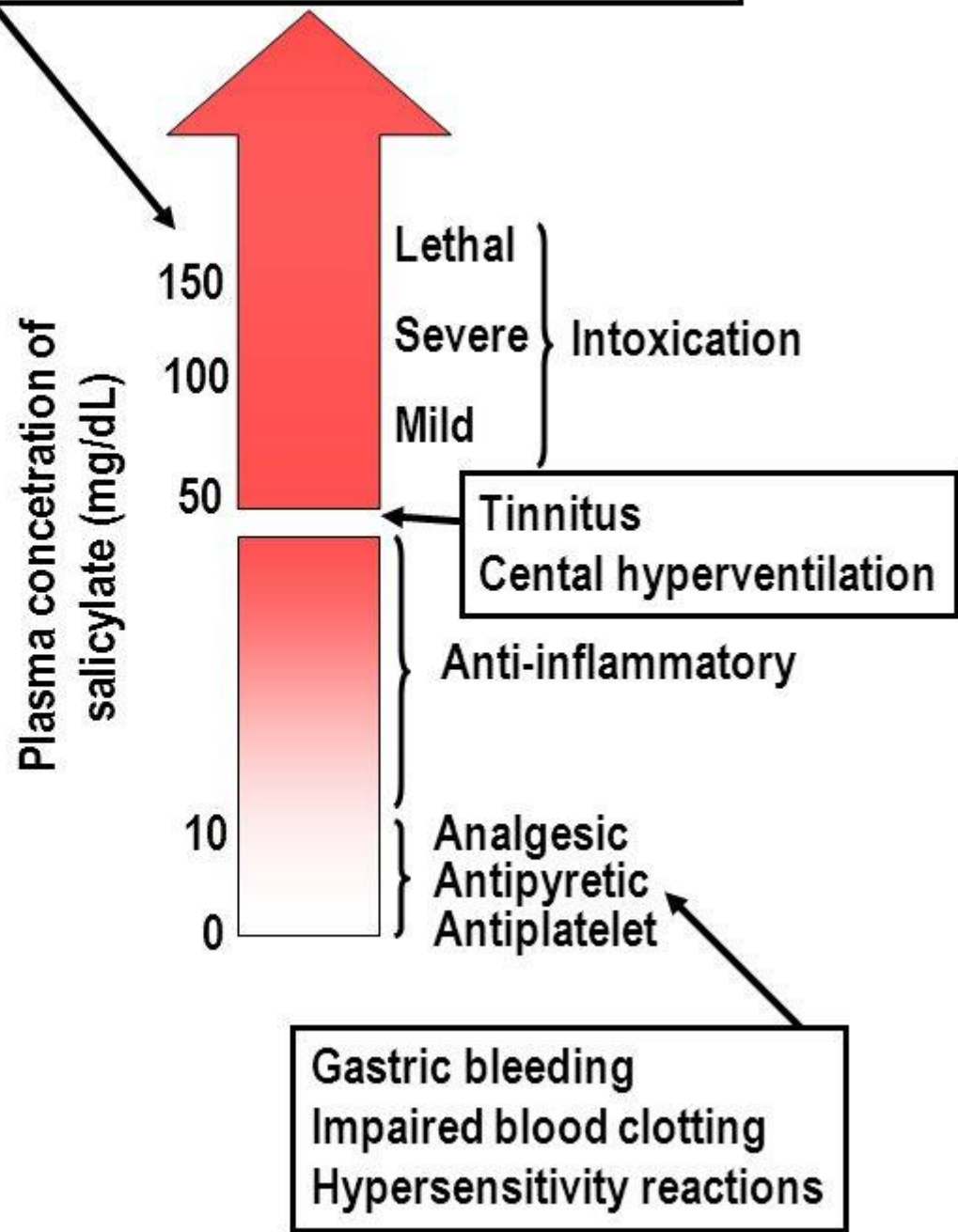


Dissolution (low in stomach)
Absorption (facilitated in stomach)
Dosage
In the liver water-soluble conjugates are formed, excreted by the kidney

Kinetics: 1 order 0 order (saturation)

Dose-dependent effects of salicylate

Vasomotor collapse; Coma; Dehydration



(according to Lippincott's Pharmacology 2006)

Aspirin

Low Doses



1) Uncoupling oxidative phosphorylation
+ O₂ + CO₂ + breathing
(respiratory center stimulation)

High Doses



2) Hyperventilation and alkalosis
(+ bicarbonate excretion)

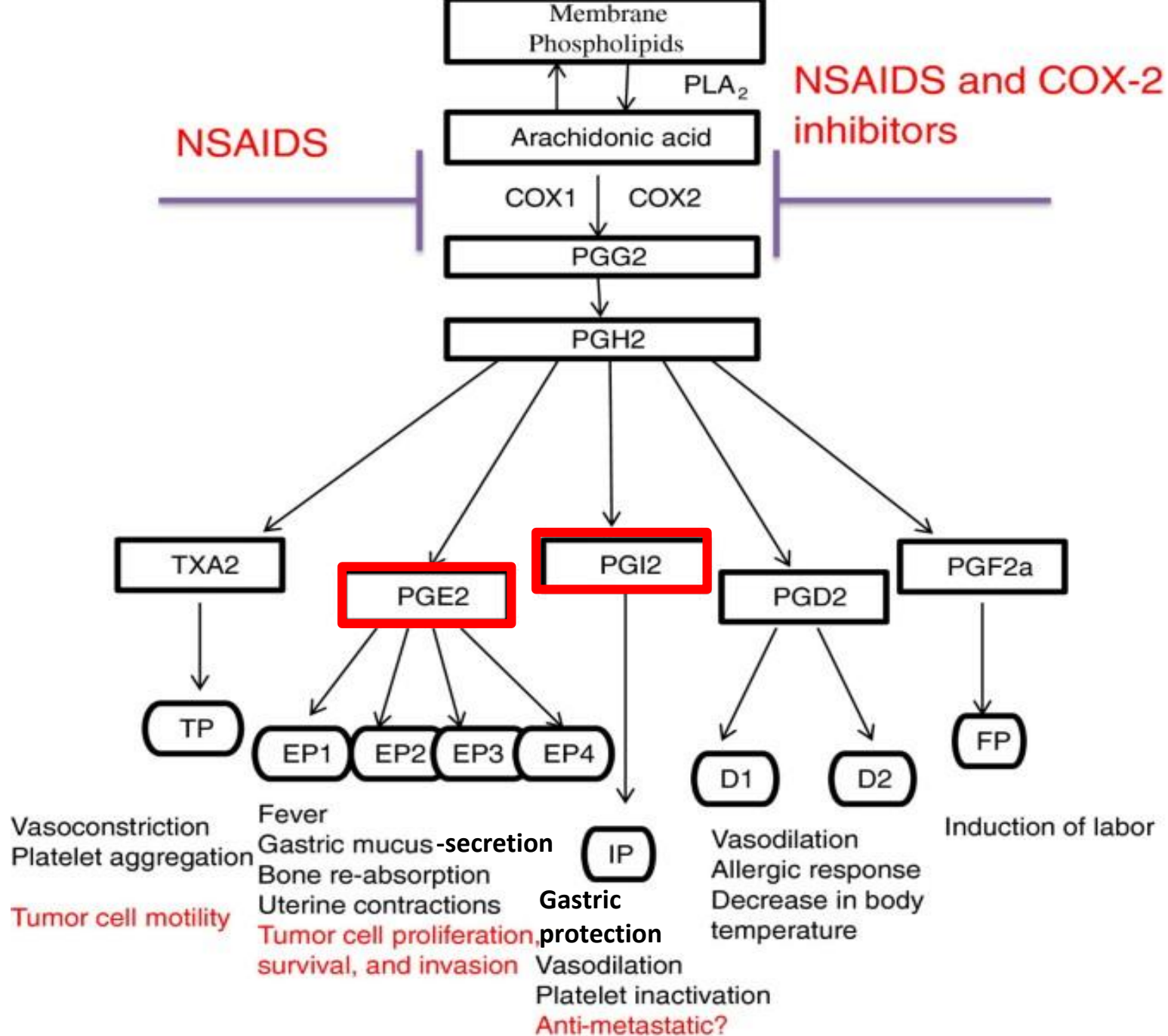
Toxic Doses



3) Respiratory paralysis + CO₂
+ excretion bicarbonate (acidosis resp.)

Reye's Syndrome (pediatric age)

Gastrointestinal effects



Risk factors for NSAID gastropathy

- General

Age

Gender (female)

Stress, anxiety

Abuse of substances: smoke, alcohol....

Other drugs interaction

- Specific

Presence of past gastritis or other gastric pathologies

Intolerance to NSAID

Tab. 4.3. Meccanismi di danno dell'epitelio gastrointestinale indotto dai FANS.

– **Azione sistemica dei FANS**

Riduzione della secrezione e della viscosità del muco

Riduzione della secrezione di bicarbonati

Inibizione dei processi di rigenerazione tissutale

Alterazioni del flusso ematico locale

Aumentata sintesi locale di sostanze ad azione lesiva

Aumentata infiltrazione mucosa da parte dei polimorfonucleati

Interazioni negative con i sali biliari

Ritardo di cicatrizzazione di lesioni preesistenti

– **Azione lesiva topica dei FANS**

Assottigliamento e riduzione della viscosità del muco

Ridotta produzione di bicarbonati

Riduzione del flusso ematico locale

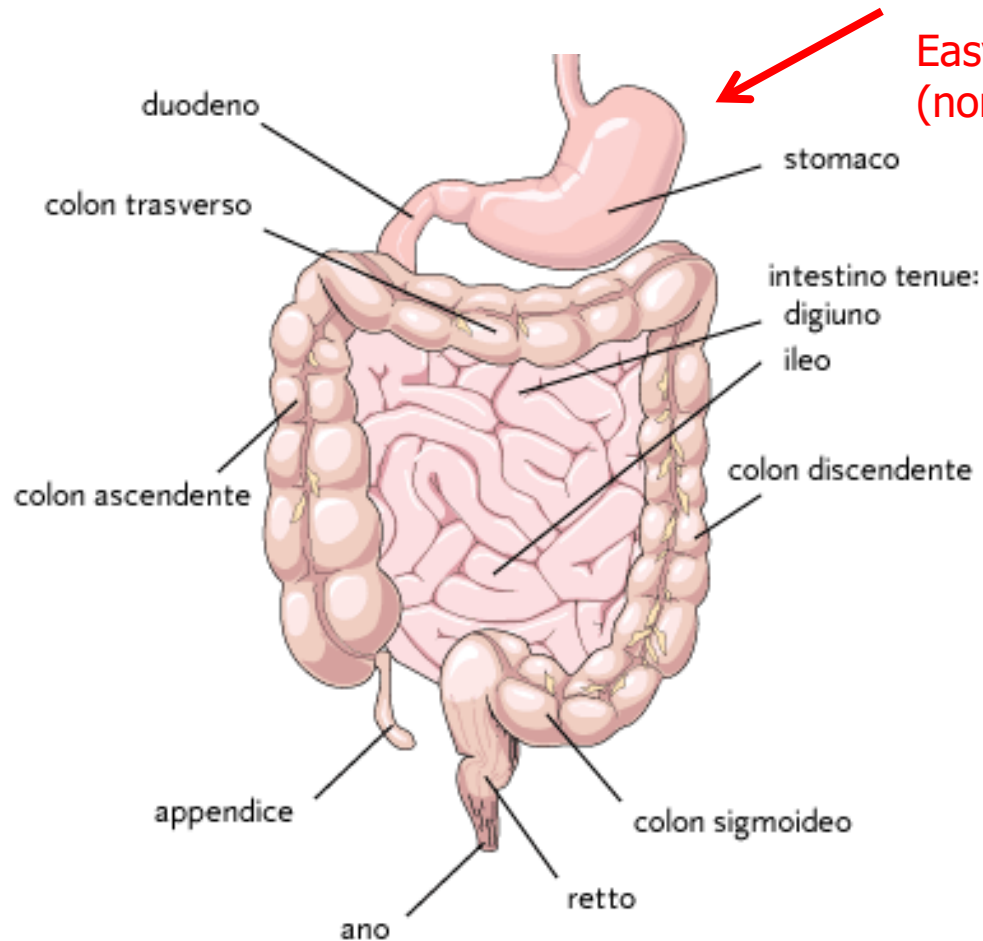
Ridotta secrezione di fattori di crescita

Penetrazione intracellulare con effetto tossico diretto (ASA)

Aspirin

Low dissolution
(Gastrointestinal side)

Easy absorption
(non-ionized, basic plasma)



NSAID gastropathy and peptic disease

	Gastropatia da FANS	Malattia peptica
Eziologia	Uso di FANS	Multifattoriale
Patogenesi	Inibizione sintesi PG Caduta difese di barriera della mucosa Modificazioni dei fattori di protezione	Ipersecrezione acida non compensata da incremento dei fattori di protezione
Localizzazione	Stomaco (70% casi)	Duodeno
Età insorgenza	Più frequente negli anziani	Più frequente nei giovani
Sintomatologia	Varia, spesso assente	Dolore, piroisi, dispepsia

Prevention and therapy of gastropathy

- **General Prevention**

Correction of risk factors

Correct diagnosis and real need to use NSAIDs

Choice of the less harmful drug

Avoid NSAIDs association

Choose the most suitable route of administration

Utilization of gastro-protective formulation

Posology

Regular clinical check (endoscopy)

Cito-protectors

- **Therapy of gastropathy**

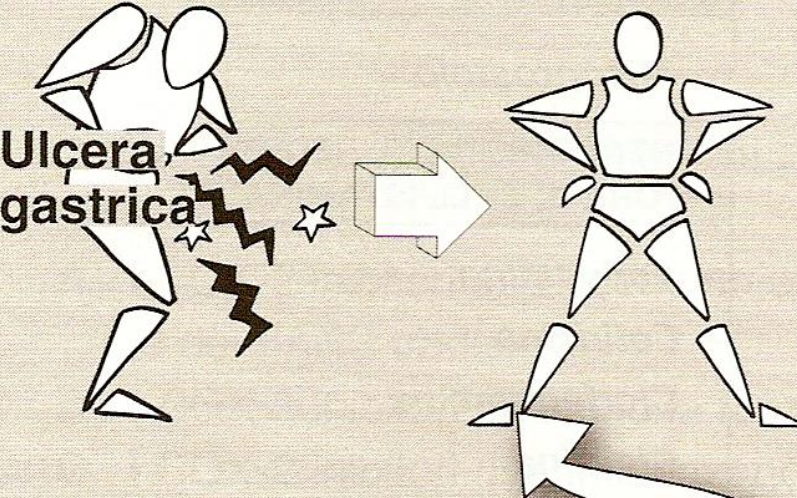
H2 inhibitors, protonic pump inhibitors

Anti-acid compounds

Mucosa protection (sucralfate)

Administration of prostaglandine analogs

Misoprostol (PGE1)



The diagram shows two stylized human figures. The figure on the left is depicted in pain, clutching its stomach, with jagged lightning bolts and stars around it, and the text 'Ulçera gastrica' (gastric ulcer) next to it. A large white arrow points from this figure to a second figure on the right, which stands upright and confident with hands on hips, representing relief from the condition.

Misoprostolo

- Inibisce la secrezione di HCl e pepsina e rafforza la resistenza della mucosa gastrica.
- Utile nei pazienti con ulcera gastrica che assumono cronicamente *acido acetilsalicilico*.

Omeprazole
(proton pump inhibitor)
Cimetidine, ranitidine
(H2 antagonists)

Antiplatelet drugs

MAIN AGGREGATING AGENTS

THROMBIN

ARACHIDONIC ACID

ADP

COLLAGEN

RISTOCETIN

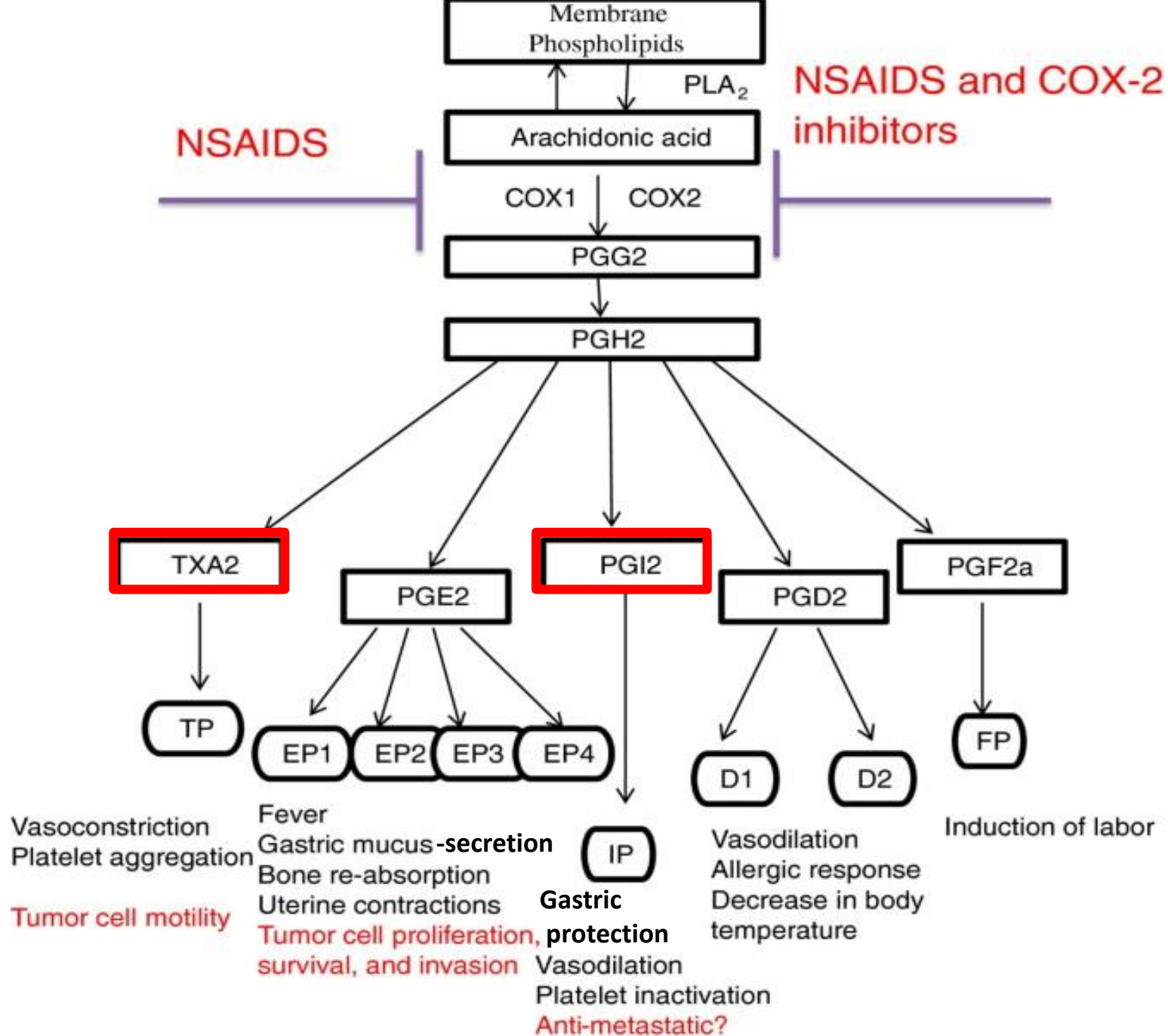
ADRENALINE

PAF

ENDOPEROXIDE ANALOGUES

IONOPHORS OF Ca^{++}

SEROTONIN



1 - Farmaci che interferiscono con il metabolismo dell'acido arachidonico
Acido acetilsalicilico

Indobufene

Sulfinpirazone

Inibitori della trombassano-sintetasi

2 - Farmaci che interferiscono con i livelli intrapiastrinici di AMP ciclico

Dipiridamolo

Prostaciclicina ed analoghi

3 - Farmaci che interferiscono con recettori specifici

Antagonisti recettoriali del PAF

Antagonisti recettoriali del trombassano

4 - Farmaci che inibiscono la trombina

Eparina

Eparine a basso peso molecolare

Peptidi sintetici

5 - Alcuni farmaci (non utilizzati come antiaggreganti) che possono alterare la
funzione piastrinica

Alpha-antagonisti

Beta-bloccanti

Anestetici locali (procaina)

Antiistaminici

Calcio antagonisti (nifedipina, verapamile)

Antidepressivi triciclici (amitriptilina, imipramina)

Fenotiazine (clorprozamina, trifluoperazina)

Papaverina

Xantinici (caffeine, teofillina)

Corticosteroidi

Antimalarici (clorochina, mepacrina)

Acido nicotinico

Beta-lattamina

Nitrofurantoina

6 - Altri

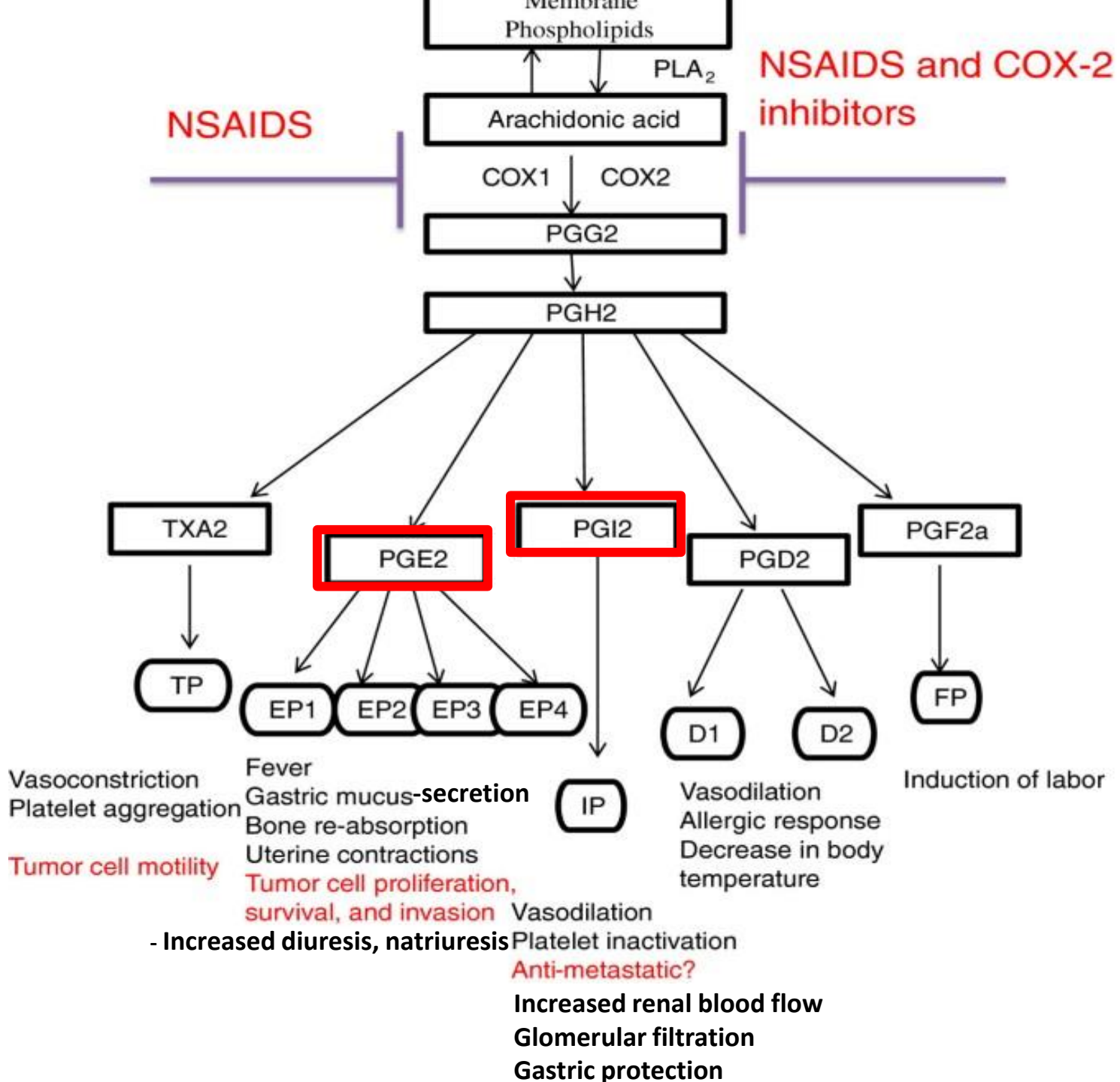
Destrano

Etanolo

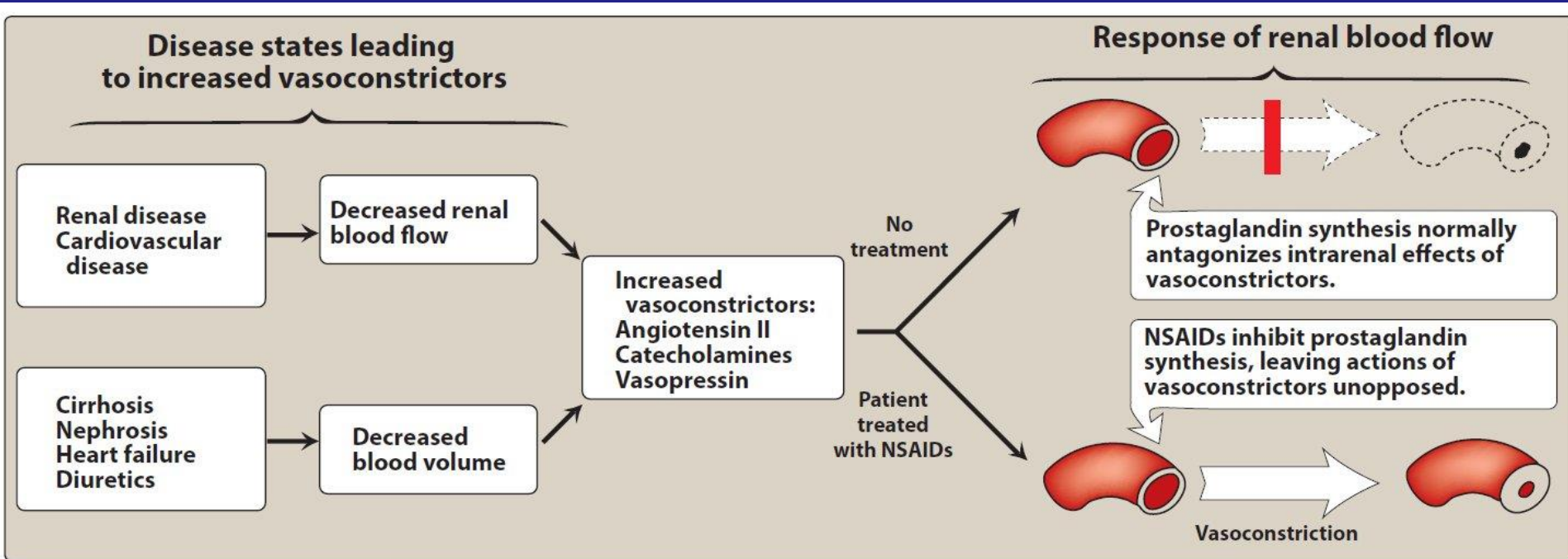
Clofibrato

Aglio

Actions on Kidney



EFFECT ON THE KIDNEY OF THE INHIBITION OF THE SYNTHESIS OF PROSTAGLANDINS BY ACETYLSALICYLIC ACID

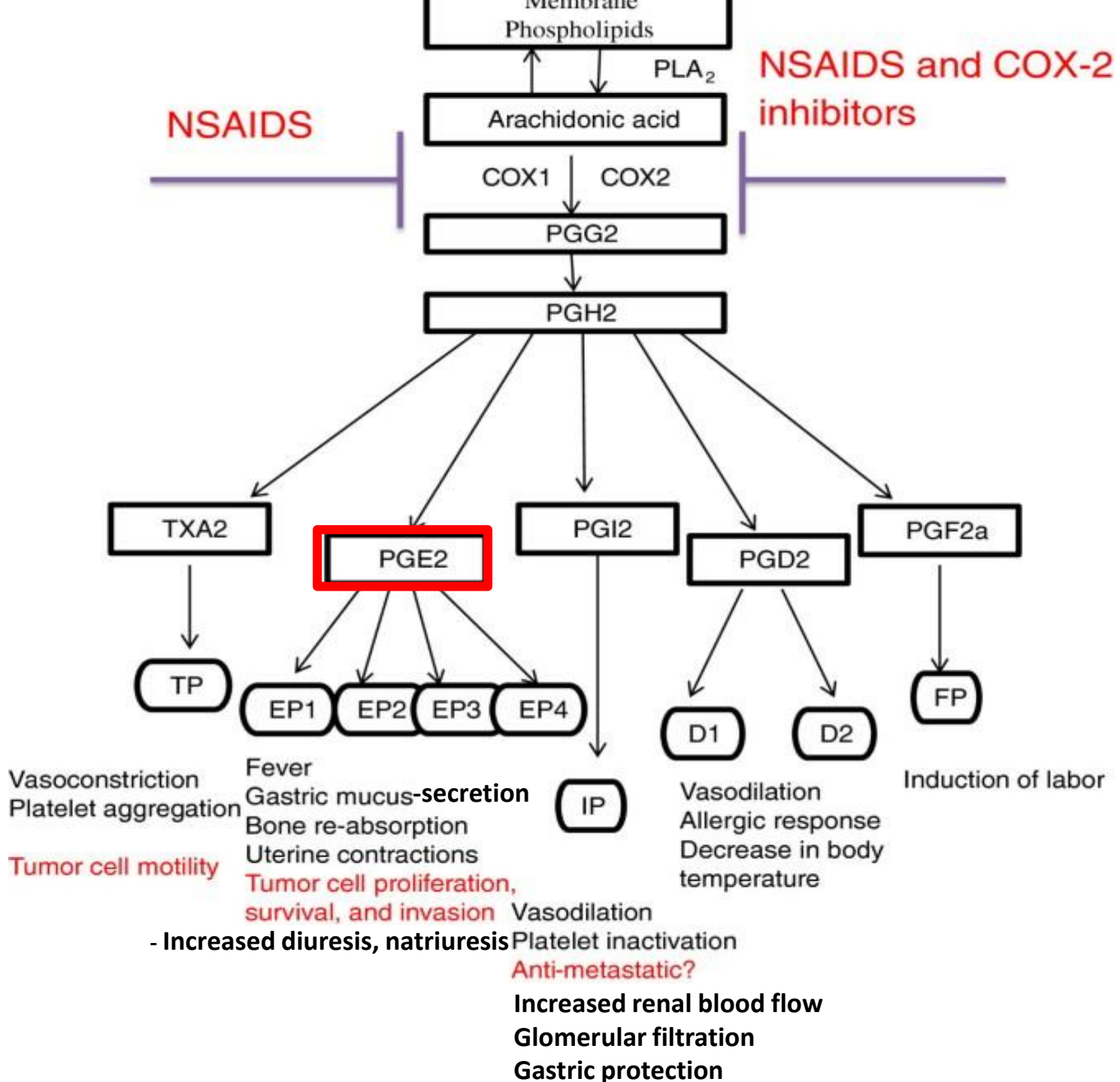


Renal effect of NSAIDs inhibition of prostaglandin synthesis. NSAIDs = nonsteroidal anti-inflammatory drugs.

Source : Lippincott Illustrated Reviews, Pharmacology - Whalen, Karen

NSAIDs: Na⁺ and H₂O retention, edema, hyperkalemia

Antipyretic effect



BACTERIA
VIRUS
TUMORS
TISSUE
DAMAGE

ENDOTOXINS
STERIODS
PYROGEN
ANTIGENS ETC.

GRANULOCYTES
MONOCYTES
MACROPHAGES
LYMPHOCYTES

IL1 and 6, TNF, IF γ
(ENDOGENOUS
PYROGENS)

HYPOTHALAMUS
↑ cAMP
↑ PGE

Fever

Heat Loss

Heat is lost from the body by the following channels:

- (1) Heat loss from the skin,
 - (a) by conduction and radiation,
 - (b) by evaporation of sweat.
- (2) Heat loss from the respiratory passages,
 - (a) from warming of the inspired air,
 - (b) by evaporation of water.

Small quantities of heat are also lost in the excretion of urine and fæces.

The centres which regulate temperature can cause an increase in the heat loss by the following means:

- (1) By causing vasodilatation of the skin, and thus increasing the heat lost by conduction and radiation.
- (2) By increasing the volume of air respired and thus increasing the heat lost by evaporation and by warming the inspired air.
- (3) By causing sweating and thus increasing the heat lost by evaporation.

HEAT EXCHANGE
IN CALORIES PER
SQUARE METRE BODY
SURFACE PER HOUR.

HEAT
PRODUCED



HEAT
LOST



50

40

30

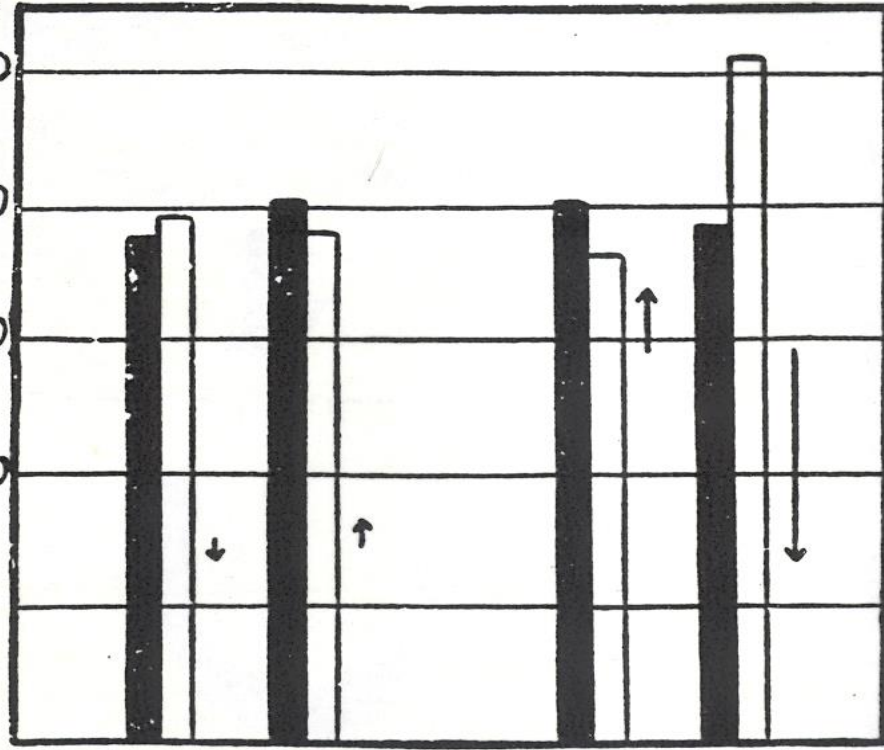
20

BODY
TEMP.
40°C

39°C

38°C

37°C

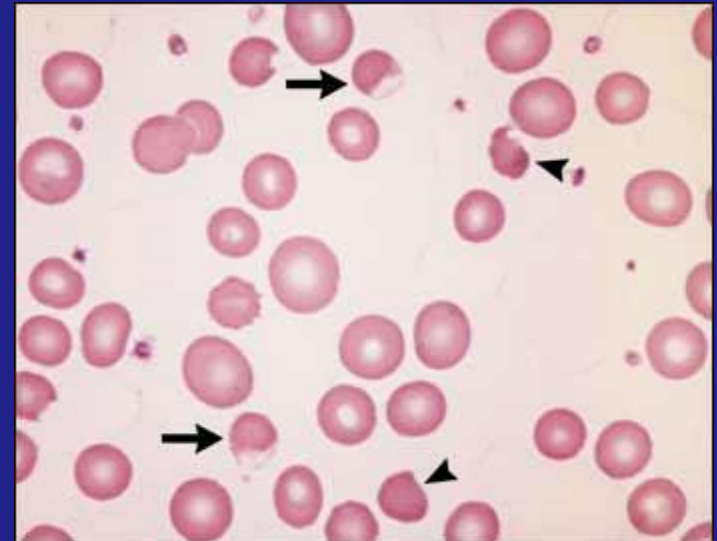
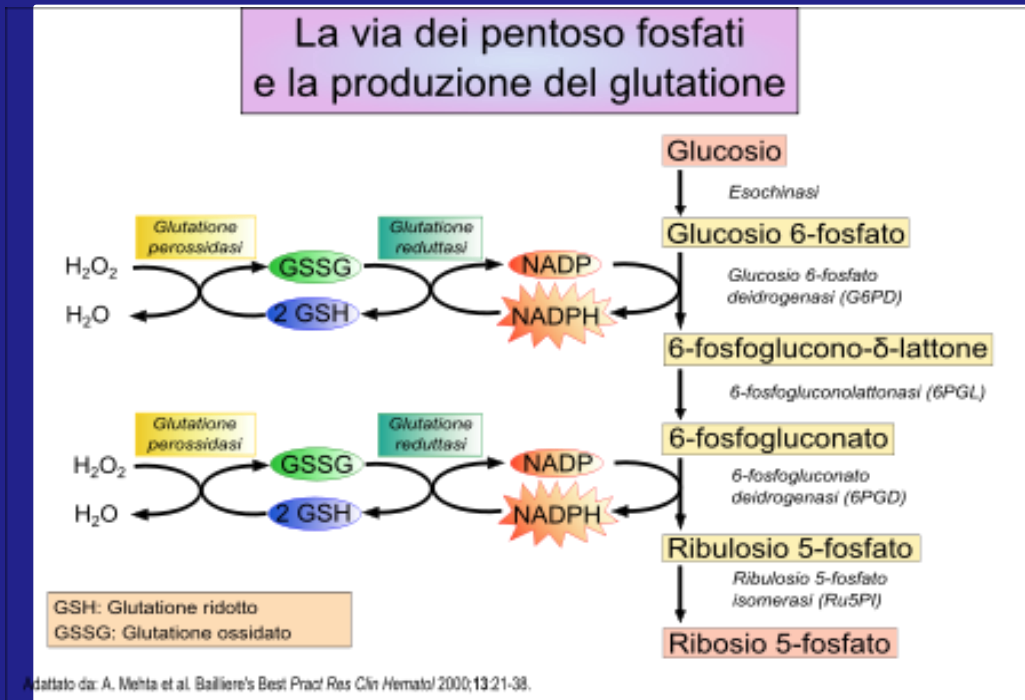


BEFORE TAKING ASPIRIN. AFTER TAKING ASPIRIN. BEFORE TAKING ASPIRIN. AFTER TAKING ASPIRIN.
NORMAL SUBJECT FEVERED SUBJECT

FIG. 19.2. Diagram showing the action of acetylsalicylic acid (1 G.) on the heat exchange in a normal and in a fevered subject. The heat exchange and the change in temperature were measured over two-hour periods before and after administration of the drug. The drug produced little effect on the normal person, but in the fevered subject it caused a great increase in heat loss and a consequent fall in temperature. The arrows show the direction and extent of the change of body temperature. (From figures by Barbour, 1919.)

G6PD deficiency

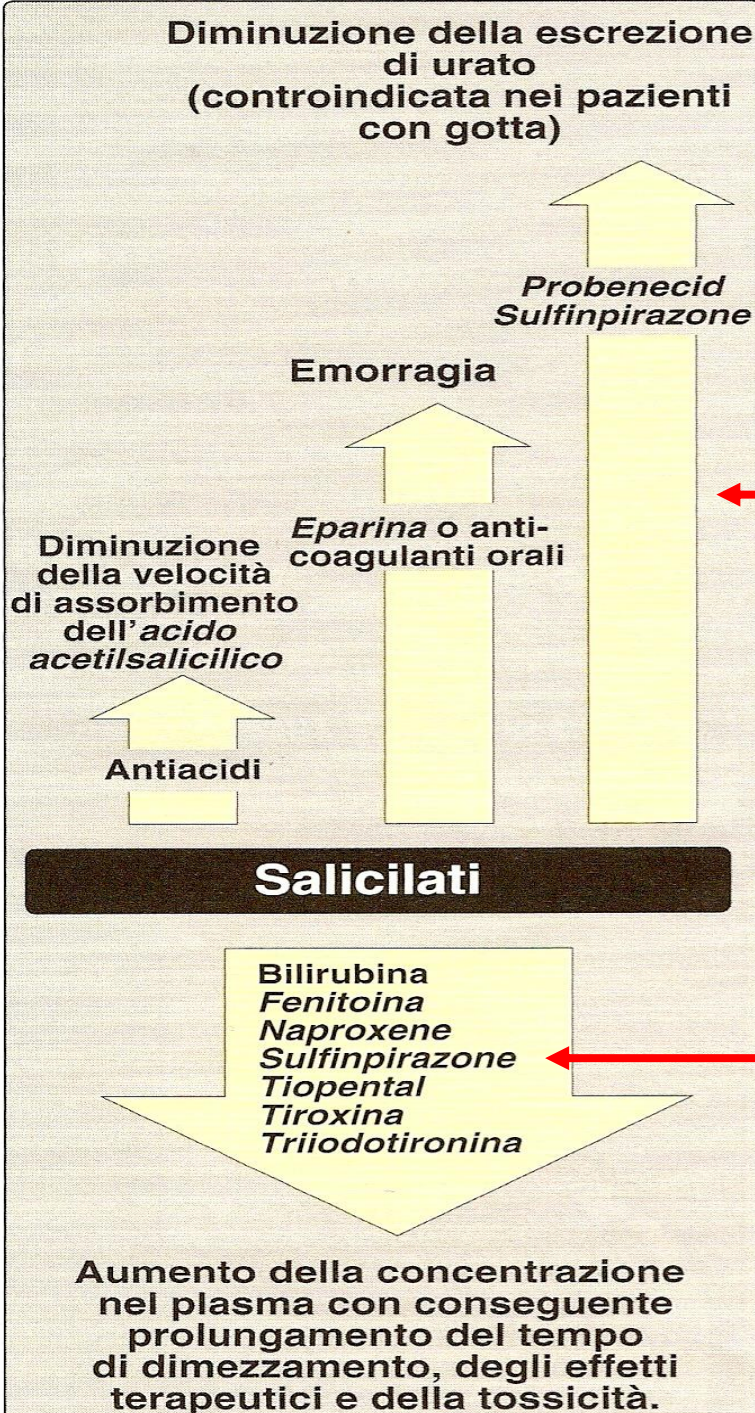
Red blood cells in hemolysis



-SH

Oxidants are produced continuously inside red blood cells, but are also generated by activated phagocytes (for example during infections). In the absence of G6PD, the ability of reduced glutathione to "detoxify" cells from the oxidants produced, which accumulate, damaging the vital components of the cells themselves, decreases. Hemoglobin oxidation produces inactive methaemoglobin and intracellular precipitates of oxidized hemoglobin known as Heinz bodies. Hemolysis represents the final outcome of oxidative damage. **Aspirin**, when taken at the usual therapeutic dosages by patients suffering from the common variants of G6PD deficiency is not responsible for hemolysis and, if this occurs, it is almost certainly due to an infection or fever. The drug is safe even when used chronically at low doses as an antiplatelet treatment. The same is true for **paracetamol**: in studies conducted in patients with even severe variants of G6PD deficiency, the drug, at therapeutic doses, did not cause hemolysis and the cases that continue to be reported can be attributed to the combination of fever or infections. The instruction card contained in paracetamol and acetylsalicylic acid still report as a prudent measure, maintained by the manufacturers, the warning to "administer with caution" these drugs to patients with G6PD deficiency

INTERACTION WITH OTHER DRUGS



Aspirin and probenecid compete for organic acid transporter
Excretion of salicylate may affect uric acid
Low doses of salicylate reduced uric acid excretion

Decrease of aspirin clearance

Chronic pain treatment

Inflammatory pain

Neuropathic pain

Antidepressants

Antiepileptic

NSAIDs

Opioids

Therapeutic Utilization of NSAID

FANS

Acido acetilsalicilico ← **Aspirina, aspro**

Diflunisal ●

Diclofenac ← **Voltaren**

Etodolac

Fenamati

Fenoprofene

Flurbiprofene ← **Froben**

Ibuprofene ← **Brufen, cibalgina**

Indometacina ← **Indoxen**

Ketoprofene ← **Orudis**

Meloxicam

Metilsalicilato

Nabumetone

Naproxene ← **Momendol**

Nimesulide

Oxaprozin

Piroxicam ← **Feldene**

Sulindac

Tolmetina

■ Acido propionico

■ Oxicami

■ Acido acetico

INIBITORI COX-2

Celecoxib , (*rofecoxib*)

ALTRI ANALGESICI

Paracetamolo ← **Efferalgan, tachipirina**

OTHER NSAIDs 1

PROPIONIC ACID DERIVATIVES

Ibuprofen, naproxen, fenoprofen, ketoprofen, flurbiprofen, oxaprozin
AR > AN / AP

DERIVATIVES OF ACETIC ACID

Indomethacin AR > AN / AP (toxicity), sulindac, etodolac (COX-2)

OXICAM

Piroxicam, meloxicam (long half-life) AR > AN / AP

FENAMATES

Mephenaminic acid, meclofenamate (little used)

AP = antipyretic

AR = arthritis-rheumatoid, arthrosis, anti-inflammatory

AN = analgesic

OTHER NSAIDs 2

ETHERO-aryl

Diclofenac AP / AR / AN + powerful than other NSAIDs

Ketorolac (+ analgesic, - AR)

Tolmetidine AR (ASA similar but better tolerated potency)

PARA-AMINOPHENOLIC

Paracetamol AP / AN

Nabumetone AR (prodrug, ASA similar potency, better tolerated)

Celecoxib, etoricoxib, lumiracoxib (~~refecoxib, valdecocoxib~~) AR

AP = antipyretic

AR = arthritis-rheumatoid, arthrosis, anti-inflammatory

AN = analgesic

Principio attivo	Effetto	Meccanismo	Comportamento clinico
Antiacidi	Preparati contenenti alluminio e carbone attivo: ridotto assorbimento dei FANS, ridotta efficacia. Preparati contenenti bicarbonato: incremento assorbimento dei FANS, aumento rischio di tossicità	Alterazione dell'assorbimento	Se ne sconsiglia la cosomministrazione
Anticoagulanti orali	Possibile aumento dell'effetto anticoagulante delle curarine, del fenindione e delle eparine a basso peso molecolare: rischio di episodi emorragici.	Spiazzamento legami proteici, inibizione del metabolismo e dell'aggregazione piastrinica	Se ne sconsiglia la cosomministrazione
Antidepressivi	Aumento del rischio di emorragia con inibitori della ricaptazione della serotonina o venlafaxina.	Non noto	Se ne sconsiglia la cosomministrazione
Antidiabetici	Aumento dell'effetto delle sulfaniluree, ipoglicemia	Non noto	Monitorare la glicemia
Antiipertensivi β -bloccanti ACE-inibitori Diuretici tiazidici	Aumento del rischio di insufficienza renale e antagonismo dell'effetto ipotensivo; aumento del rischio di ipercaliemia con ketorolac	Inibizione della sintesi renale di prostaglandine, riduzione dell'effetto natriuretico	Monitorare la risposta terapeutica
Antimicotici	Aumento della concentrazione plasmatica di parecoxib e celecoxib con fluconazolo	Inibizione del metabolismo mediato dal citocromo P450 2C9	Iniziare la cosomministrazione con la dose più bassa indicata per il coxib
Antivirali	Aumento della concentrazione plasmatica di piroxicam con ritonavir (rischio di tossicità); possibile aumento della concentrazione plasmatica di altri FANS con ritonavir; aumento del rischio di tossicità ematologica con zidovudina	Inibizione competitiva della glucuronazione	Monitorare i segni tossici
Ciclosporina	Aumento del rischio di nefrotossicità, colestasi, parestesie	Non noto	Monitorare i segni di tossicità da ciclosporina

Contraccettivi orali	Aumento del catabolismo	Non noto	Cautela nell'uso concomitante
Digossina	Aumento del rischio di tossicità da digossina (nausea, vomito, aritmie)	Non noto	Monitorare i segni di tossicità
Fluorochinoloni	Possibile aumento del rischio di crisi convulsive	Inibizione del GABA	Evitare cosomministrazione
Litio	Aumento del rischio di tossicità da litio (astenia, tremori, sete, confusione)	Riduzione della clearance del litio	Monitorare i segni di tossicità da litio
Metotrexato	Riduzione dell'eliminazione di metotrexato con aspirina, aumento di tossicità con altri FANS	Riduzione della clearance	Se ne sconsiglia la cosomministrazione
Probenecid	Aumento delle concentrazioni plasmatiche di indometacina, ketoprofene, naproxene e ketorolac	Riduzione della clearance del FANS	Monitorare l'insorgenza di tossicità
Tacrolimus	Scopenso renale acuto (principalmente ibuprofene)	Non noto	Evitare cosomministrazione
Vaccino antivaricella	Cosomministrato con aspirina comporta un aumento del rischio di sindrome di Reye	Non noto	Non somministrare salicilati prima di 6 settimane dopo la vaccinazione

Paracetamol



- Paracetamol has potent analgesic and antipyretic actions but rather weaker anti-inflammatory effects.
- It is given orally and metabolised in liver (half-life 2–4 hours).
- Toxic doses cause nausea and vomiting, then, after 24–48 hours, potentially fatal liver damage by saturating normal conjugating enzymes, causing the drug to be converted by mixed function oxidases to *N*-acetyl-*p*-benzoquinone imine. This, if not inactivated by conjugation with glutathione, reacts with cell proteins and kills the cell.
- Agents that increase glutathione (acetylcysteine intravenous or methionine orally) can prevent liver damage if given early.

FANS

Forte

Effetto
antiinfiam-
matorio

Effetto
analgesico

Effetto
anti-
piretico

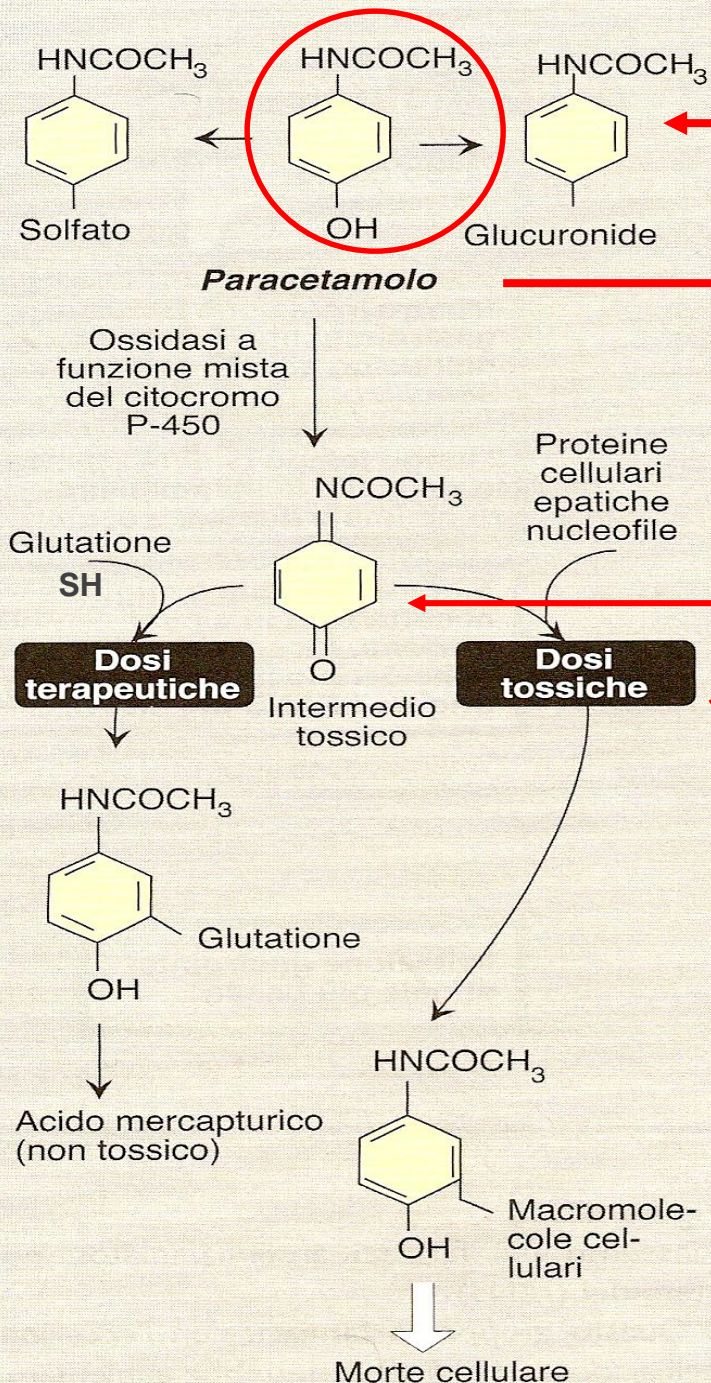
Debole

Forte

Paracetamolo



Metabolism of paracetamol



Liver

Urine

N-acetyl-p-benzoquinone

Depletion of hepatic glutathione